STIC-EIC1600/2900

376127

From: STIC-EIC1600/2900@usplo.gov

Sent: Thursday, October 13, 2011 2:37 PM

To: Winakur, Eric

Ce: STIC-EIC1600/2900

Subject: Confirmation Receipt: 1600 Search Request - 10566406

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Requester -----

Name: WINAKUR, ERIC FRANK

Organization: TC 3700 Art Unit: 3777

Employee Number
Office Location: RND-7A31
Phone Number: (571)272-4736
Email: eric.winakur@uspto.gov

Request Detail

Attachment: chemRea10566406.doc

Case/Application number: 10566406 PALM Priority App. Filing Date:

Format for Search Results: SCORE & EMAIL

Meaning of unusual acronyms or initialisms:

Identify the novelty:

Additional Comments:

Please see my attachment for details. If you need additional info, please contact me. Wasn't sure what SCORE format is, but checked it in case that is preferable to e-mail.

Request Date: Thursday, October 13, 2011 2:36 PM

=> file registry

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STRUCTURE FILE UPDATES: 17 OCT 2011 HIGHEST RN 1337015-67-4
DICTIONARY FILE UPDATES: 17 OCT 2011 HIGHEST RN 1337015-67-4

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TSCA INFORMATION NOW CURRENT THROUGH June 24, 2011.

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FILE COVERS 1907 - 18 Oct 2011 VOL 155 ISS 17
FILE LAST UPDATED: 17 Oct 2011 (20111017/ED)
REVISED CLASS FIELDS (/NCL) LAST RELGADED: Aug 2011
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Aug 2011

ZCAplus now includes complete International Patent Classification (IPC) reclassification data for the second quarter of 2011.

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This file contains CAS Registry Numbers for easy and accurate substance identification.

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FILE 'MEDLINE' ENTERED AT 11:13:17 ON 18 OCT 2011
FILE 'EMBASE' ENTERED AT 11:13:17 ON 18 OCT 2011
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=> d stat que L45

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PROCESSING COMPLETED FOR L49 L50 12 DUP REM L48 L49 (9 DUPLICATES REMOVED)

ANSWERS '1-7' FROM FILE ZCAPLUS ANSWERS '8-9' FROM FILE MEDLINE ANSWER '10' FROM FILE BIOSIS ANSWERS '11-12' FROM FILE WPIX

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L50 ANSWER 1 OF 12 ZCAPLUS COPYRIGHT 2011 ACS on STN DUPLICATE 1

ACCESSION NUMBER: 2010:784810 ZCAPLUS Full-text

DOCUMENT NUMBER: 153:118116 ENTRY DATE:

Entered STN: 25 Jun 2010 TITLE:

Method for making silicone hydrogel contact lenses INVENTOR(S): Chang, Frank; Vogt, Juergen; Pruitt, John Dallas;

Qian, Xinming; Smith, Dawn A.; Domschke, Angelika Maria; Holland, Troy Vernon

PATENT ASSIGNEE(S): Novartis AG, Switz.; Turek, Richard Charles; Wu,

Daging

SOURCE:

PCT Int. Appl., 46pp. CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

CLASSIFICATION: 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 63

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO.							TE APPLICATION NO. 100624 WO 2009-US47428 T, AU, AZ, BA, BB, BG, BH, BR, BI, CU, CZ, DE, DK, DM, DO, DZ, EK, GM, GT, HN, HR, HU, ID, IL, II, K, KZ, LA, LC, LK, LR, LS, LT, LI, W, MX, MY, MZ, NA, MG, NI, NO, NI, S, RU, SC, SD, SE, SG, SK, SL, SI, T, TZ, UA, UG, US, UZ, VC, VN, ZZ, DE, DK, EE, ES, FI, FR, GB, GI, V, MC, MK, MT, NI, NO, PL, PT, RC, CI, CK, MK, MZ, NA, SD, SL, SZ, TZ, MD, RU, TJ, TM					ATE						
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US	2009	0160	074		A1		2009	0625		US 2	-800	3169	93		2	0081	218	
US	7780	879			B2		2010	0824										
CA	2747	355			A1		2010	0624		CA 2	009-	2747	355		2	0090	616	
US	2010	0258	961		A1		2010	1014		US 2	009-	4563	64		2	0090	616	
KR	2011	1058	13		A		2011	0927		KR 2	011-	7016	544		2	0090	616	
EP	2374	031			A1		2011	1012		EP 2	009-	7898	23		2	0090	616	

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             SI, SK, TR
PRIORITY APPLN. INFO.:
                                           US 2008-316993
                                                             A 20081218
                                           US 2009-212623P P 20091414
US 2007-8554P P 20071220
WO 2009-US47428 W 20090616
PATENT CLASSIFICATION CODES:
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 WO 2010071691
               IPCI G02B0001-04 [I,A]
                IPCR G02B0001-04 [I,A]
                ECLA G02B0001-04B2+C08L51/08S; G02B0001-04B2+C08L83/04
US 20090160074 IPCI B29D0011-00 [I,A]; C08F0002-46 [I,A]; B29D0011-00
                       [I,A]; G02B0001-12 [I,A]; B29C0035-08 [I,A];
                       B29C0033-60 [I,A]; C08J0005-00 [I,A]; A61K0006-083
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                NCL
                       264/300.000; 264/319.000; 264/331.130; 264/496.000
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CA 2747355
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                ECLA G02B0001-04B2+C08L51/08S; G02B0001-04B2+C08L83/04
US 20100258961 IPCI B29D0011-00 [I,A]; C08F0002-50 [I,A]; G02C0007-04 [N,A]
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                NCL 264/001.380; 351/160.000H; 522/099.000
                ECLA G02B0001-04B2+C08L51/08S; G02B0001-04B2+C08L83/04
               IPCI B29D0011-00 [I.A]; C08G0077-26 [I.A]; C08J0005-00 [I.A]
 KR 2011105813
                IPCI G02B0001-04 [I,A]
EP 2374031
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R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU,

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT ABSTRACT:

A method for producing silicone hydrogel contact lenses comprises: providing a mold for making a soft contact lens, wherein the mold has a first mold half with a first molding surface defining the anterior surface of a contact lens and a second mold half with a second molding surface defining the posterior surface of the contact lens, wherein said first and second mold halves are configured to receive each other such that a cavity is formed between said first and second molding surfaces; introduce a monomer mixture of lens-forming materials into the cavity, wherein the monomer mixture comprises at least one hydrophilic amide-type vinylic monomer, at least one siloxane-containing (meth)acrylamide monomer, at least one polysiloxane vinylic monomer or macromer, and from about 0.05-1.5% of a photoinitiator, wherein the lens forming material is characterized by having an ability to be cured by a UV light having an UV-intensity of about 4.1 mW/cm2 within about 100 s; and irradiating, under a spatial limitation of actinic radiation, the lens-forming material in the mold for a time period of about 120 s or less, so as to crosslink the lens-forming material to form the silicone hydrogel contact lens, wherein the produced contact lens comprises an anterior surface defined by the first molding surface, an opposite posterior surface defined by the second molding surface, and a lens edge defined by the spatial limitation of actinic radiation.

SUPPL. TERM: silicone hydrogel contact lens INDEX TERM: Polysiloxanes ROLE: TEM (Technical or engineered material use); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (hydrogel; method for making silicone hydrogel contact lenses) INDEX TERM: Contact lenses (method for making silicone hydrogel contact lenses) INDEX TERM: Polymerization (photopolymn.; method for making silicone hydrogel contact lenses) INDEX TERM: Hydrogels (silicone; method for making silicone hydrogel contact INDEX TERM: 31900-57-9DP, Polydimethylsiloxane, bis(3-acrylamidopropyl)-terminated ROLE: IMF (Industrial manufacture); PREP (Preparation) (assumed monomer; method for making silicone hydrogel contact lenses) INDEX TERM: 31900-57-9DP, Polydimethylsiloxane, bis(2-hydroxyethoxypropyl)-terminated, reaction products with isophorone diisocyanate and isocyanatoethylmethacrylate, polymers ROLE: IMF (Industrial manufacture); TEM (Technical or engineered material use); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (assumed monomer; method for making silicone hydrogel contact lenses) INDEX TERM: 31900-57-9D, Polydimethylsiloxane, bis(3-aminopropyl)-terminated ROLE: RCT (Reactant); RACT (Reactant or reagent) (assumed monomer; method for making silicone hydrogel contact lenses) INDEX TERM: 79-39-0D, Methacrylamide, polymers 107-58-4D, N-tert-Butylacrylamide, polymers 924-42-5D, N-(Hydroxymethyl)acrylamide, polymers 1852-16-0D, N-(Butoxymethyl)acrylamide, polymers 2210-24-4D, N-Phenylacrylamide, polymers 2210-25-5D, 2680-03-7D. N-Isopropylacrylamide, polymers N,N-Dimethylacrylamide, polymers 2873-97-4D, Diacetone acrylamide, polymers 5205-93-6D, N-[3-(Dimethylamino)propyl]methacrylamide, polymers 6737-24-2D, 2-Acrylamidoglycolic acid, polymers 6976-91-6D, N,N-Dimethylmethacrylamide, polymers 7646-67-5D, N-(2-Hydroxyethyl) acrylamide, polymers 13749-61-6D, N-Isopropylmethacrylamide, polymers 13880-05-2D, N-[Tris(hydroxymethyl)methyl]acrylamide, 15214-89-8D, 2-Acrylamido-2-methyl-1-propanesulfonic acid, polymers 16669-59-3D, N-(Isobutoxymethyl)acrylamide, polymers 44817-99-4D, polymers 45021-77-0D, (3-Acrylamidopropyl)trimethylammonium chloride, polymers 95773-74-3D, polymers 115257-95-9D, polymers 115258-10-1D, polymers 862097-64-1D, polymers

INDEX TERM:

INDEX TERM:

INDEX TERM:

REFERENCE COUNT:

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1232277-88-1D, polymers 1232277-90-5D, polymers
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1232277-97-2D, polymers
ROLE: TEM (Technical or engineered material use); THU
(Therapeutic use); BIOL (Biological study); USES (Uses)
   (hydrogels; method for making silicone hydrogel contact
   lenses)
362523-60-2P
ROLE: IMF (Industrial manufacture); PREP (Preparation)
   (method for making silicone hydrogel contact lenses)
4098-71-9DP, Isophorone diisocvanate, reaction products with
Fomblin ZDOL and KF-6001 and 2-isocyanatoethyl methacrylate
4098-71-9DP, Isophorone diisocyanate, reaction products with
bis(2-hydroxyethoxypropyl)-polydimethylsiloxane and
isocvanatoethylmethacrylate, polymers 17096-07-0DP,
polymers 30674-80-7DP, 2-Isocyanatoethyl methacrylate,
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diisocyanate and KF-6001 30674-80-7DP, reaction products
with bis(2-hydroxyethoxypropyl)-polydimethylsiloxane and
isophorone diisocyanate, polymers 107852-51-7DP, Fomblin
Z-DOL, reaction products with isophorone diisocyanate and
KF-6001 and 2-isocvanatoethyl methacrylate 156327-07-0DP,
KF-6001, reaction products with Fomblin ZDOL and isophorone
diisocvanate and 2-isocvanatoethyl methacrylate
156327-07-0DP, reaction products with isophorone
diisocyanate and isocyanatoethylmethacrylate, polymers
ROLE: IMF (Industrial manufacture); TEM (Technical or
engineered material use); THU (Therapeutic use); BIOL
(Biological study); PREP (Preparation); USES (Uses)
   (method for making silicone hydrogel contact lenses)
814-68-6, Acrylovl chloride
                             97917-34-5
ROLE: RCT (Reactant); RACT (Reactant or reagent)
   (method for making silicone hydrogel contact lenses)
      THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS
      RECORD.
(1) Edmond, Q; WO 2006071479 A1 2006 ZCAPLUS
(2) Johnson & Johnson Vision Prod; EP 0940447 A2 1999
          ZCAPLUS
(3) Mueller Karl F; US 4605712 A 1986 ZCAPLUS
(4) Nicolson Paul C; US 20070105973 A1 2007 ZCAPLUS
(5) Novartis Ag; WO 2008008752 A2 2008 ZCAPLUS
(6) Novartis Ag: WO 2009085902 A1 2009 ZCAPLUS
(7) Phelan John C; US 20050237483 A1 2005
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L50 ANSWER 2 OF 12 ZCAPLUS COPYRIGHT 2011 ACS on STN DUPLICATE 2
ACCESSION NUMBER: 2010:84996 ZCAPLUS Full-text
DOCUMENT NUMBER: 152:170552
ENTRY DATE: Entered STN: 22 Jan 2010

TITLE: Silicone-containing polymeric materials with

hydrolyzable groups

INVENTOR(S): Chang, Frank; Smith, Dawn A.; Medina, Arturo N.;

Quinn, Michael Hugh; Chapoy, L. Lawrence

PATENT ASSIGNEE(S): Novartis AG, Switz.
SOURCE: U.S. Pat. Appl. Publ., 18 pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent

LANGUAGE: English

CLASSIFICATION: 38-3 (Plastics Fabrication and Uses)

KIND DATE

Section cross-reference(s): 63

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. K

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					A1 20100121 US 2009-499117 20090708											
CA 2730		01.		A1					CA 2009-2730506							
WO 2010		93							WO 2009-US49859							
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PRIORITY APP	LN.	INFO	.:						US 2							
									WO 2	009-	US 49	859		W 2	0090	708
PATENT CLASS										m = 011						
PATENT NO.																
US 20100014																
05 20100014	047													2000	01-1	2 [I,A]
		IPC														2 [I,A]
		NCL		351/										2000	O1 1.	2 [1,11]
		ECL		C08F										30/0:	ρ,	
				COSF												
																-14F;
				G02B												
				G02B	0001	-04B	2+C0	8L33	/08;	G02	в000	1-04	B2+C	08L3	3/10	
CA 2730506		IPC	Ι	B29D	0011	-00	[I,A]; C	08F0	230-	08 [I,A]	; C0	8F02	83-1	2
				[I,A]; C	08F0	290-	06 [I,A]	; G0	2B00	01-0	4 [I	, A]		
		IPC	R	B29D	0011	-00	[I,A]; C	08F0	230-	08 [I,A]	; C0	8F02	83-1	2
				[I,A]; C	08F0	290-	06 [I,A]	; G0	2B00	01-0	4 [I	, A]		
		ECL	A	C08F	0290	-06;	B29	D001	1-00	C4;	C08F	0008	-12+	30/0	8;	

APPLICATION NO.

DATE

C08F0008-12+20/00; C08F0283-12; C08F0283-12D; C08F0290-00; C08F0290-06F; C08F0290-14; C08F0290-14F; G02B0001-04B2+C08L51/08S; G02B0001-04B2+C08L83/04; G02B0001-04B2+C08L33/08; G02B0001-04B2+C08L33/10 WO 2010011493 IPCI B29D0011-00 [I,A]; C08F0230-08 [I,A]; C08F0283-12 [I,A]; C08F0290-06 [I,A]; G02B0001-04 [I,A] IPCR B29D0011-00 [I,A]; C08F0230-08 [I,A]; C08F0283-12 [I,A]; C08F0290-06 [I,A]; G02B0001-04 [I,A] ECLA C08F0290-06; B29D0011-00C4; C08F0008-12+30/08; C08F0008-12+20/00; C08F0283-12; C08F0283-12D; C08F0290-00; C08F0290-06F; C08F0290-14; C08F0290-14F; G02B0001-04B2+C08L51/08S; G02B0001-04B2+C08L83/04; G02B0001-04B2+C08L33/08; G02B0001-04B2+C08L33/10 EP 2307190 IPCI B29D0011-00 [I,A]; C08F0230-08 [I,A]; C08F0283-12 [I,A]; C08F0290-06 [I,A]; G02B0001-04 [I,A] ECLA C08F0290-06; B29D0011-00C4; C08F0008-12+30/08; C08F0008-12+20/00; C08F0283-12; C08F0283-12D; C08F0290-00; C08F0290-06F; C08F0290-14; C08F0290-14F; G02B0001-04B2+C08L51/08S; G02B0001-04B2+C08L83/04;

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT ABSTRACT:

The invention provides a silicone hydrogel contact lens including a hydrolyzable units. The hydrolyzable units are converted by hydrolysis into a hydrophilic units which is capable of imparting the silicone hydrogel contact lens a hydrophilic surface without post-curing surface treatment. The invention also provides an actinically crosslinkable prepolymer including hydrolyzable units and use of the prepolymer of the invention. In addition, the invention provides a method for making wettable silicone hydrogel contact lens without posterior surface treatments other than hydrolysis.

SUPPL. TERM: silicone hydrolyzable deriv hydrogel hydrophilic contact lens

INDEX TERM: Polysiloxanes

ROLE: TEM (Technical or engineered material use); USES

(Uses)

(acrylic; silicone-containing polymeric materials with hydrolyzable groups for hydrogel soft contact lenses)

G02B0001-04B2+C08L33/08; G02B0001-04B2+C08L33/10

INDEX TERM: Acrylic polymers

ROLE: TEM (Technical or engineered material use); USES

(Uses)

(polysiloxane-; silicone-containing polymeric materials with

hydrolyzable groups for hydrogel soft contact lenses)

INDEX TERM: Hydrogels

INDEX TERM:

(silicone-containing polymeric materials with hydrolyzable

groups for hydrogel soft contact lenses)

INDEX TERM: Contact lenses

(soft; silicone-containing polymeric materials with

hydrolyzable groups for hydrogel soft contact lenses) 75-01-4D, Vinyl chloride, polymers with unsatd. silicone derivs. having hydrolyzable groups 75-35-4D, Vinylidene chloride, polymers with unsatd. silicone derivs. having

hydrolyzable groups 79-06-1D, Acrylamide, polymers with unsatd. silicone derivs. having hydrolyzable groups

79-10-7D, Acrylic acid, polymers with unsatd. silicone derivs. having hydrolyzable groups 79-39-0D, Methacrylamide, polymers with unsatd, silicone derivs, having hydrolyzable groups 79-41-4D, Methacrylic acid, polymers with unsatd. silicone derivs. having hydrolyzable 80-62-6D, Methyl methacrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 88-12-0D, N-Vinv1-2-pyrrolidone, polymers with unsatd. silicone derivs. having hydrolyzable groups Methyl acrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 97-63-2D, Ethyl methacrylate, polymers with unsatd. silicone derivs. having hydrolyzable 100-42-5D, Styrene, polymers with unsatd. silicone derivs, having hydrolyzable groups 103-11-7D, 2-Ethylhexyl acrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 105-38-4D, Vinyl propionate, polymers with unsatd, silicone derivs, having hydrolyzable groups 106-98-9D, 1-Butene, polymers with unsatd. silicone derivs. having hydrolyzable groups 106-99-0D, Butadiene, polymers with unsatd. silicone derivs. having hydrolyzable groups 107-13-1D, Acrylonitrile, polymers with unsatd, silicone derivs. having hydrolyzable groups 107-18-6D, Allyl alcohol, polymers with unsatd. silicone derivs. having hydrolyzable groups 108-05-4D, Vinyl acetate, polymers with unsatd. silicone derivs. having hydrolyzable groups 109-92-2D, Vinvl ethyl ether, polymers with unsatd. silicone derivs. having hydrolyzable groups 123-20-6D, Vinyl butyrate, polymers with unsatd. silicone derivs. having hydrolyzable groups 126-98-7D, Methacrylonitrile, polymers with unsatd, silicone derivs, having hydrolyzable groups 126-99-8D, Chloroprene, polymers with unsatd. silicone derivs. having hydrolyzable groups 140-88-5D, Ethyl acrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 689-12-3D, Isopropyl acrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 818-61-1, 2-Hydroxyethyl acrylate 868-77-9D, 2-Hydroxyethyl methacrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 925-60-0D, Propyl acrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 1337-81-1D, Vinylpyridine, polymers with unsatd, silicone derivs, having hydrolyzable groups 2210-28-8D, Propyl methacrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 2235-00-9D, N-Vinylcaprolactam, polymers with unsatd. silicone derivs. having hydrolyzable groups 2680-03-7D, N, N-Dimethylacrylamide, polymers with unsatd. silicone derivs. having hydrolyzable groups 2867-47-2D, Dimethylaminoethyl methacrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 2873-97-4D, N-(1,1-Dimethyl-3-oxobutyl)acrylamide, polymers with unsatd. silicone derivs. having hydrolyzable groups 3063-94-3D, Hexafluoroisopropyl methacrylate, polymers with unsatd. silicone derivs. having hydrolyzable groups 3066-71-5D, Cyclohexyl acrylate, polymers with unsatd. silicone derivs.

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having hydrolyzable groups 3195-78-6D,
N-Vinyl-N-methylacetamide, polymers with unsatd. silicone
derivs. having hydrolyzable groups 5202-78-8D,
N-Vinylacetamide, polymers with unsatd. silicone derivs.
having hydrolyzable groups 5873-43-8D, Vinyl valerate,
polymers with unsatd. silicone derivs. having hydrolyzable
       7534-94-3D, Isobornyl methacrylate, polymers with
unsatd, silicone derivs, having hydrolyzable groups
13081-44-2D, N-(2-Dimethylaminoethyl)methacrylamide,
polymers with unsatd. silicone derivs. having hydrolyzable
       13162-05-5D, N-Vinylformamide, polymers with
groups
unsatd. silicone derivs. having hydrolyzable groups
17096-07-0D, 3-[Tris(trimethylsilyloxy)silyl]propyl
methacrylate, polymers with unsatd, silicone derivs, having
hydrolyzable groups 18151-85-4D,
3-Methacryloxypropylpentamethyldisiloxane, polymers with
unsatd, silicone derivs, having hydrolyzable groups
18547-93-8D, 1,3-Bis(3-
Methacryloxypropyl)tetramethyldisiloxane, polymers with
unsatd. silicone derivs. having hydrolyzable groups
25013-15-4D, Vinyltoluene, polymers with unsatd. silicone
derivs. having hydrolyzable groups 25584-83-2D,
Hydroxypropyl acrylate, polymers with unsatd. silicone
derivs. having hydrolyzable groups 27813-02-1D,
Hydroxypropyl methacrylate, polymers with unsatd. silicone
derivs. having hydrolyzable groups 38785-10-3D,
Trifluoroethyl methacrylate, polymers with unsatd. silicone
derivs. having hydrolyzable groups 54174-14-0D, Glycerol
methacrylate, polymers with unsatd. silicone derivs. having
hydrolyzable groups 76643-43-1D, Hexafluorobutyl
methacrylate, polymers with unsatd. silicone derivs. having
hydrolyzable groups 94086-93-8D,
2-(Methacryloyloxy) propyltrimethylammonium chloride,
polymers with unsatd. silicone derivs. having hydrolyzable
       1061202-60-5D, Aminopropyl methacrylate
hydrochloride, polymers with unsatd. silicone derivs. having
hydrolyzable groups 1204770-07-9D, polymers
1204770-08-0D, polymers 1204770-09-1
ROLE: TEM (Technical or engineered material use); USES
(Uses)
   (silicone-containing polymeric materials with hydrolyzable
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(silicone-containing polymeric materials with hydrolyzable groups for hydrogel soft contact lenses)

L50 ANSWER 3 OF 12 ZCAPLUS COPYRIGHT 2011 ACS on STN DUPLICATE 3
ACCESSION NUMBER: 2010:352928 ZCAPLUS Full-text

DOCUMENT NUMBER: 152:353415

ENTRY DATE: Entered S'

Entered STN: 19 Mar 2010

TITLE: Continuous non-invasive ophthalmic glucose sensor for diabetics

AUTHOR(S): Domschke, Angelika M.

CORPORATE SOURCE: Duluth, GA, USA SOURCE: Chimia (2010),

SOURCE: Chimia (2010), 64(1-2), 43-44
CODEN: CHIMAD; ISSN: 0009-4293
PUBLISHER: Swiss Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

CLASSIFICATION: 9-5 (Biochemical Methods)

ABSTRACT:

A contact lens with an embedded glucose sensor hologram has been

developed for continuous non-invasive monitoring of glucose levels in

diabetics. This article describes the development and initial clin. testing of this ophthalmic glucose sensor and provides a comparison to current

continuous glucose monitors.

SUPPL. TERM: glucose sensor hologram contact lens diagnosis

diabetes; spectroscopy acrylamidophenylboronate hydrogel

glucose sensor contact lens

INDEX TERM: Contact lenses

Diabetes mellitus Diagnosis

Glucose sensors

Human

Hydrogels Spectroscopy

(contact lens with embedded

glucose sensor hologram consisting of 3-acrylamidophenylboronic acid hydrogel for continuous non-invasive monitoring of glucose

levels in diabetics)

INDEX TERM:

(reflection; contact lens with

embedded glucose sensor hologram

consisting of 3-acrylamidophenylboronic acid

hydrogel for continuous non-invasive monitoring of

glucose levels in diabetics)

INDEX TERM: 50-99-7, Glucose, analysis

Holography

ROLE: ANT (Analyte); DGN (Diagnostic use); ANST (Analytical

study); BIOL (Biological study); USES (Uses)

(contact lens with embedded

glucose sensor hologram consisting of 3-acrylamidophenylboronic acid hydrogel for

continuous non-invasive monitoring of glucose levels in diabetics)

INDEX TERM: 99349-68-5, 3-Acrylamidophenylboronic acid

ROLE: ARG (Analytical reagent use); TEM (Technical or engineered material use); ANST (Analytical study); USES

(Uses)

(contact lens with embedded

glucose sensor hologram consisting of 3-acrylamidophenylboronic acid hydrogel for

continuous non-invasive monitoring of glucose levels in diabetics)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1

CITINGS)

DATE LAST CITED: Date last citing reference entered STN: 13 Sep 2011 OS.CITING.REFS: CAPLUS 2011:1143317

US.CITING.REFS: CAPLUS ZUIT:114331

REFERENCE COUNT: 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD.

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REFERENCE(S):
                  (1) Denisyuk, Y; Opt Spectrosc 1963, V18, P152
                   (2) Domschke, A; Diabetes Technol Ther 2002, V4, P49
                   (3) Edelman, S; Diabetes Technol Ther 2009, V11, P68
                   (4) Garg, S; Diabetes Technol Ther 2009, V11, P65 ZCAPLUS
                   (5) Hirsch, I; J Clin Endocrinol Metab 2009, V94, P2232
                             ZCAPLUS
                   (6) Hisamitsu, I; Pharm Res 1997, V14, P289 ZCAPLUS
                   (7) Kabilan, S; Biosensors Bioelectron 2005, V20, P1602
                             ZCAPLUS
                   (8) Kabilan, S; J Mol Recognit 2004, V17, P162 ZCAPLUS
                   (9) Lee, M; Anal Chem 2004, V76, P5748 ZCAPLUS
                   (10) March, W; Diabetes Technol Ther 2002, V4, P49
                   (11) March, W; Diabetes Technol Ther 2004, V6, P49
                   (12) Srivastav, G; AIOC Proceedings 2006
L50 ANSWER 4 OF 12 ZCAPLUS COPYRIGHT 2011 ACS on STN DUPLICATE 4
ACCESSION NUMBER:
                        2006:130936 ZCAPLUS Full-text
DOCUMENT NUMBER:
                         145:119579
ENTRY DATE:
                        Entered STN: 13 Feb 2006
TITLE:
                        Initial clinical testing of a holographic non-invasive
                        contact lens glucose sensor
AUTHOR(S):
                        Domschke, Angelika; March, Wayne F.; Kabilan,
                        Satyamoorthy; Lowe, Christopher
                        CIBA Vision Corporation, Duluth, GA, USA
CORPORATE SOURCE:
SOURCE:
                        Diabetes Technology & Therapeutics (2006), 8(1), 89-93
                        CODEN: DTTHFH; ISSN: 1520-9156
PUBLISHER:
                        Mary Ann Liebert, Inc.
DOCUMENT TYPE:
                        Journal
LANGUAGE:
                        English
CLASSIFICATION:
                        9-1 (Biochemical Methods)
ABSTRACT:
Introduction: the purpose of the present study was to determine the effectiveness
a new holog, contact lens glucose sensor for the non-invasive
monitoring of blood glucose. Methods: one fasting normal subject was given
an oral challenge consisting of 44 g of glucose. The contact lens
hologram signal and fingerstick blood glucose were measured over a 26- min
period. Results: the contact lens hologram signal appeared to track blood
glucose well. The contact lens was comfortable and well tolerated.
Conclusion: the holog, contact lens glucose sensor shows promise as a
non-invasive home glucose monitor.
SUPPL. TERM:
                   glucose sensor holog contact lens
INDEX TERM:
                   Biosensors
                    Contact lenses
                   Diagnosis
                   Electronic device fabrication
                     Glucose sensors
                   Holography
                   Human
                      (initial clin. testing of a holog. non-invasive
                      contact lens glucose
                      sensor)
INDEX TERM:
                  50-99-7, D-Glucose, analysis
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contact lens glucose
                      sensor)
INDEX TERM:
                   7473-98-5, Darocur 1173 99349-68-5, 3-Acrylamidophenyl
                   boronic acid 159073-29-7, Nelfilcon A
                   ROLE: ARU (Analytical role, unclassified); BUU (Biological
                   use, unclassified); DEV (Device component use); ANST
                   (Analytical study); BIOL (Biological study); USES (Uses)
                      (initial clin. testing of a holog. non-invasive
                      contact lens glucose
                      sensor)
OS.CITING REF COUNT: 10 THERE ARE 10 CAPLUS RECORDS THAT CITE THIS RECORD (10
                         CITINGS)
DATE LAST CITED: Date last citing reference entered STN: 21 Jun 2011
OS.CITING.REFS: CAPLUS 2011:740788; 2010:841367; 2009:799398; 2009:485976;
                         2009:92134; 2008:793135; 2008:663156; 2007:734952;
                         2007:451604; 2007:373809
                         THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                   16
                         RECORD.
REFERENCE(S):
                   (1) Ballerstadt, R; Anal Chim Acta 1997, V345, P203 ZCAPLUS
                   (2) Friends, G; J Biomed Mater Res 1992, V26, P59 ZCAPLUS
                   (3) Kabilan, S; Biosensors Bioelectron 2005, V20, P1602
                             ZCAPLUS
                   (4) Kabilan, S; J Mol Recognit 2004, V17, P162 ZCAPLUS
                   (5) Lee, M; Anal Chem 2004, V76, P5748 ZCAPLUS
                   (6) Lorand, J; J Org Chem 1959, V24, P769 ZCAPLUS
                   (7) March, W; Bioinstrumentation: Research, Developments and
                             Applications 1990, P31
                   (8) March, W; Diabetes Care 1982, V5, P259 ZCAPLUS
                   (9) March, W; Diabetes Technol Ther 2000, V2, P27 ZCAPLUS
                   (10) March, W; Diabetes Technol Ther 2002, V3, P46
                   (11) March, W; Diabetes Technol Ther 2002, V4, P49
                   (12) March, W; Diabetes Technol Ther 2004, V6, P49
                   (13) March, W; Diabetes Technol Ther 2004, V6, P782 ZCAPLUS
                   (14) Marshal, A; Anal Chem 2003, V75, P4423
                   (15) Muller, B; US 5583163 1996 ZCAPLUS
                   (16) Russell, R; Anal Chem 1999, V71, P3126 ZCAPLUS
L50 ANSWER 5 OF 12 ZCAPLUS COPYRIGHT 2011 ACS on STN DUPLICATE 5
ACCESSION NUMBER:
                        2005:300731 ZCAPLUS Full-text
                         142:351677
DOCUMENT NUMBER:
ENTRY DATE:
                        Entered STN: 07 Apr 2005
TITLE:
                        Ophthalmic device comprising a holographic sensor
INVENTOR(S):
                        Lowe, Christopher Robin; Kabilan, Satvamoorthy; Blyth,
                         Jeffrey; Domschke, Angelika; Smith, Dawn; Karangu,
                         Njeri
PATENT ASSIGNEE(S):
                         Smart Holograms Ltd., UK; Ciba Vision Corp.; Cambridge
                        University Technical Services Ltd.
SOURCE:
                        PCT Int. Appl., 24 pp.
                        CODEN: PIXXD2
DOCUMENT TYPE:
                        Patent
LANGUAGE:
                        English
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ROLE: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study) (initial clin. testing of a holog. non-invasive

CLASSIFICATION: 9-1 (Biochemical Methods)

Section cross-reference(s): 63

FAMILY ACC. NUM. COUNT: 3 PATENT INFORMATION:

PATENT I	KIN		DATE			APPL		ION			D	ATE			
WO 2005					2005									0040	927
	AE, AG											BY.			
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	GE, GH														
	LK, LR														
	NO, NZ														
	TJ, TM	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW
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	EE, ES	FI,	FR,	GB,	GR,	HU,	ΙE,	IT,	LU,	MC,	NL,	PL,	PT,	RO,	SE,
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	SN, TD														
AU 2004:	276949		A1					AU 2	004-	2769	49		2	0040	927
AU 2004:			B2		2008	0731									
CA 2540			A1		2005			CA 2							
EP 1664					2006									0040	
R:	AT, BE											NL,	SE,	MC,	PT,
JP 2007	IE, SI	, F.I.,	RO,	CY,	1K,	BG,	CZ,	EE,	HU,	PL,	SK.		2	0040	007
JP 4566	206333		T B2 A1		2007	1020		UP Z	006-	52/4	0.1		2	0040	921
US 2009	000 0001607		31		2010	0122		110 2	008-	3265	n		2	0080	216
US 7998	412		B2		2011	0122		05 2	000	3203	•			0000.	210
PRIORITY APP						0010		GB 2	003-	2248	8		A 2	0030	925
								GB 2	004-	1399			A 2	0040	122
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								WO 2 US 2	001-	GB61			W 2	0010	108
								US 2	002-	1695	02				
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								US 2	006-	5730	97		A2 2	0060	323
PATENT CLASS															
PATENT NO.							ASSI	FICA	TION	COD	ES				
WO 20050214							21			1 20		0 21		12700	01 77
WO 20050314	42 IP	-1			G02					1-/5	[IC	5,/]	; 60.	INUU.	21-77
	TD	CR	G01N							77 [т ъ 1		1 1100	21_7	0
	II.	-11	[N, A						021-	// [1,71	, 00	11400.	61-7.	o
	EC.	LA							• G0	2800	01-0	4R2+	5011	งากว	1:75R;
	20.				:77B										1
AU 20042769	49 IP	CI.	G01N											0021	-77
			[I,A												
			G01N											021-	77
	[I,C*]; G01N0021-77 [I,A]; G01N0021-78 [N,A];														
	G02B0001-04 [I,C*]; G02B0001-04 [I,A]														
	IP	CR	G01N						021-	75 [I,A]	; G0	1N00	21-7	8
	[I,A]; G02B0001-04 [I,A]														
	ECLA G01N0021-75; G01N0021-77; G02B0001-04B2; S01N0021:75R								1:75R;						
S01N0021:77B2D; S01N0021:77H2; S01N0021:78															

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CA 2540076
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                       G01N0021-75 [I,A]; G01N0021-77 [I,A]; G02B0001-04
                       [I.A]: G02C0007-04 [I.A]
                 TPCR
                       G02C0007-04 [I,A]; G01N0021-75 [I,A]; G01N0021-77
                       [I,A]; G01N0021-78 [N,A]; G02B0001-04 [I,A]
                 ECLA
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                       S01N0021:77B2D; S01N0021:77H2; S01N0021:78
EP 1664909
                 IPCI
                       G02C0007-04 [ICM, 7]; G01N0021-75 [ICS, 7]; G01N0021-77
                       [ICS, 7]; G02B0001-04 [ICS, 7]
                 IPCR
                       G01N0021-75 [I.A]; G01N0021-77 [I.A]; G01N0021-78
                       [N,A]; G02B0001-04 [I,A]
                 ECLA
                       G01N0021-75; G01N0021-77; G02B0001-04B2; S01N0021:75R;
                       S01N0021:77B2D; S01N0021:77H2; S01N0021:78
JP 2007506999
                 IPCI
                       G02C0007-04 [I,A]; G02B0005-32 [I,A]; G01N0033-483
                       [I,A]; G01N0021-77 [I,A]; G01N0033-66 [I,A];
                       G02C0007-04 [I,A]; G02B0005-32 [I,A]; G01N0033-483
                        | I.A|: G01N0033-66 | I.A|
                 IPCR
                       G02C0007-04 [I,A]; G01N0021-75 [I,A]; G01N0021-77
                       [I,A]; G01N0021-78 [N,A]; G01N0033-483 [I,A];
                       G01N0033-66 [I,A]; G02B0001-04 [I,A]; G02B0005-32 [I,A]
                 ECLA
                       G01N0021-75; G01N0021-77; G02B0001-04B2; S01N0021:75R;
                        S01N0021:77B2D; S01N0021:77H2; S01N0021:78
                 FTERM 2G045/AA25; 2G045/DA31; 2G045/FA11; 2G054/CA25;
                        2G054/EA06; 2H006/BB10; 2H006/BC00; 2H006/BC07;
                        2H006/BE05; 2H049/CA06; 2H049/CA09; 2H049/CA24;
                        2H049/CA28
US 20090021697 IPCI
                       A61B0003-125 [I,A]; A61B0005-00 [I,A]; G01N0021-00
                       II.Al
                 TPCR
                       A61B0003-125 [I,A]; A61B0005-00 [I,A]
                 NCL
                       351/219.000; 600/309.000; 422/082.050; 436/164.000;
                       436/165.000; 436/169.000; 436/528.000; 436/529.000;
                        436/530.000; 436/531.000; 600/300.000; 600/316.000;
                       600/318,000; 600/319,000
                 ECLA
                       G01N0021-75; A61B0003-125; A61B0005-145G; G01N0021-77;
                       G03H0001-18; S01N0021:75R; S01N0021:77B2D;
                        S01N0021:77H2; S01N0021:78; S03H0001:04D; S03H0001:28
ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT
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ABSTRACT:
An ophthalmic device comprises a holog. element comprising a medium and,

disposed therein, a hologram, wherein an optical characteristic of the element changes as a result of a variation of a phys. property of the medium, and wherein the variation arises as a result of interaction between the medium and an analyte present in an ocular fluid. The device may be used for the detection of analytes such as glucose.

SUPPL. TERM: ophthalmic device comprising holog sensor
INDEX TERM: Sensors
(Holog.; ophthalmic device comprising a holog. sensor)
INDEX TERM: Apparatus
(Implantable; ophthalmic device comprising a holog. sensor)
INDEX TERM: Eye
(Ocular fluid; ophthalmic device comprising a holog. sensor)

INDEX TERM: Apparatus (Ophthalmic; ophthalmic device comprising a holog, sensor) INDEX TERM: Amino group Contact lenses Crosslinking Eve Fluids Human Optical properties Physical and chemical properties Polymerization Reaction (ophthalmic device comprising a holog. sensor) INDEX TERM: Polymers, uses ROLE: DEV (Device component use); USES (Uses) (ophthalmic device comprising a holog. sensor) INDEX TERM: Monomers ROLE: RCT (Reactant); RACT (Reactant or reagent) (ophthalmic device comprising a holog. sensor) INDEX TERM: 50-99-7, D-Glucose, analysis ROLE: ANT (Analyte): ANST (Analytical study) (ophthalmic device comprising a holog. sensor) INDEX TERM: 50-21-5, Lactic acid, analysis ROLE: ARU (Analytical role, unclassified); ANST (Analytical study) (ophthalmic device comprising a holog. sensor) INDEX TERM: 98-80-6, Phenylboronic acid ROLE: DEV (Device component use); USES (Uses) (ophthalmic device comprising a holog. sensor) INDEX TERM: 7440-22-4, Silver, miscellaneous ROLE: MSC (Miscellaneous) (ophthalmic device comprising a holog. sensor) INDEX TERM: 79-06-1, Acrylamide, reactions 110-26-9, Methylenebisacrylamide 3845-76-9, N-[3-(Dimethylamino)propyl]acrylamide 6737-24-2, Acrylamidoglycolic acid 24650-42-8, 2-DiMethoxy-2-phenyl-acetophenone 99349-68-5, 3-Acrylamidophenylboronic acid 849146-10-7 ROLE: RCT (Reactant); RACT (Reactant or reagent) (ophthalmic device comprising a holog. sensor) OS CITING REF COUNT: 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS RECORD (4 CITINGS) DATE LAST CITED: Date last citing reference entered STN: 16 Feb 2009 OS.CITING.REFS: CAPLUS 2007:538645; 2007:410728; 2006:240734; 2005:1328919 REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS

RECORD.

REFERENCE(S): (1) Bradley, M; US 20030103868 A1 2003 ZCAPLUS

(2) British Technology Group Limited; WO 9526499 Al 1995

ZCAPLUS

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(4) Lednev, I; US 20030027240 Al 2003(5) Novartis Erfind Verwalt Gmbh; WO 9934244 A 1999 ZCAPLUS

L50 ANSWER 6 OF 12 ZCAPLUS COPYRIGHT 2011 ACS on STN DUPLICATE 6

ACCESSION NUMBER: 2005:141376 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 142:194009

ENTRY DATE: Entered STN: 18 Feb 2005

TITLE: Ophthalmic sensor

INVENTOR(S): Chapoy, Lawrence L.; Domschke, Angelika Maria; Smith, Dawn

PATENT ASSIGNEE(S): Novartis A.-G., Switz.; Novartis Pharma G.m.b.H.

SOURCE: PCT Int. Appl., 28 pp.
CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

CLASSIFICATION: 9-16 (Biochemical Methods)

PATENT INFORMATION:

PA	PATENT NO.					DATE								20040806 7, BZ, CA, CH, 8, FI, GB, GD, 7, KR, KZ, LC, 7, MZ, NA, NI, 8, SK, SL, SY, 1, 2A, ZM, ZW, 7, CZ, DE, DK, 17, RC, SE, 17, ML, MR, NE,			
WC	WO 2005015237													2	0040	806	
	W:	ΑE,	AG,	AL,	AM,	AT,	AU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,
		CN,	co,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,
		GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KZ,	LC,
		LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NI,
		NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,
		TJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW
	RW:	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,
		AZ,	BY,	KG,	ΚZ,	MD,	RU,	ΤJ,	TM,	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,
		EE,	ES,	FI,	FR,	GB,	GR,	HU,	ΙE,	IT,	LU,	MC,	NL,	PL,	PT,	RO,	SE,
		SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,
		SN,	TD,	TG													
EP 1654543 A1 20060510 EP 2004-763860						60		2	20040806								
EF	1654	543			B1		2010	0707									
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	ΙT,	LI,	LU,	NL,	SE,	MC,	PT,
		ΙE,	SI,	FI,	RO,	CY,	TR,	BG,	CZ,	EE,	HU,	PL,	SK				
AT	4734	44			T		2010	0715		AT 2	004-	7638	60		2	0040	806
ES	2350	933			Т3												
PRIORIT	Y APP	LN.	INFO	.:						US 2	003-	4932	41P	1	P 2	0030	807
										WO 2	004-	EP88	25	1	W 2	0040	806
PATENT																	
PATENT							AMIL	Y CL	ASSI	FICA	TION	COD	ES				
				I		0033	-66	[ICM	,7];	A61	в000	5-00	[IC	S,7]	; G0	1N00	33-58
						-58	ft.A	1 · G	01N0	033-	66 [T.A1					
IPCR G01N0033-58 [I,A]; G01N0033-66 [I,A] ECLA G01N0033-58D; G01N0033-66																	

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EP 1654543
               IPCI
                       G01N0033-66 [I,C]; G01N0033-66 [I,A]; A61B0005-00
                       [I.Cl; A61B0005-00 [I,A]; G01N0033-58 [I,C];
                       G01N0033-58 [I.A]
                 IPCR
                       G01N0033-58 [I,A]; G01N0033-66 [I,A]
                 ECLA G01N0033-58D; G01N0033-66
 AT 473444
                 IPCI G01N0033-66 [I,C]; G01N0033-66 [I,A]; A61B0005-00
                       [I,C]; A61B0005-00 [I,A]; G01N0033-58 [I,C];
                       G01N0033-58 [I,A]
                 IPCR G01N0033-66 [I.A]; A61B0005-00 [I.A]; G01N0033-58 [I.A]
                 ECLA G01N0033-58D; G01N0033-66
                 IPCI G01N0033-66 [I,A]; A61B0005-00 [I,A]; G01N0033-58 [I,A]
 ES 2350933
                 IPCR G01N0033-66 [I,A]; A61B0005-00 [I,A]; G01N0033-58 [I,A]
                ECLA G01N0033-58D; G01N0033-66
OTHER SOURCE(S):
                       MARPAT 142:194009
ABSTRACT:
This invention is generally related to a biocompatible sensor for
detecting/measuring sugar, especially glucose, in an ocular fluid in a non-invasive
or minimally invasive manner and a method for using the biocompatible sensor.
A biocompatible sensor of the invention comprises, consists essentially, or
consists of an ophthalmic device comprising a mol. sensing moiety which
interacts or, reacts with sugar to provide an optical signal which is
indicative of sugar level in an ocular fluid.
SUPPL. TERM:
                  ophthalmic sensor
INDEX TERM:
                  Biocompatibility
                  Contact lenses
                  Tear (ocular fluid)
                      (ophthalmic sensor)
INDEX TERM:
                   50-99-7, D-Glucose, analysis
                   ROLE: ANT (Analyte); BSU (Biological study, unclassified);
                   ANST (Analytical study); BIOL (Biological study)
                      (ophthalmic sensor)
INDEX TERM:
                   406719-91-3 406719-93-5
                   ROLE: BUU (Biological use, unclassified); BIOL (Biological
                  study); USES (Uses)
                      (ophthalmic sensor)
REFERENCE COUNT:
                  13
                        THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS
                         RECORD.
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REFERENCE(S):
                   (2) Asher; US 2001026946 A1 2001 ZCAPLUS
                   (3) Badugu, R; ANALYTICAL CHEMISTRY 2004, V76(3), P610
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                            PHOTOBIOLOGY, A: CHEMISTRY 2001, V143(1), P39
                            ZCAPLUS
                   (5) Dicesare, N; JOURNAL OF PHYSICAL CHEMISTRY A 2001,
                            V105(28), P6834 ZCAPLUS
                   (6) Dicesare, N; TETRAHEDRON LETTERS 2002, V43(14), P2615
                            ZCAPLUS
                   (7) Front, M: US 3958560 A 1976
                   (8) Lakowicz & Dicesare; US 2004087842 A1 2004
                   (9) March; US 2003045783 A1 2003
                   (10) Menzebach; US 5535743 A 1996
                   (11) Novartis Ag; WO 0078830 A 2000 ZCAPLUS
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L50 ANSWER 7 OF 12 ZCAPLUS COPYRIGHT 2011 ACS on STN DUPLICATE 7 2005:4048 ZCAPLUS Full-text

ACCESSION NUMBER: DOCUMENT NUMBER: 143:40470

Entered STN: 04 Jan 2005

ENTRY DATE:

TITLE: Holographic glucose sensors

AUTHOR(S): Kabilan, Satyamoorthy; Marshall, Alexander J.;

Sartain, Felicity K.; Lee, Mei-Ching; Hussain, Abid; Yang, Xiaoping; Blyth, Jeff; Karangu, Njeri; James,

Karen; Zeng, Jimmy; Smith, Dawn; Domschke,

Angelika; Lowe, Christopher R.

CORPORATE SOURCE: Institute of Biotechnology, University of Cambridge,

Cambridge, CB2 1QT, UK

SOURCE: Biosensors & Bioelectronics (2005), 20(8), 1602-1610

CODEN: BBIOE4; ISSN: 0956-5663

PUBLISHER: Elsevier B.V. DOCUMENT TYPE: Journal

LANGUAGE: English 9-16 (Biochemical Methods)

CLASSIFICATION: ABSTRACT:

A novel holog, sensor system capable of detecting dynamic changes in glucose concentration has been developed. The hologram is recorded within a bio-compatible hydrogel matrix containing phenylboronic acid derivs. On binding glucose, the color of the hologram red-shifts to longer wavelengths as the hydrogel expands and this color change is used to quantify glucose concentration However, phenylboronic acids are non-selective and bind a wide variety of cis-diols. In blood, glucose is the only sugar found free at high concentration, while other sugars are typically found as part of glycoproteins and macromol. structures. Although glycoproteins have been shown to have no effect on the sensor, phenylboronic acids can bind lactate much more readily than glucose. We have designed two polymer hydrogel systems to increase the selectivity of the sensor for glucose over lactate. The first involved the use of high concns. of 3-acrylamidophenylboronic acid (3-APB) while the second system utilized 2-acrylamido-5-fluorophenylboronic acid (5-F-2-MAPB). Both systems displayed an increased selectivity to glucose over lactate at physiol. pH and ionic strength and could be deployed as selective holog.

sensors for glucose detection in physiol. fluids. SUPPL. TERM: holog glucose sensor phenylboronate

INDEX TERM: Holography

INDEX TERM:

(apparatus; holog, glucose sensors using

hydrogel matrix containing phenylboronic acid derivs.)

Glucose sensors

Hydrogels

(holog, glucose sensors using

hydrogel matrix containing phenylboronic acid

derivs.)

50-99-7, D-Glucose, analysis INDEX TERM:

ROLE: ANT (Analyte); ANST (Analytical study)

(holog. glucose sensors using

hydrogel matrix containing phenylboronic acid

```
derivs.)
INDEX TERM:
                   98-80-6DP, Phenylboronic acid, derivs.
                   99349-68-5P, 3-Acrylamidophenylboronic acid
                   853348-12-6P
                   ROLE: ARG (Analytical reagent use); SPN (Synthetic
                  preparation); ANST (Analytical study); PREP (Preparation);
                  USES (Uses)
                      (holog, glucose sensors using
                      hydrogel matrix containing phenylboronic acid
                      derivs.)
INDEX TERM:
                   814-68-6, Acryloyl chloride 30418-59-8, 3-
                   Aminophenylboronic acid 850689-32-6
                   ROLE: RCT (Reactant); RACT (Reactant or reagent)
                      (holog, glucose sensors using
                      hydrogel matrix containing phenylboronic acid
                      derivs.)
INDEX TERM:
                   50-21-5, Lactic acid, analysis
                   ROLE: ARU (Analytical role, unclassified); ANST (Analytical
                   study)
                      (interference in glucose determination; holog.
                      glucose sensors using hydrogel matrix
                      containing phenylboronic acid derivs.)
OS.CITING REF COUNT: 50 THERE ARE 50 CAPLUS RECORDS THAT CITE THIS RECORD (50
                         CITINGS)
DATE LAST CITED: Date last citing reference entered STN: 29 Sep 2011
OS.CITING.REFS: CAPLUS 2011:1225408; 2011:1143317; 2011:1031793; 2011:834723;
                         2010:1490135; 2010:1596514; 2010:1216590;
                         2010:1183648; 2010:1061518; 2010:965019; 2010:709920;
                         2010:650773; 2010:659511; 2010:461695; 2010:430797;
                         2010:352928; 2010:269535; 2010:126617; 2010:79681;
                         2009:805644; 2009:128069; 2009:378910; 2009:202424;
                         2008:793135; 2008:663156; 2008:611503; 2008:464452;
                         2008:342715; 2008:322918; 2008:271079; 2007:1301251;
                         2007:1199499; 2007:1154354; 2007:893720; 2007:854989;
                         2007:688510; 2007:556181; 2007:388210; 2007:382879;
                         2007:373809; 2007:227238; 2007:97080; 2007:79743;
                         2006:685228; 2006:587414; 2006:158201; 2006:130941;
                         2006:130936; 2005:1120543; 2005:1059086
REFERENCE COUNT:
                         THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS
                   34
                         RECORD.
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REFERENCE(S):
                   (2) Alexeev, V; Anal Chem 2003, V75(10), P2316 ZCAPLUS
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                             Applications 1992
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V33(1), P163 ZCAPLUS

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L50 ANSWER 8 OF 12 MEDLINE on STN

ACCESSION NUMBER: 2011856536 IN-PROCESS Full-text

ZCAPLUS

DOCUMENT NUMBER: PubMed ID: 21790271

TITLE: A blueprint for telerehabilitation guidelines-october 2010.

AUTHOR: A Diueprint for telerenabilitation guidelines-october 2010
AUTHOR: Brennan David M; Tindall Lyn; Theodoros Deborah; Brown

Janet; Campbell Michael; Christiana Diana; Smith David;

Cason Jana; Lee Alan

CORPORATE SOURCE: 1 Research Division, National Rehabilitation Hospital ,

Washington, District of Columbia.

SOURCE: Telemedicine journal and e-health: the official journal of the American Telemedicine Association, (2011 Oct) Vol. 17,

No. 8, pp. 662-5. Electronic Publication: 2011-07-26. Journal code: 100959949. E-ISSN: 1556-3669. L-ISSN:

1530-5627.

PUB. COUNTRY: United States

DOCUMENT TYPE: Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: NONMEDLINE; IN-DATA-REVIEW; IN-PROCESS; NONINDEXED;

Priority Journals

ENTRY DATE: Entered STN: 24 Sep 2011

Last Updated on STN: 24 Sep 2011

ABSTRACT:

[Formula: see text] Committee Members Co-Chairs David Brennan, M.B.E., Senior Research Engineer, National Rehabilitation Hospital, Washington, District of Columbia. Lyn Tindall, Ph.D., Speech-Language Pathologist, Department of Veterans Affairs Medical Center, Lexington, Kentucky. Writing Committee Janet Brown, M.A., CCC-SLP, Director, Health Care Services, American

Speech-Language-Hearing Association, Rockville, Maryland. Mike Campbell, M.S.,

M.B.A., CCC-SLP, Director of the Speech and Hearing Program, The University of North Carolina at Greensboro, Browns Summit, North Carolina. Jana Cason, D.H.S., OTR/L, Assistant Professor, Auerbach School of Occupational Therapy, Louisville, Kentucky. Diana Christiana, M.A.T., CCC-SLP, President/CEO, Clinical Communications, Sugar Land, TX. Alan Lee, Ph.D., P.T., D.P.T., C.W.S., G.C.S., Associate Professor, Mount St. Mary's College, Doctor of Physical Therapy Program, Los Angeles, California. David R. Smith, Director, Telehealth Resource Center, Marquette General Hospital, Marquette, Michigan. Deborah Theodoros, Ph.D., Associate Professor and Head, Division of Speech Pathology, School of Health and Rehabilitative Sciences, The University of Queensland, Queensland, Australia. Contributors Working Group Members [WG], Consultants [C], Reviewers [R], Telerehabilitation Special Interest Group Chair [TR], ATA Standards and Guidelines Committee Member [SG], and ATA Staff [S] Nina Antoniotti, R.N., M.B.A., Ph.D., Chair of SG, Director of Telehealth, Marshfield Clinic, Marshfield, Wisconsin. Jordana Bernard, M.B.A., S, Senior Director, Program Services, American Telemedicine Association, Washington, District of Columbia. Anne Burdick, M.D., M.P.H., SG, Associate Dean for Telemedicine and Clinical Outreach, Professor of Dermatology, Director, Leprosy Program, University of Miami Miller School of Medicine, Miami, Florida. Jerry Cavallerano, Ph.D., OD, SG, Staff Optometrist, Assistant to the Director, Joslin Diabetes Center, Beetham Eve Institute, Boston, Massachusetts. Ellen Cohn, Ph.D., CCC-SLP, TR, WG, Associate Dean for Instructional Development, School of Health and Rehabilitative Sciences, University of Pittsburgh, Pittsburgh, Pennsylvania. Paul Cox, MSEE, WG, President, PERL Research, Huntsville, Alabama. Mary Fran Delaune, P.T., MPT, R, Director, Practice Department, American Physical Therapy Association, Alexandria, Virginia. Matt Elrod, P.T., D.P.T., M.Ed., N.S.C., R, Associate Director, Practice Department, American Physical Therapy Association, Alexandria, Virginia. Brian Grady, M.D., [SG], VISN5 TMH Lead and Director, TeleMental Health, School of Medicine, University of Maryland, Baltimore, Maryland, Elizabeth Krupinski, Ph.D., Vice Chair of SG, Associate Director, Program Evaluation, University of Arizona, Arizona Telemedicine Program, Department of Radiology, Research Professor, Department of Radiology Research, Tucson, Arizona. Jonathan D. Linkous, MPA, S, Chief Executive Director, American Telemedicine Association, Washington, District of Columbia. Michael Pramuka, Ph.D., CRC, WG, Rehabilitation Counselor, Walter Reed Medical Center, Washington, District of Columbia. Richard Schein, Ph.D., WG, Postdoctoral Associate, Department of Rehabilitation Science and Technology, University of Pittsburgh, Pittsburgh, Pennsylvania. Lou Theurer, SG, Grant Administrator, Burn Telemedicine Program, University of Utah Health Sciences Center, Salt Lake City, Utah. Jill Winters, Ph.D., R.N., WG, SG, Dean and Professor, Columbia College of Nursing, Milwaukee, Wisconsin.

L50 ANSWER 9 OF 12 MEDLINE on STN ACCESSION NUMBER: 1974147497 MEDLINE Full-text DOCUMENT NUMBER: PubMed ID: 4274688

TITLE: High molecular-weight heparan sulfate from the cell

surface.

AUTHOR: Kraemer P M; Smith D A

SOURCE: Biochemical and biophysical research communications, (1974

Jan 23) Vol. 56, No. 2, pp. 423-30.

Journal code: 0372516. ISSN: 0006-291X. L-ISSN: 0006-291X.

PUB. COUNTRY: United States

DOCUMENT TYPE: Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 197406

ENTRY DATE: Entered STN: 10 Mar 1990

> Last Updated on STN: 10 Mar 1990 Entered Medline: 19 Jun 1974

CONTROLLED TERM: Check Tags: Female; Male

Animals

Binding Sites Borohydrides

Cell Line

Cell Membrane: AN, analysis

Chromatography, Gas Chromatography, Gel

Cricetinae Dialysis Glucosamine

*Glvcosaminoglvcans

Heparitin Sulfate: AN, analysis Heparitin Sulfate: PH, physiology

Hyaluronoglucosaminidase Hydrogen-Ion Concentration

Molecular Weight

Ovary Pronase

Sulfur Radioisotopes Testis: EN, enzymology

Tritium

Ultracentrifugation

CAS REGISTRY NO.: 10028-17-8 (Tritium); 3416-24-8 (Glucosamine); 9050-30-0

(Heparitin Sulfate)

CHEMICAL NAME: Borohydrides; Glycosaminoglycans; Sulfur Radioisotopes;

EC 3.2.1.35 (Ryaluronoglucosaminidase); EC 3.4.24.-

(Pronase)

OS.CITING REF COUNT: 7 There are 7 MEDLINE records that cite this record

L50 ANSWER 10 OF 12 BIOSIS COPYRIGHT (c) 2011 The Thomson Corporation on

STN ACCESSION NUMBER: 1970:222911 BIOSIS Full-text

DOCUMENT NUMBER: PREV197051132911; BA51:132911

TITLE: YIELD AND CHEMICAL COMPOSITION OF LEAVES AND STEMS OF ALFALFA-D AT INTERVALS UP THE SHOOTS.

AUTHOR(S): SMITH D

SOURCE: Journal of Agricultural and Food Chemistry, (1970) Vol. 18,

No. 4, pp. 652-656.

CODEN: JAFCAU. ISSN: 0021-8561.

DOCUMENT TYPE: Article

FILE SEGMENT: LANGUAGE: Unavailable

CONCEPT CODE: Biochemistry studies - General 10060

Biochemistry studies - Proteins, peptides and amino acids

10064

Biochemistry studies - Lipids 10066 Biochemistry studies - Carbohydrates 10068

Biochemistry studies - Minerals 10069

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Nutrition - General studies, nutritional status and methods
                   13202
                   Temperature - General measurement and methods 23001
                    Plant physiology - Temperature 51503
                    Plant physiology - Growth, differentiation
                                                               51510
                    Plant physiology - Chemical constituents 51522
                    Agronomy - Forage crops and fodder
INDEX TERMS:
                   Major Concepts
                       Agronomy (Agriculture); Biochemistry and Molecular
                       Biophysics; Development; Methods and Techniques;
                       Physiology
INDEX TERMS:
                   Miscellaneous Descriptors
                       SUGAR FIBER POTASSIUM BORON COPPER FAT CALCIUM
                       MAGNESIUM ALUMINUM BARIUM STRONTIUM MANGANESE STARCH
                      TEMPERATURE PHOSPHORUS IRON ASH ZINC
                   Classifier
ORGANISM:
                       Leguminosae
                                   26260
                    Super Taxa
                       Dicotyledones; Angiospermae; Spermatophyta; Plantae
                   Taxa Notes
                      Angiosperms, Dicots, Plants, Spermatophytes, Vascular
                       Plants
                    7440-09-7 (POTASSIUM)
REGISTRY NUMBER:
                    7440-42-8 (BORON)
                    7440-50-8 (COPPER)
                    7440-70-2 (CALCIUM)
                    7439-95-4 (MAGNESIUM)
                    7429-90-5 (ALUMINUM)
                   7440-39-3 (BARIUM)
                    7440-24-6 (STRONTIUM)
                    7439-96-5 (MANGANESE)
                    9005-25-8 (STARCH)
                    7723-14-0 (PHOSPHORUS)
                   7439-89-6 (IRON)
                   7440-66-6 (ZINC)
L50 ANSWER 11 OF 12 WPIX COPYRIGHT 2011
                                               THOMSON REUTERS on STN
ACCESSION NUMBER:
                    2005-182103 [200519] WPIX Full-text
DOC. NO. CPI:
                     C2005-058144 [200519]
DOC. NO. NON-CPI:
                    N2005-151918 [200519]
TITLE:
                     Ophthalmic device useful for monitoring an analyte,
                      e.g. glucose level in ocular fluid comprises a
                      hologram-based sensor
DERWENT CLASS:
                      A14; A96; B04; D22; P31; P84; P81; S05
INVENTOR:
                      DOMSCHKE A; DOMSCHKE A M; XIAODONG H; ZHOU J; ZHOU J
                     S: HU X
PATENT ASSIGNEE:
                     (NOVS-C) NOVARTIS AG; (NOVS-C) NOVARTIS PHARMA GMBH;
                     (DOMS-I) DOMSCHKE A M; (HUXX-I) HU X; (ZHOU-I) ZHOU J S;
                     (EYES-N) EYESENSE AG
COUNTRY COUNT:
                     107
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PATENT INFORMATION:

PATENT NO	KIND DATE	WEEK	LA PG	MAIN IPC
WO 2005015 EP 1651946 US 2007000 US 7927519	A1 20060 2470 A1 20070	217 (200519 503 (200629 104 (200703 419 (201128) EN) EN	1

APPLICATION DETAILS:

PATENT NO KIND	APPLICATION DATE
WO 2005015184 A1	WO 2004-EP6492 20040616
US 20070002470 A1 Provisional	US 2003-491014P 20030730
EP 1651946 A1	EP 2004-736912 20040616
EP 1651946 A1	WO 2004-EP6492 20040616
US 20070002470 A1	WO 2004-EP6492 20040616
US 20070002470 A1	US 2006-564323 20060803
US 7927519 B2 Provisional	US 2003-491014P 20030730
US 7927519 B2 PCT Application	WO 2004-EP6492 20040616
US 7927519 B2	US 2006-564323 20060803

FILING DETAILS:

EP 1651946 A1	Based on WO 2005015184 A
US 7927519 B2	Based on WO 2005015184 A
PRIORITY APPLN. INFO:	US 2003-491014P 20030730
	US 2006-564323 20060803
INT. PATENT CLASSIF .:	
IPC ORIGINAL:	G02B0007-02 [I,A]
IPC RECLASSIF.:	A61B0005-00 [N,A]; B29C0035-08 [I,A]; B29D0011-00 [I,A];
	G01N0021-77 [I,A]; G03H0001-00 [I,A]
ECLA:	A61B0005-145G; A61B0005-1455; B29C0035-08M2;
	B29D0011-00C4Y2; B29D0011-00J; G01N0021-75; G01N0021-77;
	G03H0001-00; G03H0001-04F
ICO:	K61B0005:145G; L29C0035:08B4; L29C0551:04;
	S01N0021:77B2G; S01N0021:77H2; S01N0021:77H4

NCLS: 264/001.310

PATENT NO KIND

USCLASS NCLM: NCLS: BASIC ABSTRACT:

WO 2005015184 A1 UPAB: 20110503

264/001.320; 359/819.000

NOVELTY - Ophthalmic device (D1) comprising a hologram-based sensor for monitoring an analyte level in ocular fluid, is new.

DETABLED DESCRIPTION - INDEPENDENT CLAIMS are also included for:

PATENT NO

(1) a fluid composition for making a biocompatible sensor (S1) containing a reflection hologram comprising at least one prepolymer, optionally a vinylic monomer, a molecular sensing group associated the prepolymer or vinylic monomer. The sensing group can interact or react with an analyte of interest to provide an optical signal which is indicative of a change in at least one optical property of the reflection holograms; and

(2) making (S1) involving introducing crosslinkable and/or polymerizable fluid material (M1) into a cavity formed by a mold, and producing and recording a pattern of interference fringes while polymerizing/crosslinking (M1) in the cavity to form the blocompatible sensor, thereby recording the pattern the sensor to form reflection hologram. The mold has a first mold half defining a first molding surface and a second mold half defining a second mold half are configured to receive each other such that the cavity is formed between the surfaces.

USE - The device is used for detecting/measuring an analyte (e.g. glucose) level in ocular fluid (claimed).

ADVANTAGE - Production of the biocompatible halographic sensor is cost effective. The device is biocompatible and stable over a long period of time.

MANUAL CODE: CFI: All-BOI; All-CO2B; Al2-V02A; Al2-V03C2; B04-B04L; B04-C03; B05-B01A; B10-A07A; B11-C04A; B11-C07B6; B11-C09B; B11-C09B

B11-C08B; B11-C08J; B12-K04A; D09-C01A EPI: S05-F05

- TI Ophthalmic device useful for monitoring an analyte, e.g. glucose level in ocular fluid comprises a hologram-based sensor
- TT TT: OPHTHALMIC DEVICE USEFUL MONITOR ANALYTE GLUCOSE LEVEL OCULAR
 FLUID COMPRISE HOLOGRAM BASED SENSE
- IN DOMSCHKE A; DOMSCHKE A M; XIAODONG H; ZHOU J; ZHOU J S; HU X
- NOV NOVELTY Ophthalmic device (D1) comprising a hologram-based sensor for monitoring an analyte level in ocular fluid, is new.
- DETD DETAILED DESCRIPTION INDEPENDENT CLAIMS are also included for:
 - (1) a fluid composition for making a biocompatible sensor (S1) containing a reflection hologram comprising at least one prepolymer, optionally a vinylic monomer, a molecular sensing group associated the prepolymer or vinylic monomer. The sensing group can interact or react with an analyte of interest to provide an optical signal which is indicative of a change in at least one optical property of the reflection holograms; and

(2) making (S1) involving introducing crosslinkable and/or polymerizable fluid material (M1) into a cavity formed by a mold, and producing and recording a pattern of interference fringes while polymerizing/crosslinking (M1) in the cavity to form the biocompatible sensor, thereby recording the pattern the sensor to form reflection hologram. The mold has a first mold half defining a first molding surface and a second mold half defining a second molding surface. The first and second mold half are configured to receive each other such that the cavity is formed between the surfaces.

USE

USE - The device is used for detecting/measuring an analyte (e.g. glucose) level in ocular fluid (claimed).

ADVANTAGE - Production of the biocompatible halographic sensor is cost effective. The device is biocompatible and stable over a long period of time.

TECH

INSTRUMENTATION AND TESTING - Preferred Device: (D1) is selected from contact lens, corneal onlay and implantable ophthalmic device (particularly implantable subconjunctival device). (D1) comprises a reflection hologram—based sensor comprising (M1) where a molecular sensing group is incorporated. The group can interact or react with the analyte of interest to provide a signal which is indicative of a change in

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at least one optical property of the reflection hologram.
Preferred Components: (M1) is a polymer obtainable by crosslinking and/or
polymerizing a prepolymer selected from: a) a polyhydroxy compound having
molecular weight of at least about 2000 and comprises poly(vinyl alcohol),
of units of formulae (I), (II) and/or (III) (0.5 to - 80%); b) a vinyl
group-terminated polyurethane which is obtained by reacting an
isocyanate-capped polyurethane with an ethytenlcally unsaturated amine
(preferably primary or secondary amine) or an ethylenically unsaturated
monohydroxy compound. The isocyanate-capped polyurethane is a
copolymerization product of at least one polyalkylene glycol, a compound
containing at least 2 hydroxyl groups, and at least one compound with at
least 2 isocyanate groups; or c) a derivative of a polyvinyl alcohol,
polyethyleneimine or polyvinylamine containing polyvinyl alcohol,
polyethyleneimine or polyvinylamine of the formula
-(CH2-CH(-R4-C(0)C(Ra)(Rb)-NHC(0)-C(Rc)=CH2))-(VI) and
-(CH2CH2N(C(0)C(Ra)(Rb)-NHC(0)-C(Rc)=CH2)) (VII) having average molecular
weight 2000 - 1,000,000.
(M1) is a polymer obtainable by crosslinking and/or polymerizing a
prepolymer selected from a crosslinkable polyacrylamide; a crosslinkable
statistical copolymer of a vinyllactam, methyl methacrylate and a
comonomer, a crosslinkable copolymer of a vinvllactam, vinvl acetate and
vinyl alcohol; a polyalkylene glycol polyether-polyester copolymer with
crosslinkable side chain; a branched polyalkylene glycol-urethane
prepolymer, a polyalkylene glyco/-tetra(meth)acrylate prepolymer; and a
crosslinkable polyallylamine gluconolactone prepolymer.
(M1) comprises a molecular sensing group that is obtainable by
crosslinking and/or polymerizing the crosslinkable and/or polymerizable
fluid material in an aqueous solution. The aqueous solution includes a low
molecular weight additive (preferably sodium chloride) which exhibit a
limited compatibility with a polymer matrix re-suited from the
crosslinkable and/or polymerizable fluid material, but good compatibility
with water. The additive is present in an amount sufficient to increase
refractive index differences between high and low irradiated areas
resulted from the pattern of interference fringes. The molecular sensing
group is a phenyl boronic acid formula (IV) - (VI). (M1) comprises a
water soluble prepolymer obtained by reacting an acryloylchloride or an
isocyanate group containing (meth)acrylate with polymerization product of
NH2 terminated polyalkylene glycol and di/polyisocyanates optionally in
the presence of triamine; or reacting an isocyanate-capped polyurethane
with ethylenically unsaturated monohydroxy compound. The isocyanate-capped
polyurethane is copolymerization product of at least one polyalkylene
glycol, a compound containing at least two hydroxyl groups and at least
one compound with at least two isocyanate groups.
R = 1-12C alkylene;
R1 = hydrogen or 1-7C lower alkyl;
R2 = olefinically unsaturated, electron-withdrawing, crosslinkable radical
having up to 25 carbon atoms (preferably CONH-(R5NHCOO)q-R6-OCOR4 or
(CONH-(R5NHCOO)g-R6-O)pCOR4);
R3 = hydrogen, 1-6C alkyl group or alkyl;
R7 = amino or quaternary amino group of the formula N+(R')3X-;
R' = hydrogen or 1-4C alkyl;
X = HSO4-, F-, Cl-, Br-, I-, CH3COO-, OH, BF- or H2PO4;
R8 = the radical of a mono/di/tribasic, saturated or unsaturated organic
acid or sulfonic acid;
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Ra and Rb = H, 1-8C alkyl, aryl or cyclohexyl;
     Rc = H \text{ or } 1-8C \text{ alkyl;}
     Rd = 0 \text{ or } -NH;
     R14 and R18 = olefinically unsaturated, crosslinkable radicals;
     R15 and R16 = 1-12C alkylene;
     R17 = 6-12C arylene, 6-12C saturated bivalent cycloaliphatic group, 7-14C
     arylenealkylene, 7-14C alkylenearylene or 13-16C arylenealkylenearyfene;
     p and q = 0 or 1;
     R5 and R6 = 2-8C lower alkylene or R17; and
     R4 = 2-24C olefinically unsaturated crosslinkable radical
     Preferred Method: Making (S1) containing a reflection hologram involving
     spraying at least one (M1) onto the first surface of an article and
     irradiating (M1) with at least two beams of coherent light. The spraying
    process selected from an air-assisted atomization and dispensing process.
    an ultrasonic-assisted atomization and dispensing process, a piezoelectric
    assisted atomization and dispensing process, an electro-mechanical jet
     printing process, a piezo-electric jet printing process, a plaza-electric
     with hydrostatic jet printing process, and a thermal jet printing process.
     Producing and recording occurs by irradiating (M1) with at least two beams
     of coherent light, where one of the two beams is directed to (M1) through
     the first molding surface and the other beam is directed to (M1) through
    at least a portion of the second molding surface. The two beams of
    coherent light form the pattern while polymerizing/crosslinking (M1) to
     form the biocompatible sensor. The pattern is recorded in (S1) to form
     the reflection hologram. (P1) additionally involves partially crosslinking
     and/or polymerizing (M1) by actinic irradiation, before the step of
     producing and recording. The partially crosslinking and/or polymerizing is
    performed by exposing (M1) to a UV light with an energy level sufficient
    high to initiate crosslinking and/or polymerizing but low enough not to
    completely cross-link and/or polymerize (M1); or by exposing (M1) to a UV
     light for a period of time short enough not to completely cross-link
     and/or polymerize (M1). The method also involves introducing a second (M1)
    into the cavity formed by the mold; and polymerizing/crosslinking in the
    cavity to form the biosensor. The coating has the reflection hologram is
     transferred from one of the molding surfaces into the biosensor and become
     an integral part of the biosensor during polymerizing/crosslinking of the
     second (M1) in the cavity
L50 ANSWER 12 OF 12 WPIX COPYRIGHT 2011
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2004-118884 [200412] WPIX Full-text
ACCESSION NUMBER:
                     2002-443846; 2002-674752; 2003-239210; 2009-A71011;
CROSS REFERENCE:
                     2009-H71947; 2010-B16679; 2010-E16114; 2010-H14318;
                     2010-M60977
TITLE:
                     Nanoscale articles useful as sensing elements for
                     microneedle probes for implantation into living subjects.
                     comprises free-standing and bulk-doped semiconductors
DERWENT CLASS:
                     A89; B04; D16; L03; Q68; S03; U11; U12
INVENTOR:
                     CUI Y; DUAN X; GUDIKSEN M; GUDIKSEN M S; HUANG Y; LAUHON
                     L J; LIANG W; LIEBER C M; PARK H; SMITH D C; WANG D;
                     WANG J; WEI O; ZHONG Z
                     (CUIY-I) CUI Y; (DUAN-I) DUAN X; (GUDI-I) GUDIKSEN M;
PATENT ASSIGNEE:
                     (HARD-C) HARVARD COLLEGE; (HUAN-I) HUANG Y; (LAUH-I)
                     LAUHON L J; (LIAN-I) LIANG W; (LIEB-I) LIEBER C M;
                     (PARK-I) PARK H; (SMIT-I) SMITH D C; (WANG-I) WANG D;
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(WANG-I) WANG J; (WEIQ-I) WEI Q; (ZHON-I) ZHONG Z

COUNTRY COUNT: 103

PATENT INFORMATION:

PAT	TENT NO	KIN	DATE	WEEK	LA	PG	MAIN IPC
WO AU	20030089899 2004038767 2003298525	A2 A1	20040506 20040513	(200468)	EN EN	162[78]	
	2003298525 7301199		20051103 20071127		EN		

APPLICATION DETAILS:

PATENT NO KIND	APPLICATION DATE
US 20030089899 Al Provisional US 20030089899 Al Provisional US 20030089899 Al Provisional	US 2000-254745P 20001211
US 20030089899 Al Provisional	
US 20030089899 A1 Provisional	US 2001-292045P 20010518
US 20030089899 A1 Provisional	US 2001-292121P 20010518
US 20030089899 A1 CIP of	
US 20030089899 Al Provisional	
US 20030089899 Al Provisional	
	US 2002-152490 20020520
	US 2002-196337 20020716
	AU 2003-298525 20030716
AU 2003298525 A8	AU 2003-298525 20030716
	WO 2003-US22061 20030716 US 2000-226835P 20000822
	US 2000-226835P 20000822 US 2000-254745P 20001211
US 7301199 B2 Provisional US 7301199 B2 Provisional	US 2001-291896P 20010518
US 7301199 B2 Provisional	
US 7301199 B2 Provisional	
	US 2001-935776 20010822
	US 2001-348313P 20011109
	US 2001-20004 20011211
US 7301199 B2 Provisional	
	US 2002-152490 20020520
US 7301199 B2	US 2002-196337 20020716
US 7301199 B2 Cont of	US 2002-196337 20020716

FILING DETAILS:

PATENT NO	KIND	PATENT NO
AU 2003298525 AU 2003298525 US 7301199 B2	A8 Based on	WO 2004038767 A WO 2004038767 A US 7129554 B
PRIORITY APPLN, INF	O: US 2002-196337	20020716

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US 2002-152490
                                          20020520
                     US 2002-354642P
                                          20020206
                     US 2001-348313P
                                          20011109
                     US 2001-935776
                                           20010822
                     US 2001-291896P
                                           20010518
                     US 2001-292035P
                                           20010518
                     US 2001-292045P
                                          20010518
                     US 2001-292121P
                                          20010518
                      US 2000-254745P
                                          20001211
                      US 2000-226835P
                                          20000822
                      US 2001-20004
                                          20011211
INT. PATENT CLASSIF . .
          MAIN:
                     B82B003-00
                     C30B011-00; C30B029-60
      SECONDARY:
  IPC ORIGINAL:
                     H01L0029-06 [I,A]; H01L0029-76 [I,A]; H01L0029-94 [I,A]
 IPC RECLASSIF.:
                     C30B0011-00 [I,A]; C30B0025-00 [I,A]; G01N0027-12 [I,A];
                     G01N0027-414 [I,A]; G01N0033-543 [I,A]; G01Q0070-00 [I,A]
                      ; G11C0011-56 [I,A]; G11C0013-02 [I,A]; H01L0023-532
                      [I,A]; H01L0029-06 [I,A]; H01L0029-20 [I,A];
                      H01L0029-207 [I,A]; H01L0029-267 [I,A]; H01L0029-73 [I,A]
                      ; H01L0031-0352 [I,A]; H01L0031-08 [I,A]; H01L0033-18
                      [N,A]; H01L0051-00 [N,A]; H01L0051-30 [I,A]
ECLA:
                     B82Y0010-00; B82Y0015-00; B82Y0030-00;
                     C30B0011-00+29/60D; C30B0025-00; C30B0025-00F;
                      C30B0025-00F+29/60D; C30B0029-60; G01N0027-414;
                     G01N0033-543K2; G11C0011-56; G11C0013-02N; G11C0013-02R3;
                     G11C0013-02R3B; H01L0023-532M3; H01L0029-06C6;
                     H01L0029-16G; H01L0029-20B; H01L0029-207; H01L0029-267;
                     H01L0029-73; H01L0031-0352; H01L0031-08; H01L0051-00M4D
                     L82T0201:01S; T01L0033:18; T01L0051:00M4D
USCLASS NCLM:
                      257/009.000
                      257/E23.165; 257/E29.081; 257/E29.093; 257/E29.174;
                      257/E31.032; 257/E31.052; 257/E33.003; 257/E33.005;
                      850/052.000
BASIC ABSTRACT:
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UPAB: 20090509 US 20030089899 A1

NOVELTY - A nanoscale article comprising a free-standing and bulk-doped semiconductor having at least one portion with smallest or maximum width of less than 500 nm, is new. The semiconductor is doped during growth.

DETAILED DESCRIPTION - INDEPENDENT CLAIMS are also included for:

- (1) a method of detecting analyte by contacting a nanoscopic wire with a sample suspected of containing analyte, e.g. DNA, and determining a change in a property of nanoscopic wire;
- (2) a sensor comprising at least one nanoscale wire, and a detector for measuring a change in a property of nanoscale wire;
- (3) a method of generating light by applying energy to semiconductor(s), causing the semiconductor to emit light;
 - (4) a method of assembling elongated structures on a surface by:
- (i) conditioning the surface with at least one functionality that attracts the elongated structures to particular positions on the surface, and aligning the elongated structures;
- (ii) depositing elongated structures on the surface, and electrically charging the surface to produce electrostatic forces between at least two of the elongated structures;

- (iii) dispersing the elongated structures on surface of a liquid-phase to form Langmuir-Blodgett film, compressing the film, and transferring the compressed film on a surface; or
- (iv) dispersing the elongated structures in a flexible matrix, stretching the flexible matrix in a direction to produce a shear force on the elongated structures to align the elongated structure in the direction, removing the flexible matrix, and transferring the aligned elongated structures to a surface;
- (5) a system for growing doped semiconductor comprising a mechanism for providing molecules of semiconductor substrate and molecules of dopant, and a mechanism for doping molecules of semiconductor with the molecules of dopant during growth of semiconductor;
 - (6) a system for assembling elongated structures; and
- (7) an analyte-gated field effect transistor useful as chemical and/or biological sensor, comprising a substrate of first insulating material, source and drain electrodes disposed on the substrate, a semiconductor nanowire disposed between the source and drain electrodes, and an analyte-specific binding agent disposed on surface of nanowire.
- USE The article is useful as sensing element for a microneedle probe implantable into a living subject. It is designed for use in semiconductor device; computational device; or electrical component such as Schottky diode, photodiode, light-emission source, single photon emitter, photoluminescent device, electroluminescent device, field effect transistor, bipolar junction transistor, single-electron transistor, rectifier, inverter, complimentary inverter, photodetector, p-n solar cell, single photon detector, tunnel diode, light-sensing device, gate, AND gate, NAND gate, OR gate, XOR gate, NOR gate, latch, flip-flop, register, switch, clock circuit, static memory device, dynamic memory device, programmable circuit, amplifier, analog circuit, active transistor, mixed signal device, signal processing circuit, or light-emitting diode. The article may be one of the nanoscopic wire sensors in a sensor array formed on surface of glass, polymer, or silicon dioxide-coated silicon substrate. The article may form sensing elements for an integrated dip-probe sensor, a plug-and-play sensor array (claimed)

ADVANTAGE - The article is capable of simultaneously monitoring physiological characteristics. It is capable of determining oxygen concentration, carbon dioxide concentration and/or glucose levels in a subject.

DESCRIPTION OF DRAWINGS - The figure is a schematic diagram showing nanoscale wire growth by vapor deposition in or on elongated template.

MANUAL CODE:

CPI: A12-E01; A12-V03C2; A12-W11L; B05-A01B; B05-A02; B05-A03A; B05-A03B; B05-B02A; B05-B02B; B05-B02C; B05-C03; B05-C04; B05-C08; B10-A07; B11-C04A; B11-C08E6; B12-K04A; D05-H09; D05-H10; D05-H12; D05-H18; L04-C02C; C04-B; L04-B01; L04-E01A; L04-E01D; L04-E03; L04-B05 EPI: S03-B15; U11-C01J3; U11-C02A; U11-C02J7; U11-C18B9; U12-R01B2

- TI Nanoscale articles useful as sensing elements for microneedle probes for implantation into living subjects, comprises free-standing and bulk-doped semiconductors
 IN CUI Y; DUAN X; GUDIKSEN M; GUDIKSEN M S; HUANG Y; LAUHON L J; LIANG W;
- LIEBER C M; PARK H; SMITH D C; WANG D; WANG J; WEI Q; ZHONG Z
 DETD DETAILED DESCRIPTION INDEPENDENT CLAIMS are also included for:
 - (1) a method of detecting analyte by contacting a nanoscopic wire with a sample suspected of containing analyte, e.g. DNA, and determining a

USE

change in a property of nanoscopic wire;

- (2) a sensor comprising at least one nanoscale wire, and a detector for measuring a change in a property of nanoscale wire;
- (3) a method of generating light by applying energy to semiconductor(s), causing the semiconductor to emit light;
 - (4) a method of assembling elongated structures on a surface by:
- (i) conditioning the surface with at least one functionality that attracts the elongated structures to particular positions on the surface, and aliquing the elongated structures;
- (ii) depositing elongated structures on the surface, and electrically charging the surface to produce electrostatic forces between at least two of the elongated structures;
- (iii) dispersing the elongated structures on surface of a liquid-phase to form Langmuir-Blodgett film, compressing the film, and transferring the compressed film on a surface; or
- (iv) dispersing the elongated structures in a flexible matrix, stretching the flexible matrix in a direction to produce a shear force on the elongated structures to align the elongated structure in the direction, removing the flexible matrix, and transferring the aligned elongated structures to a surface;
- (5) a system for growing doped semiconductor comprising a mechanism for providing molecules of semiconductor substrate and molecules of dopant, and a mechanism for doping molecules of semiconductor with the molecules of dopant during growth of semiconductor;
 - (6) a system for assembling elongated structures; and
- (7) an analyte-gated field effect transistor useful as chemical and/or biological sensor, comprising a substrate of first insulating material, source and drain electrodes disposed on the substrate, a semiconductor nanowire disposed between the source and drain electrodes, and an analyte-specific binding agent disposed on surface of nanowire.

USE - The article is useful as sensing element for a microneedle probe implantable into a living subject. It is designed for use in semiconductor device; computational device; or electrical component such as Schottky diode, photodiode, light-emission source, single photon emitter, photoluminescent device, electroluminescent device, field effect transistor, bipolar junction transistor, single-electron transistor, rectifier, inverter, complimentary inverter, photodetector, p-n solar cell, single photon detector, tunnel diode, light-sensing device, gate, AND gate, NAND gate, NAND gate, NAND gate, NAND gate, latch, flip-flop, register, switch, clock circuit, static memory device, dynamic memory device, programmable circuit, amplifier, analog circuit, active transistor, mixed signal device, signal processing circuit, or light-emitting diode. The article may be one of the nanoscopic wire sensors in a sensor array formed on surface of glass, polymer, or

elements for an integrated dip-probe sensor, a plug-and-play sensor array (claimed)

ADV ADVANTAGE - The article is capable of simultaneously monitoring physiological characteristics. It is capable of determining oxygen concentration, carbon dioxide concentration and/or glucose levels in a subject.

silicon dioxide-coated silicon substrate. The article may form sensing

TECH
INSTRUMENTATION AND TESTING - Preferred Components: The article includes a

core comprising a first semiconductor and at least one shell comprising a semiconductor different from the first semiconductor. The shell surrounds a portion of the core. The core induces a change in the shell. The shell may be an atomic monolayer or an inductive shell. The shell is delta-doped, polarizable, ferromagnetic, mechanically inducible, oxidizable, reducible, or photoactivatable. The article has more than one light-producing region. The semiconductor is elongated and has a longitudinal section. The ratio of length of longitudinal section to a longest width is greater than 4:1 (preferably greater than 1000:1). The semiconductor is a single-walled, multi-walled or unmodified nanoscopic wire having a diameter of 0.5-200 nm and an aspect ratio of more than 2, or a nanotube. It comprises a single crystal. Preferred Dimensions: The shell has a thickness of less than 5 (preferably less than 1) nm. The smallest width is less than 200 (preferably less than 5) nm. The semiconductor has an aspect ratio of at least 100:1. Preferred Method: Additional material(s) is applied to a surface of doped semiconductor. The doped semiconductor is grown by applying energy to a collection of molecules including molecules of semiconductor and molecules of dopant. Doping of semiconductor includes controlling an extent of doping by controlling the ratio of an amount of semiconductor molecules to an amount of dopant molecules, vaporizing the molecules using a laser to form vaporized molecules, condensing the vaporized molecules into liquid crystal, and growing the semiconductor from the liquid cluster. ORGANIC CHEMISTRY - Preferred Components: The shell comprises functional moiety which can be activated by light, amorphous oxide, and/or reaction entity such as nucleic acid, antibody, sugar, carbohydrate, protein, or catalyst. The functional moiety is hydroxy, formyl, carboxy, sulfo, cyano, amino, mercapto, thiocarboxy, oxycarbonyl and/or halide, methyl, hydrazide, aldehyde, aryl azide, fluorinated aryl azide, or benzophenone. POLYMERS - Preferred Components: The functional moiety may comprise a polymer chain having a chain length of less than the diameter of nanoscopic wire. The polymer is polyamide, polyester, polyimide, and/or polyacrylic. The functional moiety comprises a thin coating of metallic element, oxide, sulfide, nitride, selenide, polymer, or polymer gel covering the surface of nanoscopic wire. INORGANIC CHEMISTRY - Preferred Material: The semiconductor comprises elemental semiconductor, e.g. silicon, germanium, tin, selenium, tellurium, boron, diamond, or phosphorus; or solid solution of elemental semiconductor, e.g. boron carbide, boron hexaphosphide, boron silicide, silicon carbide, silicon germanium, silicon tin, or germanium tin. The semiconductor may comprise IV-IV semiconductor, e.g. silicon carbide; III-V semiconductor, e.g. boron nitride/boron phosphide/boron arsenide, aluminum nitride/aluminum phosphide/aluminum arsenide/aluminum antimonide, gallium nitride/gallium phosphide/gallium arsenide/gallium antimonide, and/or indium nitride/indium phosphide/indium arsenide/indium antimonide; II-VI semiconductor, e.g. zinc oxide/zinc sulfide/zinc selenide/zinc telluride, cadmium sulfide/cadmium selenide/cadmium telluride, mercury sulfide/mercury selenide/mercury telluride, or beryllium sulfide/beryllium selenide/beryllium telluride/magnesium sulfide/magnesium selenide; IV-VI semiconductor, e.g. germanium sulfide, germanium selenide, germanium telluride, tin sulfide, tin selenide, tin telluride, lead oxide, lead sulfide, lead selenide, or lead telluride; I-VII semiconductor, e.g. copper fluoride, copper chloride, copper bromide, copper iodide, silver fluoride, silver chloride, silver bromide, or silver iodide; and beryllium silicon nitride (BeSiN2), calcium carbon nitride (CaCN2), zinc germanium phosphide (ZnGeP2), cadmium tin arsenide (CdSnAs2), zinc tin antimonide (ZnSnSb2), copper germanium phosphide (CuGeP3), copper silicon phosphide (CuSi2P3), (Cu, Ag) (Al, Ga, In, Tl, Fe)(S, Se, Te)2, silicon nitride (Si3M4), germanium nitride (Ge3M4), aluminum oxide, (Al, Ga, In)2(S, Se, Te)3, or aluminum carbon oxide (Al2CO). The semiconductor comprises p-type dopant, e.g. boxon, aluminum, indium, magnesium, zinc, cadmium, mercury, carbon, or silicon; and/or n-type dopant, e.g. phosphorus, arsenide, antimony, silicon, germanium, tin, sulfur, selenium, or tellurium.

=> file registry

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DICTIONARY FILE UPDATES: 17 OCT 2011 HIGHEST RN 1337015-67-4

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=> file zcaplus

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FILE COVERS 1907 - 18 Oct 2011 VOL 155 ISS 17
FILE LAST UPDATED: 17 Oct 2011 (20111017/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Aug 2011
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Aug 2011

ZCAplus now includes complete International Patent Classification (IPC) reclassification data for the second quarter of 2011.

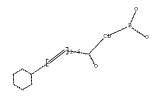
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This file contains CAS Registry Numbers for easy and accurate substance identification.



Structure attributes must be viewed using STN Express query preparation. L9 26 SEA FILE=REGISTRY SSS FUL L7 L10 STR



Structure attributes must be viewed using STN Express query preparation.

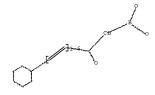
L12 16 SEA FILE=REGISTRY SUB=L9 SSS FUL L10 L13

18 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L12

=> d stat que L27 L7 STR



Structure attributes must be viewed using STN Express query preparation. L9 $_{\rm 26}$ SEA FILE-REGISTRY SSS FUL L7 $_{\rm L10}$ STR



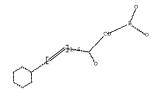
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Structure attributes must be viewed using STN Express query preparation.
L12
            16 SEA FILE-REGISTRY SUB-L9 SSS FUL L10
L13
            18 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L12
L14
             2 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON GLUCOSE/CN
L15
        699325 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L14 OR ?GLUCOS?
L16
        325389 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON ?SACCHARID?
L17
           490 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON
                                                      ?OPTHALM?
L18
         32416 SEA FILE-ZCAPLUS SPE-ON ABB-ON PLU-ON
                                                       ?OPHTHALM?
L19
        166464 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON
                                                      EYE
L20
        717477 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON ?FLUORESC?
L21
       1438953 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON OPTIC?
L22
        444462 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON SENSOR?
L23
         25592 SEA FILE-ZCAPLUS SPE-ON ABB-ON PLU-ON OCULAR?
L24
          7970 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON CONTACT LEN?
       2633827 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON POLYMER?
L26
        125187 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON SENSING?
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L27 11 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L13 AND (L15 OR L16 OR L17 OR L18 OR L19 OR L20 OR L21 OR L22 OR L23 OR L24 OR L25 OR L26)

=> d stat que L30 L7 STF



Structure attributes must be viewed using STN Express query preparation. L9 $_{\rm 26}$ SEA FILE=REGISTRY SSS FUL L7 L10 STR



Structure attributes must be viewed using STN Express query preparation.

L12 16 SEA FILE=REGISTRY SUB=L9 SSS FUL L10
L13 18 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L12

L29 411686 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON SUGAR?

L30 4 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L13 AND L29

 \Rightarrow d stat que L31 L7 STR



Structure attributes must be viewed using STN Express query preparation. L9 $_{\rm 26}$ SEA FILE-REGISTRY SSS FUL L7 L10 $_{\rm STR}$

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Structure attributes must be viewed using STN Express query preparation.
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L12 16 SEA FILE=REGISTRY SUB=L9 SSS FUL L10

L13 18 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L12

L31 4 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON BLOOD? AND L13

=> s L13 or L27 or L30 or L31

L51 18 L13 OR L27 OR L30 OR L31

=> d iall hitstr L51 1-18

L51 ANSWER 1 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2011:236462 ZCAPLUS Full-text DOCUMENT NUMBER: 154:284035

ENTRY DATE: Entered STN: 24 Feb 2011

TITLE: Preparation of boronic acid derivatives of resveratrol

for use in the treatment of cancer

INVENTOR(S): Brown, Milton L.; Yenugonda, Venkata M.; Kong, Yali

PATENT ASSIGNEE(S): Georgetown University, USA

SOURCE: PCT Int. Appl., 85pp. CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English

CLASSIFICATION: 26-9 (Biomolecules and Their Synthetic Analogs)

Section cross-reference(s): 1, 29, 63

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION: DATENT NO

PATEN	T I	.00			KIN	D	DATE			APPL	ICAT	I NOI	NO.		D	ATE		
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WO 20	0110	0225	02		A1		2011	0224		WO 2	010-	US45	925		2	0100	818	
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		CA,	CH,	CL,	CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DO,	DZ,	EC,	EE,	EG,	
		ES,	FI,	GB,	GD,	GE,	GH,	GM,	GT,	HN,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	
		KE,	KG,	KM,	KN,	KP,	KR,	KZ,	LA,	LC,	LK,	LR,	LS,	LT,	LU,	LY,	MA,	
		MD,	ME,	MG,	MK,	MN,	MW,	MX,	MY,	MZ,	NA,	NG,	NI,	NO,	NZ,	OM,	PE,	
		PG,	PH,	PL,	PT,	RO,	RS,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SM,	ST,	SV,	
		SY,	TH,	TJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	ZA,	ZM,	ZW
F	RW:	AL,	AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	HR,	
		HU,	ΙE,	IS,	IT,	LT,	LU,	LV,	MC,	MK,	MT,	NL,	NO,	PL,	PT,	RO,	SE,	
		SI,	SK,	SM,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	
		NE,	SN,	TD,	TG,	BW,	GH,	GM,	KE,	LR,	LS,	MW,	ΜZ,	NA,	SD,	SL,	SZ,	
		TZ,	UG,	ZM,	ZW,	AM,	AZ,	BY,	KG,	KZ,	MD,	RU,	TJ,	TM				
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WO 2011022502 IPCI C07F0005-02 [I,A]; A61K0031-69 [I,A] IPCR C07F0005-02 [I,A]; A61K0031-69 [I,A]

ECLA C07F0005-02C

OTHER SOURCE(S): CASREACT 154:284035; MARPAT 154:284035

GRAPHIC IMAGE:

III

ABSTRACT:

Disclosed are compds. A-L-C [A is (un)substituted cycloalkyl, aryl, heteroaryl, heterocyclyl; L is present or absent, when present L is a linker; C is an (un) substituted cycloalkyl, aryl, heteroaryl, heterocyclyl, wherein at least one position in the compds. is substituted with B(OH)2, and at least one position in the compound is substituted with alkoxy, alkoxydialkylamino or hydroxy], or a pharmaceutically acceptable, salt, prodrug, clathrate, tautomer or solvate thereof and methods related to boronic acid derivs. of resveratrol. Further disclosed are compds. I [R1 - R5 are independently, H, B(OH)2, mild Lewis acid, strong acid, weak acid, alkyl, alkenyl, alkynyl, halogen, alkoxy, NH2, alkylamino, dialkylamino, CN, NO2, CHO, CO2H, alkoxycarbonyl, alkoxydialkylamino, aminocarbonyl, alkylaminocarbonyl, dialkylaminocarbonyl, haloalkyl, haloalkoxy, haloalkylamino, di(haloalkyl)amino or sugars; L, when present, is C1-6-alkyl, C2-6-alkenyl, aryl, heteroaryl, cycloalkyl, heterocyclyl, P-O-S; wherein P = C1-6-alkyl, C2-6-alkenyl, aryl, heteroaryl, cycloalkyl, heterocyclyl; Q is NR11, O, S, C(:O); R11 = H or C1-3-alkyl; S is present or absent and when present is C1-6-alkyl, C2-6-alkenyl, aryl, heteroaryl, cycloalkyl or heterocyclyl] and II [R6 - R10 are independently, H, B(OH)2, mild Lewis acid, strong acid, weak acid, alkyl, alkenyl, alkynyl, halogen, alkoxy, NH2, alkylamino, dialkylamino, CN, NO2, CHO, CO2H, alkoxycarbonyl, alkoxydialkylamino, aminocarbonyl, alkylaminocarbonyl, dialkylaminocarbonyl, haloalkyl, haloalkoxy, haloalkylamino, di(haloalkyl)amino or sugars; R8 and R9 are optionally cyclized to form cycloalkyl, aryl, heteroaryl or heterocyclyl, optionally substituted with B(OH)2]. Thus, YK-5-104 (III) was prepared from 3,5-dimethoxybenzyl bromide via phosphinylation with PPh3 in THF; Wittig reaction with pinacolborane IV in THF containing BuLi/hexane; flash column chromatog. separation of cis/trans isomers; and O-demethylation/de-esterification with BBr3 in CH2Cl2. Certain of these derivs. have enhanced efficacy relative to resveratrol, function as irreversible modulators, and act at the G1/S phase of the cell cycle. The biol. and pharmacol. activity of III was determined [GI50 = 36.6 µM WST-1 cell survival assay; GI50 = 31.10 µM vs. breast cancer (MCF-7 cell line); GI50 = 49.09 vs. breast cancer (multidrug resistant CL 10.3 cell line)].

SUPPL. TERM:

irreversible modulator; cell cycle irreversible modulator resveratrol boronic acid deriv INDEX TERM: Multidrug resistance (CL 10.3 cell line; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Cell cycle regulatory proteins ROLE: BSU (Biological study, unclassified); BIOL (Biological study) (G1, expression level, irreversible modulators; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Interphase (cell cycle) (G1/S boundary, irreversible modulators; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Animal cell line (MCF-7; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Antiproliferative agents (irreversible; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Antitumor agents Apoptosis Human Mammary gland, neoplasm Neoplasm (preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Boronic acids ROLE: PAC (Pharmacological activity); RCT (Reactant); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent); USES (Uses) (resveratrol derivs.; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Natural products (stilbenes, resveratrol boronic acid derivs.; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Estrogen receptors ROLE: BSU (Biological study, unclassified); BIOL (Biological study) (a, expression of, by cancer; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: Estrogen receptors ROLE: BSU (Biological study, unclassified); BIOL (Biological $(\beta$, expression of, by cancer; preparation of boronic acid

resveratrol boronic acid deriv prepn cancer treatment

derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: 637-69-4P, 4-Methoxystyrene ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (Suzuki coupling reaction of, with (iodophenyl)boronate derivative; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) 128376-64-7 INDEX TERM: ROLE: RCT (Reactant); RACT (Reactant or reagent) (Wittig reaction of, with benzylphosphonate or benzylphosphonium bromide derivs.; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: 877-88-3, 3,5-Dimethoxybenzyl bromide ROLE: RCT (Reactant); RACT (Reactant or reagent) (phosphinylation or phosphitylation of; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: 344324-66-9 ROLE: RCT (Reactant); RACT (Reactant or reagent) (phosphitylation of; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: 24131-30-4P, (3,5-Dimethoxybenzyl)triphenylphosphonium bromide ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (prepared and Wittig reaction of, with (formylphenyl)boronic acid pinacol ester; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: 108957-75-1P, Diethyl (3,5-dimethoxybenzyl)phosphonate ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (preparation and Horner-Wadsworth-Emmons reaction of, with (formylphenyl)boronate; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: 868629-88-3P ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (preparation and O-demethylation/deesterification of; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer) INDEX TERM: 1073338-99-4P ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (preparation and Suzuki coupling reaction of, with vinylanisole; preparation of boronic acid derivs. of resveratrol for use in the treatment of cancer)

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INDEX TERM:
                  1266682-75-0P
                   ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
                   (Preparation); RACT (Reactant or reagent)
                      (preparation and deesterification of; preparation of boronic
acid
                      derivs. of resveratrol for use in the treatment of
                      cancer)
INDEX TERM:
                   1266682-77-2P
                   ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
                   (Preparation); RACT (Reactant or reagent)
                      (preparation and deesterification or
                      O-demethylation/deesterification of; preparation of boronic
                     acid derivs. of resveratrol for use in the treatment of
                      cancer)
INDEX TERM:
                   1266682-73-8P
                                  1266682-74-9P
                                                  1266682-76-1P
                   1266682-78-3P 1266682-79-4P
                                                  1266682-80-7P
                   1266682-81-8P 1266682-82-9P 1266682-83-0P
                   1266682-84-1P
                                  1266682-85-2P
                   1266682-86-3P
                                 1266682-87-4P 1266682-88-5P
                   1266682-89-6P 1266682-90-9P 1266682-91-0P
                   1266682-92-1P
                   ROLE: PAC (Pharmacological activity); SPN (Synthetic
                  preparation); THU (Therapeutic use); BIOL (Biological
                   study); PREP (Preparation); USES (Uses)
                      (preparation of boronic acid derivs, of resveratrol for use
                      the treatment of cancer)
INDEX TERM:
                   501-36-0, Resveratrol 33069-62-4, Paclitaxel
                   ROLE: PAC (Pharmacological activity); THU (Therapeutic use);
                   BIOL (Biological study); USES (Uses)
                      (preparation of boronic acid derivs. of resveratrol for use
in
                     the treatment of cancer)
REFERENCE COUNT:
                        THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS
                        RECORD.
                   (1) Das. B: TETRAHEDRON LETTERS 2009, P3031 ZCAPLUS
REFERENCE(S):
                   (2) Dicesare, N; JOURNAL OF PHYSICS CHEMISTRY A 2001, P6834
                             ZCAPLUS
                   (3) Japan Science & Tech Agency; JP 2005306865 A 2005
                            ZCAPLUS
                   (4) venugonda, V; ACS: "Abstract of Papers, 238th ACS
                            National Meeting" 2009
                    1266682-85-2P 1266682-86-3P
    1266682-84-1P
     RL: PAC (Pharmacological activity); SPN (Synthetic preparation); THU
     (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES
     (Uses)
        (preparation of boronic acid derivs. of resveratrol for use in the
treatment
       of cancer)
RN
    1266682-84-1 ZCAPLUS
     Boronic acid, B-[3-[(2E)-3-(3,5-dimethoxyphenyl)-1-oxo-2-propen-1-
     vllphenvll- (CA INDEX NAME)
```

Double bond geometry as shown.

RN 1266682-85-2 ZCAPLUS

CN Boronic acid, B-[3-[(2E)-1-oxo-3-(3,4,5-trimethoxyphenyl)-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

- RN 1266682-86-3 ZCAPLUS
- CN Boronic acid, B-[4-[(2E)-3-(3,5-dimethoxyphenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

L51 ANSWER 2 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2010:82926 ZCAPLUS Full-text

DOCUMENT NUMBER: ENTRY DATE:

TITLE:

AUTHOR(S):

Entered STN: 21 Jan 2010 A boronic acid chalcone analog of combretastatin $A\!-\!4$ as a potent anti-proliferation agent

Kong, Yali; Wang, Kan; Edler, Michael C.; Hamel, Ernest; Mooberry, Susan L.; Paige, Mikell A.; Brown,

Milton L.

152:350293

CORPORATE SOURCE: Department of Oncology, Drug Discovery Program,

Georgetown University Medical Center, Washington, DC,

20057, USA

SOURCE: Bioorganic & Medicinal Chemistry (2010), 18(2),

971-977

CODEN: BMECEP; ISSN: 0968-0896

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

CLASSIFICATION: 1-3 (Pharmacology)

Section cross-reference(s): 25

CASREACT 152:350293 OTHER SOURCE(S):

ABSTRACT:

Chalcones represent a class of natural products that inhibits tubulin assembly. In this study we designed and synthesized boronic acid analogs of chalcones in an effort to compare biol. activities with combretastatin A-4, a potent inhibitor of tubulin polymerization Systematic evaluation of the positional

effects

INDEX TERM:

of the carbonyl moiety towards inhibition of tubulin polymerization, cancer cell proliferation and angiogenesis revealed that placement of the carbonyl adjacent to the trimethoxybenzene A-ring resulted in more active compds. than when the carbonyl group was placed adjacent to the C-ring. Our study identified a boronic acid chalcone with inhibition towards 16 human cancer cell lines in the 10-200 nM range, and another three cell lines with GI50-values below 10 nM. Furthermore, this drug has significant anti-angiogenesis effects demonstrated by HUVEC tube formation and aortic ring assay.

SUPPL. TERM: boronic acid chalcone analog combretastatin prepn antitumor

angiogenesis structure INDEX TERM: Antiangiogenic agents

Antitumor agents

Central nervous system, neoplasm

Colon neoplasm

Human

Kidney, neoplasm

Mammary gland, neoplasm

Melanoma

Non-small-cell lung carcinoma Structure-activity relationship

Ovary, neoplasm

Prostate gland, neoplasm

(boronic acid chalcone analogs of combretastatin A-4

preparation as potential antitumor agents)

INDEX TERM: Tubulins

ROLE: BSU (Biological study, unclassified); BIOL (Biological

study)

(boronic acid chalcone analogs of combretastatin A-4

preparation as potential antitumor agents)

181644-49-5P 1215281-19-8P 1215281-21-2P

1215281-22-3P

ROLE: DMA (Drug mechanism of action); PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES

(Uses)

(boronic acid chalcone analogs of combretastatin A-4 preparation as potential antitumor agents) INDEX TERM: 117048-59-6, Combretastatin A-4 ROLE: DMA (Drug mechanism of action); PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (boronic acid chalcone analogs of combretastatin A-4 preparation as potential antitumor agents) INDEX TERM: 86-81-7 578-57-4 621-59-0 1131-62-0 1136-86-3 127972-02-5 ROLE: RCT (Reactant); RACT (Reactant or reagent) (boronic acid chalcone analogs of combretastatin A-4 preparation as potential antitumor agents) INDEX TERM: 6100-74-9P 35310-75-9P 81224-34-2P 1215281-20-1P ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (boronic acid chalcone analogs of combretastatin A-4 preparation as potential antitumor agents) INDEX TERM: 866824-84-2 ROLE: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (boronic acid chalcone analogs of combretastatin A-4 preparation as potential antitumor agents) THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (6 OS.CITING REF COUNT: 6 CITINGS) DATE LAST CITED: Date last citing reference entered STN: 12 Sep 2011 OS.CITING.REFS: CAPLUS 2011:993054; 2011:979897; 2011:567539; 2011:443531; 2010:1085552; 2010:1058412 REFERENCE COUNT: THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. REFERENCE(S): (1) Anon; http://dtp.nci.nih.gov/branches/btb/ivclsp.html (2) Autom, R; Cancer Lett 1995, V97, P33 (3) Brossi, A; J Org Chem 1967, V32, P1269 ZCAPLUS (4) Dhar, D; J Org Chem 1958, V23, P1159 ZCAPLUS (5) Dicesare, N; Tetrahedron Lett 2002, V43, P2615 ZCAPLUS (6) Ducki, S; Bioorg Med Chem Lett 1998, V8, P1051 MEDLINE (7) Edwards, M; J Med Chem 1990, V33, P1948 ZCAPLUS (8) Hamel, E; Cell Biochem Biophys 2003, V38, P1 ZCAPLUS (9) Ikeda, S; JP 08188546 1996 ZCAPLUS (10) Kong, Y; Chem Biol 2005, V12, P1007 ZCAPLUS (11) Kumar, S; J Med Chem 2003, V46, P2813 ZCAPLUS (12) Laali, K; J Org Chem 1993, V58, P1385 ZCAPLUS (13) Lin, C; Biochemistry 1989, V28, P6984 ZCAPLUS (14) Monks, A; J Natl Cancer Inst 1991, V83, P757 ZCAPLUS (15) Nielsen, S; Bioorg Med Chem 2004, V12, P3047 ZCAPLUS (16) Pettit, G; Can J Chem 1982, V60, P1374 ZCAPLUS (17) Skehan, P; J Natl Cancer Inst 1990, V82, P1107 ZCAPLUS (18) Tinley, T; Cancer Res 2003, V63, P3211 ZCAPLUS (19) Verdier-Pinard, P; Mol Pharmacol 1998, V53, P62 ZCAPLUS (20) Won, S; Eur J Med Chem 2005, V40, P103 ZCAPLUS (21) Zi, X; Cancer Res 2005, V65(8), P3479 ZCAPLUS 1215281-21-2P RL: DMA (Drug mechanism of action); PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study);

PREP (Preparation); USES (Uses)
(boronic acid chalcone analogs of combretastatin A-4 preparation as potential antitumor agents)

RN 1215281-21-2 ZCAPLUS

CN Boronic acid, B-[2-methoxy-5-[(2E)-1-oxo-3-(3,4,5-trimethoxyphenyl)-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

L51 ANSWER 3 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2010:51187 ZCAPLUS Full-text

DOCUMENT NUMBER: 152:169553

ENTRY DATE: Entered STN: 14 Jan 2010

TITLE: Trapping glucose probe in pores of polymer

INVENTOR(S): Chow, Pei Yong Edwin; Ying, Jackie Y.

PATENT ASSIGNEE(S): Agency for Science, Technology and Research, Singapore SOURCE: PCT Int. Appl. 42pp.

SOURCE: PCT Int. Appl., 42pp.
CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

CLASSIFICATION: 35-4 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 38

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PA:	TENT	NO.			KIN	D	DATE			APPL	ICAT	I NOI	.00		D.	ATE	
WO	2010	0053	98		A1	_	2010	0114		WO 2	009-	SG24	5		2	0090	
	W:	ΑE,	AG,	AL,	AM,	AO,	AT,	AU,	AZ,	BA,	BB,	BG,	BH,	BR,	BW,	BY,	BZ,
		CA,	CH,	CL,	CN,	co,	CR,	CU,	CZ,	DE,	DK,	DM,	DO,	DZ,	EC,	EE,	EG,
		ES,	FI,	GB,	GD,	GE,	GH,	GM,	GT,	HN,	HR,	HU,	ID,	IL,	IN,	IS,	JP,
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		MD,	ME,	MG,	MK,	MN,	MW,	MX,	MY,	MZ,	NA,	NG,	NI,	NO,	NZ,	OM,	PE,
		PG,	PH,	PL,	PT,	RO,	RS,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SM,	ST,	SV,
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		SK,	SM,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,
		SN,	TD,	TG,	BW,	GH,	GM,	KΕ,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,
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IE, IS, IT, LI, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, SE,
             SI, SK, SM, TR, AL, BA, RS
     US 20110136929
                        A1 20110609
                                           US 2011-2800
                                                                  20110106
                                           US 2008-129646P P 20080709
WO 2009-SG245 W 20090709
PRIORITY APPLN. INFO.:
PATENT CLASSIFICATION CODES:
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 WO 2010005398 IPCI C08F0002-32 [I,A]; G01N0021-64 [I,A]; A61B0005-00
                       [I,A]; G02C0007-04 [I,A]
                 IPCR C08F0002-32 [I,A]; A61B0005-00 [I,A]; G01N0021-64
                       [I,A]; G02C0007-04 [I,A]
                 ECLA G01N0021-77; A61B0005-145G; A61B0005-1455; C08F0002-22;
                       S01N0021:77B2G; S01N0021:77H6; S01N0021:78
 EP 2318443
                IPCI C08F0002-32 [I,A]; G01N0021-64 [I,A]; A61B0005-00
                       [I.A]: G02C0007-04 [I.A]
                 IPCR C08F0002-32 [I,A]; A61B0005-00 [I,A]; G01N0021-64
                       [I,A]; G02C0007-04 [I,A]
                 ECLA G01N0021-77; A61B0005-145G; A61B0005-1455; C08F0002-22;
                       S01N0021:77B2G; S01N0021:77H6; S01N0021:78
US 20110136929 IPCI C08J0009-00 [I,A]; C08L0033-10 [I,A]
                NCL 521/105.000; 521/149.000
ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT
OTHER SOURCE(S):
                       MARPAT 152:169553
ABSTRACT:
This invention relates to a polymer matrix defining pores that is formed by
polymerizing polymer precursors in a precursor solution The precursor solution
comprises a bicontinuous microemulsion of a first fluid in a first continuous
phase and a second fluid in a second continuous phase. The first fluid
comprises the polymer precursors. The second fluid comprises the glucose
probe. Some internal pores are connected to surface pores in the matrix
through openings sized to allow passage of glucose mols. but restrict passage
of the glucose probe. As the glucose probe is dispersed in the precursor
solution prior to polymerization, some glucose probe mols. are trapped in the
internal pores after polymerization The formed polymer may be used in an
ophthalmic device such as contact lens, for detecting the presence of
glucose in an ocular fluid.
SUPPL. TERM:
                  contact lenses porous membrane; glucose probe; boronic
                  acid; radical polymn
INDEX TERM:
                  Microemulsions
                      (bicontinuous; trapping glucose probe in pores
                      of polymer)
INDEX TERM:
                  Apparatus
                      (ophthalmic: trapping glucose probe
                      in pores of polymer)
INDEX TERM:
                  Porous materials
                     (polymer; trapping glucose probe in
                      pores of polymer)
INDEX TERM:
                  Membranes, nonbiological
                      (polymeric; trapping glucose probe in
                      pores of polymer)
INDEX TERM:
                  Polymerization
                      (radical; trapping glucose probe in pores of
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10/566406

polymer) INDEX TERM: Contact lenses (trapping glucose probe in pores of polymer) INDEX TERM: 97-90-5, Ethyleneglycol dimethacrylate ROLE: RCT (Reactant); RACT (Reactant or reagent) (crosslinker; trapping glucose probe in pores of polymer) INDEX TERM: 24650-42-8, 2,2-Dimethoxy-2-phenyl acetophenone ROLE: CAT (Catalyst use); USES (Uses) (initiator; trapping glucose probe in pores of polymer) INDEX TERM: 197648-90-1 ROLE: RCT (Reactant); RACT (Reactant or reagent) (macromonomer; trapping glucose probe in pores of polymer) INDEX TERM: 80-62-6, Methyl methacrylate 868-77-9 ROLE: RCT (Reactant); RACT (Reactant or reagent) (trapping glucose probe in pores of polymer) INDEX TERM: 94-41-7, 1,3-Diphenvlprop-2-en-1-one 614-57-3, 1,5-Diphenylpenta-2,4-dien-1-one 13780-71-7D, Boronic acid, unsubstituted or alkyl, aryl or combination substituted boronic acid 58367-01-4, Glucose 406719-92-4, Chalc-1 ROLE: TEM (Technical or engineered material use); USES (Uses) (trapping glucose probe in pores of polymer) REFERENCE COUNT: THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. REFERENCE(S): (1) Agency For Science Technology And Research; WO 2006014138 A1 2006 ZCAPLUS (2) Chapoy; US 20070030443 A1 2007 (3) Geddes; US 20070020182 A1 2007 ZCAPLUS (4) March: US 20030045783 A1 2003 (5) Suri; US 20080027245 A1 2008 ZCAPLUS 58367-01-4, Glucose 406719-92-4, Chalc-1 RL: TEM (Technical or engineered material use); USES (Uses) (trapping glucose probe in pores of polymer) 58367-01-4 ZCAPLUS RN CN Glucose (CA INDEX NAME)

Relative stereochemistry.

RN 406719-92-4 ZCAPLUS

CN Boronic acid, B-[4-[3-[4-(dimethylamino)phenyl]-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

L51 ANSWER 4 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2009:1339654 ZCAPLUS Full-text DOCUMENT NUMBER: 151:564335

ENTRY DATE: Entered STN: 02 Nov 2009

TITLE: Pharmacophore Modeling for Qualitative Prediction of

Antiestrogenic Activity

Brogi, Simone; Kladi, Maria; Vagias, Constantinos; AUTHOR(S):

Papazafiri, Panagiota; Roussis, Vassilios; Tafi,

Andrea

CORPORATE SOURCE: Dipartimento Farmaco Chimico Tecnologico, Universita

degli Studi di Siena, Siena, I-53100, Italy

SOURCE: Journal of Chemical Information and Modeling (2009), 49(11), 2489-2497

CODEN: JCISD8; ISSN: 1549-9596

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

CLASSIFICATION: 1-3 (Pharmacology)

ABSTRACT:

A ligand-based pharmacophore approach for the prediction of antiestrogenic activity to be used as an in silico screening tool for bioactive compds. including natural products was developed using Catalyst HypoGen. The generated pharmacophore hypothesis (HYPO-7) consisted of five features, namely, one hydrophobic (HY1), two hydrophobic aromatic (HY2), one hydrogen-bond acceptor (HBA), and one hydrogen-bond donor (HBD). HYPO-7 successfully predicted the lack of cytotoxicity of a number of new metabolites isolated from the red alga Laurencia glandulifera. Furthermore, a screening of the Asinex Gold Collection database was performed by coupling HYPO-7 with a docking filtration, which resulted in a restricted set of 12 new scaffolds to be investigated as potential SERMs. The inhibitory activity of these compds. was evaluated in vitro using MCF7 human breast adenocarcinoma cell line. Ten out of the twelve compds. exhibited inhibitory activity with IC50 values between 26 and 188 uM. This result shows that application of HYPO-7 could assist in the selection of potentially active compds., thus expediting the hit discovery process.

SUPPL. TERM: pharmacophore modeling antiestrogen drug screening discovery

INDEX TERM: Mammary gland, neoplasm

(adenocarcinoma; pharmacophore modeling for qual.

prediction of antiestrogenic activity)

INDEX TERM: Estrogen receptors

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ROLE: BSU (Biological study, unclassified); BIOL (Biological
                  study)
                     (binding to; pharmacophore modeling for qual. prediction
                     of antiestrogenic activity)
INDEX TERM:
                  Human
                     (cell line; pharmacophore modeling for qual. prediction
                     of antiestrogenic activity)
INDEX TERM:
                  Protein motifs
                     (ligand binding domain, of estrogen receptor;
                     pharmacophore modeling for qual. prediction of
                     antiestrogenic activity)
INDEX TERM:
                  Adenocarcinoma
                     (mammary adenocarcinoma; pharmacophore modeling for qual.
                     prediction of antiestrogenic activity)
INDEX TERM:
                  Antiestrogens
                  Antitumor agents
                  Drug discovery
                  Hydrogen bond
                  Hydrophobicity
                  Laurencia glandulifera
                  Molecular modeling
                  Natural products, pharmaceutical
                  Pharmacophores
                  Selective estrogen receptor modulators
                     (pharmacophore modeling for qual. prediction of
                     antiestrogenic activity)
INDEX TERM:
                  Drug screening
                     (virtual; pharmacophore modeling for qual, prediction of
                     antiestrogenic activity)
INDEX TERM:
                  446-72-0 486-66-8 614-46-0
                                                 7728-73-6
                                                             16982-00-6
                  57766-60-6 57794-64-6 59403-81-5 63676-22-2
                  84449-90-1, Raloxifene
                                         86161-35-5
                                                      104714-15-0
                  124526-56-3, [1,1':3',1''-Terphenyl]-4,4''-diol
                  139407-74-2 168766-36-7 302952-72-3 306752-82-9
                  315707-07-4 330674-11-8 331271-35-3 335206-32-1
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                                1197188-96-7
                                              1197188-97-8 1197188-98-9
                  ROLE: PAC (Pharmacological activity); THU (Therapeutic use);
                  BIOL (Biological study); USES (Uses)
                     (pharmacophore modeling for qual. prediction of
                     antiestrogenic activity)
OS.CITING REF COUNT: 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS RECORD (4
                        CITINGS)
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DATE LAST CITED: Date last citing reference entered STN: 12 Sep 2011 OS.CITING.REFS: CAPLUS 2011:898875; 2011:918943; 2011:704341; 2010:809214 REFERENCE COUNT: 39 THERE ARE 39 CITED REFERENCES AVAILABLE FOR THIS RECORD.

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 - 562823-90-9 562823-91-0 562823-92-1
- 562823-93-2 1197188-80-9
 - RL: PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses)
 - (pharmacophore modeling for qual. prediction of antiestrogenic

10/566406

activity)

RN 562823-90-9 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3,4-dichlorophenyl)-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-91-0 ZCAPLUS

CN Boronic acid, B=[4-[(2E)-3-(3-chloro-4-fluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-92-1 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3,4-difluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

$$\mathsf{H} \circ \mathsf{B} = \mathsf{F}$$

- RN 562823-93-2 ZCAPLUS
- CN Boronic acid, B-[4-[(2E)-3-(3-bromo-4-fluoropheny1)-1-oxo-2-propen-1-y1]pheny1]- (CA INDEX NAME)

Double bond geometry as shown.

1197188-80-9 ZCAPLUS RN

CN Boronic acid, B-[4-[(2E)-3-(3-bromo-4-iodophenyl)-1-oxo-2-propen-1vl]phenvl]- (CA INDEX NAME)

Double bond geometry as shown.

L51 ANSWER 5 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2009:480379 ZCAPLUS Full-text

DOCUMENT NUMBER: 151:48488

ENTRY DATE: Entered STN: 22 Apr 2009 TITLE: Chalcones in cancer: understanding their role in terms

of OSAR

AUTHOR(S): Katsori, A.-M.; Hadjipavlou-Litina, D.

CORPORATE SOURCE: Department of Pharmaceutical Chemistry, School of Pharmacy, Aristotle University of Thessaloniki,

Thessaloniki, 54124, Greece

SOURCE: Current Medicinal Chemistry (2009), 16(9), 1062-1081

CODEN: CMCHE7; ISSN: 0929-8673 Bentham Science Publishers Ltd.

PUBLISHER: DOCUMENT TYPE: Journal

LANGUAGE: English

ABSTRACT:

1-3 (Pharmacology) CLASSIFICATION:

Chalcones are a group of plant-derived polyphenolic compds. belonging to the flavonoids family and possess a wide variety of cytoprotective and modulatory functions. The results showed that chalcones inhibit the proliferation of MCF-7 and MDA-MB-231 by inducing apoptosis and blocking cell cycle progression in the G2/M phase. Immunoblot assay showed that chalcones significantly decreased the expression of cyclin B1, cyclin A and Cdc2 protein, as well as increased the expression of p21 and p27 in a p53-independent manner,

contributing to cell cycle arrest. In this research we tried to review the anticancer effect of chalcones derivs., and to evaluate new QSARs which will help in the understanding of the role of chalcones and of their analogs on cancer. Simultaneously a comparative study will be presented. Our QSAR results reveal that in almost all cases, the clog P parameter plays an important part in the QSAR relationships (linear or bilinear model). In some cases the steric factors such as the CMR or the substituents MR (linear) are important. Electronic effects are comparatively unimportant. The study shows that log P as calculated from the Clog P program is suitable for this form of QSAR study.

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SUPPL. TERM:
                chalcone cancer OSAR
INDEX TERM:
                Antitumor agents
                Human
                Mammary gland, neoplasm
                 QSAR (quantitative structure-activity relationship)
                   (chalcones role in cancer in terms of OSAR)
INDEX TERM:
                 94-41-7D, Chalcone, analogs 1896-62-4 10596-48-2
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                             780014-67-7 786616-56-6 876338-36-2
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                 ROLE: PAC (Pharmacological activity); THU (Therapeutic use);
                BIOL (Biological study); USES (Uses)
                   (chalcones role in cancer in terms of QSAR)
```

OS.CITING REF COUNT: 13 THERE ARE 13 CAPLUS RECORDS THAT CITE THIS RECORD (13

2011:567539; 2010:1305186; 2010:1232937; 2010:1023598;

CITINGS)

DATE LAST CITED: Date last citing reference entered STN: 03 Oct 2011

OS.CITING.REFS: CAPLUS 2011:1190265; 2011:1077241; 2011:1001367; 2011:743876;

2010:831193;	2010:893598;	2010:262014;	2010:26105;
2009:1390139			

REFERENCE COUNT:

59 THERE ARE 59 CITED REFERENCES AVAILABLE FOR THIS RECORD.

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- 562823-92-1 562823-93-2
 - 265853-35-1 265853-33-5
 - RL: PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses)
 - (chalcones role in cancer in terms of QSAR)
- RN 562823-84-1 ZCAPLUS
- CN Boronic acid, B-[4-[(2E)-3-(4-iodophenyl)-1-oxo-2-propen-1-yl]phenyl](CA INDEX NAME)

Double bond geometry as shown.

- RN 562823-90-9 ZCAPLUS
- CN Boronic acid, B-[4-[(2E)-3-(3,4-dichlorophenyl)-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-91-0 ZCAPLUS

CN Boronic acid, B=[4-[(2E)-3-(3-chloro-4-fluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-92-1 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3,4-difluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-93-2 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3-bromo-4-fluorophenyl)-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

ACCESSION NUMBER: 2007:1393130 ZCAPLUS Full-text

DOCUMENT NUMBER: 148:417386 ENTRY DATE: Entered STN: 07 Dec 2007

Trans-4-lodo, 4'-boranyl-chalcone induces antitumor TITLE:

activity against malignant glioma cell lines in vitro

and in vivo AUTHOR(S):

Sasayama, Takashi; Tanaka, Kazuhiro; Mizukawa, Katsu; Kawamura, Atsufumi; Kondoh, Takeshi; Hosoda, Kohkichi;

Kohmura, Eiji

English

Department of Neurosurgery, Kobe University Graduate CORPORATE SOURCE: School of Medicine, 7-5-1, Kusunoki-cho, Chuo-ku,

Kobe, 650-0017, Japan

SOURCE: Journal of Neuro-Oncology (2007), 85(2), 123-132

CODEN: JNODD2; ISSN: 0167-594X

PUBLISHER: Springer DOCUMENT TYPE: Journal

1-6 (Pharmacology) CLASSIFICATION:

LANGUAGE: ABSTRACT:

SUPPL. TERM:

INDEX TERM:

Chalcones are considered the precursors of flavonoids and have been identified as interesting compds, with antitumor properties. Boronic-chalcone derivs, are more toxic to breast cancer cells compared to normal breast cells. Here, we studied the antitumor activities of trans-4-lodo, 4'-boranyl-chalcone (TLBC), which is a boronic-chalcone derivative, in several glioma cell lines. TLBC showed a dose-dependent inhibition with inhibitory concentration 50% value in the µM range (5.5-25.5 uM) in various glioma cell lines. Flow cytometric and western blot assay demonstrated that TLBC induced apoptosis independent of changes to the tumor suppressor p53. This cytotoxic effect was the caspase-dependent manner. Also, TLBC lowered levels of anti-apoptotic Bcl-2 and/or Bcl-XL protein in several of the cell lines. To examine the antitumor effect of TLBC in vivo, we used a malignant glioma xenograft model. This result showed that in the mice treated with TLBC at 20 mg/kg, mean tumor volume was reduced by 43.9% (P < 0.01) in comparison with the control group. Immunohistochem. and western

blot anal. showed that Bcl-2 protein levels were decreased and Bax protein levels were slightly increased in the tumors injected with 20 mg/kg TLBC compared with the control tumors. Therefore, we conclude that TLBC may be a potential chemotherapeutic agent for human glioma.

apoptosis INDEX TERM: Gene, animal

ROLE: BSU (Biological study, unclassified); BIOL (Biological

study)

(Bc1-2; trans-4-lodo, 4'-boranvl-chalcone reduced mRNA

level of Bc1-2 in human glioblastoma cell)

anticancer trans lodo boranyl chalcone glioblastoma

INDEX TERM: Bcl-x proteins

ROLE: BSU (Biological study, unclassified); BIOL (Biological

(Bcl-xL; trans-4-lodo, 4'-boranyl-chalcone reduced mRNA

level of Bc1-XL in human glioblastoma cell)

Neuroglia, neoplasm

(glioblastoma; trans-4-lodo, 4'-boranyl-chalcone induced cell cycle arrest, apoptosis and decreased antiapoptotic Bc1-2, XL protein expression in human glioblastoma cell

and in mouse with glioblastoma xenograft) INDEX TERM: p53 (protein) ROLE: BSU (Biological study, unclassified); BIOL (Biological study) (trans-4-lodo, 4'-boranyl-chalcone induced cell cycle arrest and induced apoptosis independent of p53 expression in human malignant glioblastoma cell) INDEX TERM: Cytotoxic agents Human (trans-4-lodo, 4'-boranvl-chalcone induced cell cycle arrest, apoptosis and decreased antiapoptotic Bc1-2 as well as Bcl-XL protein expression in human glioblastoma cell) INDEX TERM: Bcl-2 proteins ROLE: BSU (Biological study, unclassified); BIOL (Biological study) (trans-4-lodo, 4'-boranyl-chalcone induced cell cycle arrest, apoptosis and decreased antiapoptotic Bc1-2 as well as Bcl-XL protein expression in human glioblastoma cell) INDEX TERM: Apoptosis (trans-4-lodo, 4'-boranyl-chalcone induced cell cycle arrest, apoptosis and decreased antiapoptotic Bcl-2, XL protein expression in human glioblastoma cell and in mouse with glioblastoma xenograft) INDEX TERM: Cell death (trans-4-lodo, 4'-boranyl-chalcone induced cell death in human glioblastoma cell) INDEX TERM: Antitumor agents Proliferation inhibition (trans-4-lodo, 4'-boranyl-chalcone reduced tumor volume in mouse with human glioblastoma xenograft) INDEX TERM: Bax proteins ROLE: BSU (Biological study, unclassified); BIOL (Biological (trans-4-lodo, 4'-boranyl-chalcone slightly increased Bax protein expression in human glioblastoma cell and in mouse with glioblastoma xenograft) INDEX TERM: 562823-84-1 ROLE: PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (trans-4-lodo, 4'-boranyl-chalcone induced cell cycle arrest, apoptosis and decreased antiapoptotic Bc1-2 as well as Bcl-XL protein expression in human glioblastoma cell) OS.CITING REF COUNT: 6 THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (6 CITINGS) DATE LAST CITED: Date last citing reference entered STN: 22 Dec 2010 OS.CITING.REFS: CAPLUS 2010:1454435; 2010:1346932; 2010:1162550; 2009:541802; 2008:1164081; 2008:1074659 REFERENCE COUNT: THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS 26 RECORD. REFERENCE(S): (1) Behin, A; Lancet 2003, V361, P323 (2) Bratton, S; Exp Cell Res 2000, V256, P27 ZCAPLUS

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- 562823-84-1
- RL: PAC (Pharmacological activity); THU (Therapeutic use); BIOL
 - (Biological study); USES (Uses)
 - (trans-4-lodo, 4'-boranyl-chalcone induced cell cycle arrest, apoptosis and decreased antiapoptotic Bcl-2 as well as Bcl-XL protein expression in human glioblastoma cell)
- RN 562823-84-1 ZCAPLUS
 - Boronic acid, B-[4-[(2E)-3-(4-iodophenyl)-1-oxo-2-propen-1-yl]phenyl]-(CA INDEX NAME)

Double bond geometry as shown.

L51 ANSWER 7 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2007:90998 ZCAPLUS Full-text DOCUMENT NUMBER: 146:158818

ENTRY DATE: Entered STN: 26 Jan 2007 TITLE: Quaternary nitrogen heterocyclic compounds for detecting aqueous monosaccharides in physiological fluids INVENTOR(S): Geddes, Chris D.; Badugu, Ramachandram; Lakowicz, Joseph R. PATENT ASSIGNEE(S): University of Maryland Biotechnology Institute, USA SOURCE: U.S. Pat. Appl. Publ., 72 pp., Cont.-in-part of Appl. No. PCT/US2004/022717. CODEN: USXXCO DOCUMENT TYPE: Patent LANGUAGE: English CLASSIFICATION: 9-16 (Biochemical Methods) Section cross-reference(s): 14, 63 FAMILY ACC. NUM. COUNT: 2 PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE A1 20070125 US 2005-318663 US 20070020182 20051227 US 7718804 B2 20100518 WO 2005000109 A2 20050106 WO 2004-US22717 20040628 WO 2005000109 A2 20050100 WO 2005000109 A3 20050310 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG US 20100297016 A1 20101125 US 2010-781899 US 2010-781899 20100518 US 2003-483124P P 20030627 US 2003-483202P P 20030627 WO 2004-US22717 A2 20040628 US 2005-318663 A3 20051227 20100518 PRIORITY APPLN. INFO.: PATENT CLASSIFICATION CODES:

PATENT NO.		PATENT FAMILY CLASSIFICATION CODES
US 20070020182	INCL	424009600; 546013000; 544229000; 436095000
	IPCI	A61K0049-00 [I,A]; C07F0005-02 [I,A]; G01N0033-00
		[I,A]; C07D0215-38 [I,A]
	IPCR	A61K0049-00 [I,A]; A61B [I,S]; A61K0031-44 [I,A];
		A61K0031-47 [I,A]; C07D0213-02 [I,A]; C07D0215-00
		[I,A]; C07F0005-02 [I,A]; G01N0033-00 [I,A];
		C07D0215-38 [I,A]
	NCL	424/009.600; 436/095.000; 544/229.000; 546/013.000;
		546/171.000
	ECLA	A61K0049-00P4F4C; C07F0005-02C
WO 2005000109	IPCI	A61B [ICM, 7]
	IPCR	A61B [I,S]; A61K0031-44 [I,A]; A61K0031-47 [I,A];
		A61K0049-00 [I,A]; C07D0213-02 [I,A]; C07D0215-00

[I,A]; C07F0005-02 [I,A]
ECLA A61K0049-00P4F4C; C07F0005-02C
A61K0049-00 [I,A]; C07F0005-02 [I,A]; C07H0001-00
[I,A]; G01N0032-76 [I,A]; C12C0001-02 [I,A];
G01N0033-00 [I,A]
IPCR A61K0049-00 [I,A]; A61B [I,S]; A61K0031-44 [I,A];
A61K0031-47 [I,A]; C07F00213-02 [I,A]; C07F00215-00
[I,A]; C07F0005-02 [I,A]; C07H0001-00 [I,A];
C12C0001-02 [I,A]; G01N0021-76 [I,A]; G01N0033-00 [I,A]
NCL 424/009.100; 435/029.000; 436/035.000; 436/172.000;
536/001.100; 544/29.000; 546/013.000

ECLA A61k0049-00P4F4C; C07F0005-02C
ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT ABSTRACT:

Quaternary nitrogen heterocyclic boronic acid-containing compds. are described, which are sensitive to glucoae and fructose, as well as a variety of other physiol. important analytes, such as aqueous chloride and iodide, and a method of using the compds. Also disclosed is a contact lens doped with the quaternary nitrogen heterocyclic boronic acid-containing compound, and a method of using the doped contact lens to measure the concentration of analyte in tears under physiol. conditions.

SUPPL. TERM: quaternary nitrogen heterocyclic compd monosaccharide body fluid INDEX TERM: Eve (Ocular fluid; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Medical goods (Ophthalmic, implantable; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Medical goods Sensors (Ophthalmic; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids)

INDEX TERM: Fluorometry
(Ratiometric; quaternary nitrogen heterocyclic compds.
for detecting aqueous monosaccharides in physiol.

fluids)
INDEX TERM: Eye

INDEX TERM.

(aqueous humor; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol.

fluids)

INDEX TERM: Fluorescent substances

(boronic acid-containing; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in

physiol. fluids) Boronic acids

ROLE: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation);

USES (Uses)

(fluorophores containing; quaternary nitrogen heterocyclic

physiol. fluids) INDEX TERM: Body fluid (interstitial; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Heterocyclic compounds ROLE: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (nitrogen, Quaternary; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Heterocyclic compounds ROLE: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (nitrogen, boronic acid-containing, quaternary; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Blood analysis Body fluid Contact lenses Containers Fluorescence Fluorometry Optical anisotropy Permeation Sensors Solutions Tear Test kits UV and visible spectroscopy (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Monosaccharides ROLE: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Polymers ROLE: TEM (Technical or engineered material use); USES (Uses) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: 50-99-7, D-Glucose, analysis 57-48-7, D-Fructose, analysis 16887-00-6, Chloride, analysis 20461-54-5, Iodide, analysis ROLE: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: 920491-56-1 920491-57-2 920491-58-3 ROLE: ARG (Analytical reagent use); ANST (Analytical study);

compds. for detecting aqueous monosaccharides in

INDEX TERM:

USES (Uses)

(quaternary nitrogen heterocyclic compds. for detecting

aqueous monosaccharides in physiol. fluids)

INDEX TERM: 406719-91-3 406719-92-4 406719-93-5

406719-94-6 406719-95-7

ROLE: ARG (Analytical reagent use); PRP (Properties); ANST

(Analytical study); USES (Uses)

(quaternary nitrogen heterocyclic compds. for detecting

aqueous monosaccharides in physiol. fluids)

63816-10-4P 784146-24-3P 784146-26-5P 784146-27-6P

784146-28-7P

ROLE: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

(quaternary nitrogen heterocyclic compds. for detecting

aqueous monosaccharides in physiol. fluids)

INDEX TERM: 91-62-3, 6-Methylquinoline 100-39-0, Benzyl bromide 5263-87-6, 6-Methoxyquinoline 51323-43-4, m-Boronobenzyl

5253-67-6, 6-Methoxyquinoiine 51523-45-4, M-Boronobenzyi bromide 68162-47-0, p-Boronobenzyi bromide 91983-14-1, o-Boronobenzyi bromide

o-Boronobenzyl bromide

ROLE: RCT (Reactant); RACT (Reactant or reagent)

(quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids)

INDEX TERM: 784146-23-2P 784146-25-4P 819814-02-3P

ROLE: SPN (Synthetic preparation); PREP (Preparation) (quaternary nitrogen heterocyclic compds. for detecting

aqueous monosaccharides in physiol. fluids)
OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2
CITINGS)

DATE LAST CITED: Date last citing reference entered STN: 01 Apr 2011 OS.CITING.REFS: CAPLUS 2011:370337; 2010:51187

IT 50-99-7, D-Glucose, analysis

RL: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study)

(quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids)

RN 50-99-7 ZCAPLUS

CN D-Glucose (CA INDEX NAME)

Absolute stereochemistry.

IT 406719-92-4 406719-94-6

RL: ARG (Analytical reagent use); PRP (Properties); ANST (Analytical study); USES (Uses)

(quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids)

RN 406719-92-4 ZCAPLUS

Boronic acid, B-[4-[3-[4-(dimethylamino)phenyl]-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

406719-94-6 ZCAPLUS RN

Boronic acid, B-[4-[5-[4-(dimethylamino)phenyl]-1-oxo-2,4-pentadien-1vl]phenvl]- (CA INDEX NAME)

L51 ANSWER 8 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2005:1114291 ZCAPLUS Full-text

DOCUMENT NUMBER: 145:58663

ENTRY DATE: Entered STN: 18 Oct 2005

TITLE: A glucose-sensing contact lens: a new approach to noninvasive continuous physiological glucose

monitoring

AUTHOR(S): Badugu, Ramachandram; Lakowicz, Joseph R.; Geddes,

Chris D.

Cent. fluorescence Spectroscopy, Dep. Biochem. & Mol. CORPORATE SOURCE: Biol., Univ. of Maryland School of Medicine, MD,

21201, USA Proceedings of SPIE-The International Society for

SOURCE: Optical Engineering (2004), 5317(Optical Fibers and Sensors for Medical Applications IV), 234-245

CODEN: PSISDG; ISSN: 0277-786X

SPIE-The International Society for Optical Engineering

PUBLISHER: DOCUMENT TYPE: Journal

LANGUAGE: English

CLASSIFICATION: 9-16 (Biochemical Methods)

ABSTRACT:

The authors have developed a new technol. for the non-invasive continuous monitoring of tear glucose using a daily use, disposable contact lens, embedded with sugar-sensing boronic acid containing fluorophores. The authors' findings show that the authors' approach may be suitable for the continuous monitoring of tear glucose levels in the range 50-500 µM, which track blood glucose levels that are typically ≈ 5-10-fold higher. The

SUPPL. TERM:

authors initially tested the sensing concept with well-established, previously published, boronic acid probes and the results could conclude the used probes, with higher pKa values, are almost insensitive toward glucose within the contact lens, attributed to the low pH and polarity inside the lens. Subsequently, the authors have developed a range of probes based on the quinolinium backbone, having considerably lower pKa values, which enables them to be suitable to sense the physiol. glucose in the acidic pH contact lens. Herein the authors describe the results based on the authors' findings towards the development of glucose sensing contact lens and therefore an approach to non-invasive continuous monitoring of tear glucose using a contact lens.

glucose sensor contact lens tear boronic acid

INDEX TERM: Contact lenses Fluorescent substances Glucose sensors Tear (ocular fluid) (noninvasive continuous physiol. glucose monitoring in contact lens) 50-99-7, D-Glucose, analysis 57-48-7, INDEX TERM: D-Fructose, analysis ROLE: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study) (noninvasive continuous physiol, glucose monitoring in contact lens) INDEX TERM: 162254-07-1, ANDBA 406719-91-3, CSTBA 406719-92-4, Chalc 1 406719-93-5, DSTBA 406719-95-7 890653-41-5 890653-42-6 ROLE: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses) (noninvasive continuous physiol. glucose monitoring in contact lens) INDEX TERM: 98-80-6. PhenvlBoronic acid ROLE: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses) (noninvasive continuous physiol. glucose monitoring in contact lens using boronic acid probes) INDEX TERM: 22559-70-2, Quinolinium ROLE: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses) (noninvasive continuous physiol. glucose monitoring in contact lens using

quinolinium probes)
OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1
CTINGS)

DATE LAST CITED: Date last citing reference entered STN: 16 Feb 2009 OS.CITING.REFS: CAPLUS 2005:150225

REFERENCE COUNT: 55 THERE ARE 55 CITED REFERENCES AVAILABLE FOR THIS RECORD.

REFERENCE(S): (1) Badugu, R; Anal Chem. In press

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physiological glucose using a monosaccharide-sensing contact lens-2, manuscript under preparation

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- 50-99-7, D-Glucose, analysis
 - RL: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study)
 - (noninvasive continuous physiol. glucose monitoring in contact lens)
- 50-99-7 ZCAPLUS RN
- D-Glucose (CA INDEX NAME) CN

Absolute stereochemistry.

- IT 406719-92-4, Chalc 1
 - RL: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses)
 - (noninvasive continuous physiol. glucose monitoring in contact lens)
- RN 406719-92-4 ZCAPLUS
- CN Boronic acid, B-[4-[3-[4-(dimethylamino)phenyl]-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

L51 ANSWER 9 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2005:952799 ZCAPLUS Full-text DOCUMENT NUMBER: 143:398794

ENTRY DATE: Entered STN: 01 Sep 2005

TITLE: Monitoring the Effects of Antagonists on

Protein-Protein Interactions with NMR Spectroscopy AUTHOR(S): D'Silva, Lovola; Ozdowy, Przemyslaw; Krajewski, Marcin; Rothweiler, Ulli; Singh, Mahavir; Holak, Tad

CORPORATE SOURCE: Max Planck Institute for Biochemistry, Martinsried,

D-82152, Germany

Journal of the American Chemical Society (2005), SOURCE:

127(38), 13220-13226

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

CLASSIFICATION: 1-1 (Pharmacology)

Section cross-reference(s): 9

ABSTRACT:

We describe an NMR method that directly monitors the influence of ligands on protein-protein interactions. For a two-protein interaction complex, the size of one component should be small enough (less than .apprx.15 kDa) to provide a good quality 15N (13C) HSQC spectrum after 15N(13C) labeling. The size of the second unlabeled component should be large enough so that the mol. weight of the preformed complex is larger than .apprx. 40 kDa. When the smaller protein binds to a larger one, broadening of NMR resonances results in the disappearance of most of its cross-peaks in the HSQC spectrum. Addition of an antagonist that can dissociate the complex would restore the HSOC spectrum of the smaller component. The method directly shows whether an antagonist releases proteins in their wild-type folded states or whether it induces their denaturation, partial unfolding, or precipitation We illustrate the method by

studvina lead compds. that have recently been reported to block the MDM2-p53 interaction. Activation of p53 in tumor cells by inhibiting its interaction with MDM2 offers new strategy for cancer therapy.

protein interaction antagonist monitoring NMR spectroscopy; SUPPL. TERM:

MDM2 p53 interaction inhibitor cancer therapy protein unfolding

INDEX TERM: p53 (protein)

ROLE: BSU (Biological study, unclassified); BIOL (Biological

study)

(MDM2 complexes; monitoring effects of antagonists on protein-protein interactions with NMR spectroscopy)

INDEX TERM: Mdm2 protein

p53 (protein) ROLE: BSU (Biological study, unclassified); PRP

(Properties); BIOL (Biological study)

(conformation; monitoring effects of antagonists on protein-protein interactions with NMR spectroscopy)

INDEX TERM: Conformation

(folded; monitoring effects of antagonists on

INDEX TERM: Antitumor agents Drug screening NMR spectroscopy Neoplasm Protein unfolding Protein-protein interaction (monitoring effects of antagonists on protein-protein interactions with NMR spectroscopy) INDEX TERM: Mdm2 protein ROLE: BSU (Biological study, unclassified); BIOL (Biological study) (p53 complexes; monitoring effects of antagonists on protein-protein interactions with NMR spectroscopy) INDEX TERM: Conformation (protein; monitoring effects of antagonists on protein-protein interactions with NMR spectroscopy) INDEX TERM: 59541-35-4, NSC 279287 548472-68-0, Nutlin-3 562823-90-9 ROLE: PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (monitoring effects of antagonists on protein-protein interactions with NMR spectroscopy) OS.CITING REF COUNT: 25 THERE ARE 25 CAPLUS RECORDS THAT CITE THIS RECORD (26 CITINGS) DATE LAST CITED: Date last citing reference entered STN: 22 Apr 2011 OS.CITING.REFS: CAPLUS 2011:472182; 2010:1396327; 2010:933915; 2010:882825; 2010:694488; 2009:1171980; 2009:1193603; 2009:915670; 2009:494539; 2009:554469; 2009:412604; 2008:939303; 2008:891340; 2008:723802; 2008:544780; 2008:508001; 2007:903019; 2007:707103; 2007:454213; 2006:1143616; 2006:920972; 2006:723279; 2006:536751; 2006:157006; 2005:1172226 REFERENCE COUNT: 45 THERE ARE 45 CITED REFERENCES AVAILABLE FOR THIS RECORD. (1) Ayed, A; Nat Struct Biol 2001, V8, P756 ZCAPLUS REFERENCE(S): (2) Bell, S; J Mol Biol 2002, V322, P917 ZCAPLUS (3) Blommers, M; J Am Chem Soc 1997, V119, P3425 ZCAPLUS (4) Bottger, A; Curr Biol 1997, V7, P860 ZCAPLUS (5) Bottger, V; Oncogene 1996, V13, P2141 ZCAPLUS (6) Chen, J; Mol Cell Biol 1993, V13, P4107 ZCAPLUS (7) Chene, P; Nat Rev Cancer 2003, V3, P102 ZCAPLUS (8) Coles, M; Drug Discovery Today 2003, V8, P803 ZCAPLUS (9) Dawson, R; J Mol Biol 2003, V332, P1131 ZCAPLUS (10) Dehner, A: Chembiochem 2003, V4, P870 ZCAPLUS (11) Dyson, H; Nat Rev Mol Cell Biol 2005, V6, P197 ZCAPLUS (12) Fernandez, C; EMBO J 2004, V23, P2039 ZCAPLUS (13) Fischer, P; Trends Pharmacol Sci 2004, V25, P343 ZCAPLUS (14) Fry, D; J Biomol NMR 2004, V30, P163 ZCAPLUS (15) Galatin, P; J Med Chem 2004, V47, P4163 ZCAPLUS (16) Grasberger, B; J Med Chem 2005, V48, P909 ZCAPLUS (17) Hu, Q; EMBO J 1990, V9, P1147 ZCAPLUS (18) Huang, S; EMBO J 1990, V9, P1815 ZCAPLUS

protein-protein interactions with NMR spectroscopy)

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 - 562823-90-9
 RL: PAC (Pharmacological activity); THU (Therapeutic use); BIOL
 - (Biological study); USES (Uses)
 (monitoring effects of antagonists on protein-protein interactions with
- NMR spectroscopy) RN 562823-90-9 ZCAPLUS
- CN Boronic acid, B-[4-[(2E)-3-(3,4-dichlorophenyl)-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

DOCUMENT NUMBER: 142:89411

ENTRY DATE: Entered STN: 07 Jan 2005

TITLE: Quaternary nitrogen heterocyclic compounds for detecting aqueous monosaccharides in physiological

fluids

INVENTOR(S): Geddes, Chris D.; Badugu, Ramachandran; Lakowitz,

Joseph R.

PATENT ASSIGNEE(S): University of Maryland Biotechnology Institute, USA

SOURCE: PCT Int. Appl., 120 pp.

CODEN: PIXXD2
OCUMENT TYPE: Patent

DOCUMENT TYPE: Patent LANGUAGE: English

CLASSIFICATION: 9-16 (Biochemical Methods)

Section cross-reference(s): 14, 63

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

	PATENT NO.								APPLICATION NO.									
				A2 20050106			WO 2004-US22717											
		W:										, BG,						
												, EC,						
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		RW:										, SL,						
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						BF,	BJ,	CF,	CG,	CI,	CM,	, GA,	GN,	GQ,	GW,	ML,	MR,	NE,
		1611		TD,				2006	0.110			2004	7700	0.5			0010	
	EP 1644330			B1 20110817								20040626						
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		7718									00 .	2005	3100	05		-	0031	221
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PRIC								2010	1125			2003-						
- 1110	PRIORITY APPLN. INFO.:										2003-							
												2003						

PATENT CLASSIFICATION CODES:

SUPPL. TERM:

INDEX TERM:

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IPCR C07D0213-02 [I,A]; A61B [I,S]; A61K0031-44 [I,A];
                       A61K0031-47 [I,A]; A61K0049-00 [I,A]; C07D0215-00
                       [I,A]; C07F0005-02 [I,A]
                ECLA
                       A61K0049-00P4F4C; C07F0005-02C
AT 520661
                IPCI
                       C07D0213-02 [I,A]; A61K0031-44 [I,A]; A61K0031-47
                       [I,A]; C07D0215-00 [I,A]; C07F0005-02 [I,A]
                IPCR
                       A61B [I,S]; A61K0049-00 [I,A]
                ECLA
                       A61K0049-00P4F4C; C07F0005-02C
US 20070020182 IPCI
                       A61K0049-00 [I.A]; C07F0005-02 [I.A]; G01N0033-00
                       [I,A]; C07D0215-38 [I,A]
                IPCR
                       A61K0049-00 [I,A]; A61B [I,S]; A61K0031-44 [I,A];
                       A61K0031-47 [I,A]; C07D0213-02 [I,A]; C07D0215-00
                       [I,A]; C07F0005-02 [I,A]; G01N0033-00 [I,A];
                       C07D0215-38 [I.A]
                       424/009.600; 436/095.000; 544/229.000; 546/013.000;
                NCL
                       546/171.000
                ECLA
                       A61K0049-00P4F4C; C07F0005-02C
US 20100297016
                IPCI
                       A61K0049-00 [I,A]; C07F0005-02 [I,A]; C07H0001-00
                       [I,A]; G01N0021-76 [I,A]; C12Q0001-02 [I,A];
                       G01N0033-00 [I,A]
                IPCR
                       A61K0049-00 [I,A]; A61B [I,S]; A61K0031-44 [I,A];
                       A61K0031-47 [I,A]; C07D0213-02 [I,A]; C07D0215-00
                       [I,A]; C07F0005-02 [I,A]; C07H0001-00 [I,A];
                       C12Q0001-02 [I,A]; G01N0021-76 [I,A]; G01N0033-00 [I,A]
                NCL
                       424/009.100; 435/029.000; 436/095.000; 436/172.000;
                       536/001.110; 544/229.000; 546/013.000
                ECLA
                      A61K0049-00P4F4C; C07F0005-02C
ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT
                       MARPAT 142:89411
OTHER SOURCE(S):
ABSTRACT:
Disclosed are quaternary nitrogen heterocyclic boronic acid-containing compds.
which are sensitive to glucose and fructose, as well as a variety of other
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physiol. important analytes, such as aqueous chloride and iodide, and a method of using the compds. Also disclosed is a contact lens doped with the quaternary nitrogen heterocyclic boronic acid-containing compound, and a method of using the doped contact lens to measure the Concentration of analyte in tears under physiol. conditions.

quaternary nitrogen heterocyclic compd detecting aq

		monosaccharide physiol fluid
INDEX	TERM:	Eye
		(Ocular fluid; quaternary nitrogen heterocyclic
		compds. for detecting aqueous monosaccharides in
		physiol. fluids)
INDEX	TERM:	Medical goods
		(Ophthalmic, implantable; quaternary nitrogen
		heterocyclic compds. for detecting aqueous
		monosaccharides in physiol. fluids)
INDEX	TERM:	Medical goods
		Sensors
		(Ophthalmic; quaternary nitrogen heterocyclic
		compds. for detecting aqueous monosaccharides in
		physiol. fluids)

Fluorometry

(Ratiometric; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Eye (aqueous humor; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Fluorescent substances (boronic acid-containing; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Acids, preparation Group IIIA element compounds ROLE: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (boronic acids, fluorophores containing; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Bond (covalent; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Body fluid (interstitial; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Heterocyclic compounds ROLE: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (nitrogen, Quaternary; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Heterocyclic compounds ROLE: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (nitrogen, boronic acid-containing, quaternary; quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Adhesion, physical Blood analysis Body fluid Concentration (condition) Contact Lenses Containers Fluorescence Fluorometry Optical anisotropy Permeation Reaction Sensors Solutions

Surface Tear (ocular fluid) Test kits UV and visible spectroscopy (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Monosaccharides ROLE: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: Polymers, uses ROLE: DEV (Device component use); USES (Uses) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: 50-99-7, D-Glucose, analysis 57-48-7, D-Fructose, analysis 16887-00-6, Chloride, analysis 20461-54-5, Iodide, analysis ROLE: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: 406719-91-3 406719-92-4 406719-93-5 406719-94-6 406719-95-7 ROLE: ARG (Analytical reagent use); PRP (Properties); ANST (Analytical study); USES (Uses) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: 63816-10-4P 784146-24-3P 784146-26-5P 784146-28-7P ROLE: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: 91-62-3, 6-Methylquinoline 100-39-0, Benzyl bromide 5263-87-6, 6-Methoxyquinoline 51323-43-4, m-Boronobenzyl bromide 68162-47-0, p-Boronobenzyl bromide 91983-14-1, o-Boronobenzyl bromide ROLE: RCT (Reactant); RACT (Reactant or reagent) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) INDEX TERM: 784146-23-2P 784146-25-4P 819814-02-3P ROLE: SPN (Synthetic preparation); PREP (Preparation) (quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids) OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2 CITINGS) DATE LAST CITED: Date last citing reference entered STN: 26 Nov 2010 OS.CITING.REFS: CAPLUS 2010:1428388; 2006:1289696 REFERENCE COUNT: THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD. REFERENCE(S): (1) Anon: WO 0186264 A1 ZCAPLUS 50-99-7, D-Glucose, analysis RL: ANT (Analyte); BSU (Biological study, unclassified); ANST (Analytical

10/566406

study); BIOL (Biological study)

(quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids)

- 50-99-7 ZCAPLUS CN D-Glucose (CA INDEX NAME)

Absolute stereochemistry.

406719-92-4 406719-94-6

RL: ARG (Analytical reagent use); PRP (Properties); ANST (Analytical study); USES (Uses)

(quaternary nitrogen heterocyclic compds. for detecting aqueous monosaccharides in physiol. fluids)

- 406719-92-4 ZCAPLUS RN
- CN Boronic acid, B-[4-[3-[4-(dimethylamino)phenyl]-1-oxo-2-propen-1vl]phenvl]- (CA INDEX NAME)

- 406719-94-6 ZCAPLUS RN
- CN Boronic acid, B-[4-[5-[4-(dimethylamino)phenyl]-1-oxo-2,4-pentadien-1yl]phenyl]- (CA INDEX NAME)

ACCESSION NUMBER: DOCUMENT NUMBER: ENTRY DATE:

TITLE:

142:234524 Entered STN: 04 Aug 2004

L51 ANSWER 11 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN 2004:621102 ZCAPLUS Full-text

Cvanide-sensitive fluorescent probes

AUTHOR(S): Badugu, Ramachandram; Lakowicz, Joseph R.; Geddes,

Chris D.

CORPORATE SOURCE: Center for Fluorescence Spectroscopy, Department of

Biochemistry and Molecular Biology, Medical

Biotechnology Center, University of Maryland School of

Medicine, Baltimore, MD, 21201, USA

Dyes and Pigments (2005), 64(1), 49-55

CODEN: DYPIDX; ISSN: 0143-7208

PUBLISHER: Elsevier Ltd. DOCUMENT TYPE: Journal LANGUAGE: English

CLASSIFICATION: 4-1 (Toxicology)

ABSTRACT:

SOURCE:

We characterize the response of several boronic acid containing fluorophores, which are widely used for sugar determination, towards aqueous cyanide. In two recent

reports

we have shown that boronic acid containing fluorophores can be used to sense

aqueous

cyanide through physiol. safeguard levels. In this report we show that our new sensing mechanism is not just specific to our recently reported probes, but is indeed generic to the boronic acid moiety itself. Subsequently a wide range of cyanide-sensitive probes can now be realized, offering several modalities for fluorescence based cyanide sensing such as: intensity, lifetime, ratiometric, polarization and modulation fluorescence sensing.

SUPPL. TERM: cyanide sensing fluorescent probe

INDEX TERM: Fluorescent indicators

> (boronic acid-containing; cyanide sensing by boronic acid-containing fluorescent probes)

INDEX TERM: Fluorometry

(cyanide sensing by boronic acid-containing

fluorescent probes)

INDEX TERM: 57-12-5, Cyanide, analysis

ROLE: ANT (Analyte); ANST (Analytical study) (cyanide sensing by boronic acid-containing

fluorescent probes)

INDEX TERM: 162254-07-1, ANDBA 387869-15-0, PANSBA 406719-91-3,

CSTBA 406719-92-4, Chalc 1 406719-93-5, DSTBA

ROLE: ARG (Analytical reagent use); PRP (Properties); ANST

(Analytical study); USES (Uses)

(cvanide sensing by boronic acid-containing

fluorescent probes)

OS.CITING REF COUNT: 56 THERE ARE 56 CAPLUS RECORDS THAT CITE THIS RECORD (56 CITINGS)

DATE LAST CITED: Date last citing reference entered STN: 04 Oct 2011

OS.CITING.REFS: CAPLUS 2011:1223218; 2011:797568; 2011:805145; 2011:800011;

2011:498892; 2011:476192; 2011:223030; 2011:330297; 2011:317839; 2010:1592856; 2011:82874; 2010:1467081; 2010:1424734; 2010:1125778; 2010:1083809; 2010:938709; 2010:797026; 2010:372843; 2010:430797; 2010:67917;

2009:1581793; 2009:1508405; 2009:1256622;

2009:1073942; 2009:788190; 2009:850677; 2009:362350; 2009:467324; 2009:111292; 2009:50982; 2008:1203090; 2008:1126750; 2008:973436; 2008:873394; 2008:806353;

2008:507194; 2008:502873; 2008:210493; 2008:97338; 2007:1011172; 2007:970817; 2007:915860; 2007:544629; 2007:257600; 2007:4114; 2006:1191619; 2006:1166915; 2006:1133685; 2006:937699; 2006:718797 REFERENCE COUNT: 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD. REFERENCE(S): (1) Badugu, R; Anal Biochem 2004, V327(1), P82 ZCAPLUS (2) Badugu, R; Anal Chem 2004, V76(3), P610 ZCAPLUS (3) Badugu, R; Dyes Pigments 2004, V61(3), P227 ZCAPLUS (4) Badugu, R; In preparation (5) Badugu, R; Sensors Actuators B Chem, in press (6) Badugu, R; Submitted for publication (7) Diceasre, N; Langmuir 2002, V18, P7785 (8) Dicesare, N; Anal Biochem 2001, V294, P154 ZCAPLUS (9) Dicesare, N; Anal Biochem 2002, V301(1), P111 ZCAPLUS (10) Dicesare, N; Chem Commun 2001, P2022 ZCAPLUS (11) Dicesare, N; J Biomed Optics 2002, V7(4), P538 ZCAPLUS (12) Dicesare, N; J Fluoresc 2002, V12(2), P147 ZCAPLUS (13) Dicesare, N; J Photochem Photobiol A Chem 2001, V143, P39 ZCAPLUS (14) Dicesare, N; J Phys Chem A 2001, V105, P6834 ZCAPLUS (15) Dicesare, N; Org Lett 2001, V3(24), P3891 ZCAPLUS (16) Dicesare, N; Tetrahedron Lett 2002, V43, P2615 ZCAPLUS (17) Filipovic-Kovaceic, Z; Eur Food Res Technol 2002, V215(4), P347 (18) Geddes, C; Meas Sci Technol 2001, V12, PR53 ZCAPLUS (19) Gryczynski, Z; Methods Enzymol 2002, V360, P44 (20) Ishii, A; Anal Chem 1998, V70(22), P4873 ZCAPLUS (21) Karnati, V; Bioorg Med Chem Lett 2002, V12, P3373 ZCAPLUS (22) Lakowicz, J; Principles of fluorescence spectroscopy 2nd ed 1997 (23) Licht, S; Anal Chem 1996, V68(6), P954 ZCAPLUS (24) Lu, J; Anal Chim Acta 1995, V304(3), P369 ZCAPLUS (25) Moriya, F; J For Sci 2001, V46(6), P1421 ZCAPLUS (26) Ng, B; J Electrochem Soc 2000, V147(6), P2350 ZCAPLUS (27) Presmasiri, W; J Raman Spectrosc 2001, V32(11), P919 (28) Rao, V; Bull Electrochem 1997, V13(7), P327 ZCAPLUS (29) Recalde-Ruiz, D; Quim Anal 1999, V18, P111 ZCAPLUS (30) Tessier, P; Appl Spectrosc 2002, V56(12), P1524 ZCAPLUS ΙT 406719-92-4, Chalc 1 RL: ARG (Analytical reagent use); PRP (Properties); ANST (Analytical study); USES (Uses) (cyanide sensing by boronic acid-containing fluorescent probes) RN 406719-92-4 ZCAPLUS CN Boronic acid, B-[4-[3-[4-(dimethylamino)phenyl]-1-oxo-2-propen-1vllphenvll- (CA INDEX NAME)

L51 ANSWER 12 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2003:1011575 ZCAPLUS Full-text

DOCUMENT NUMBER: 140:195805

ENTRY DATE: Entered STN: 30 Dec 2003

TITLE: Noninvasive continuous monitoring of physiological

glucose using a monosaccharide-sensing contact lens Badugu, Ramachandram; Lakowicz, Joseph R.; Geddes, AUTHOR(S): Chris D.

CORPORATE SOURCE: Center for Fluorescence Spectroscopy, Department of

Biochemistry and Molecular Biology, Medical Biotechnology Center, University of Maryland School of

Medicine, Baltimore, MD, 21201, USA

SOURCE: Analytical Chemistry (2004), 76(3), 610-618

CODEN: ANCHAM; ISSN: 0003-2700

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

CLASSIFICATION: 9-16 (Biochemical Methods)

ABSTRACT:

We have tested the feasibility of tear glucose sensing using a daily, disposable contact lens embedded with boronic acid-containing fluorophores as a potential alternative to current invasive glucose-monitoring techniques. Our findings show that our approach may, indeed, be suitable for the continuous monitoring of tear glucose levels in the range 50-500 µM, which track blood glucose levels that are .apprx.5-10-fold higher. We compare the response of the boronic acid probes in the contact lens to solution-based measurements and can conclude that both the pH and polarity within the contact lens need to be considered with respect to choosing/designing and optimizing glucose-sensing probes for contact lenses.

SUPPL. TERM: glucose monosaccharide sensing contact eye lens INDEX TERM: Blood analysis

Contact lenses Fluorometry

Hyperglycemia Tear (ocular fluid)

(noninvasive continuous monitoring of physiol.

glucose using monosaccharidesensing contact lens)

INDEX TERM: 50-99-7, D-Glucose, analysis 57-48-7,

Fructose, analysis

ROLE: ANT (Analyte); ANST (Analytical study)

(noninvasive continuous monitoring of physiol. glucose using monosaccharide-

sensing contact lens)

```
INDEX TERM:
                  163927-91-1 357638-58-5 661459-47-8
                   661459-48-9
                               661459-49-0
                   ROLE: ARU (Analytical role, unclassified); ANST (Analytical
                   study)
                      (noninvasive continuous monitoring of physiol.
                     glucose using monosaccharide-
                      sensing contact lens)
OS.CITING REF COUNT: 41 THERE ARE 41 CAPLUS RECORDS THAT CITE THIS RECORD (41
                        CITINGS)
DATE LAST CITED: Date last citing reference entered STN: 02 Sep 2011
OS.CITING.REFS: CAPLUS 2011:1031793; 2008:169850; 2011:320394; 2010:126934;
                         2009:1021213; 2009:799396; 2009:227665; 2008:793135;
                         2008:180791; 2007:1476414; 2007:581067; 2007:544629;
                         2007:266837; 2007:266452; 2007:12104; 2007:11745;
                         2006:1198206; 2006:1113050; 2006:1092856;
                         2006:1027247; 2006:552493; 2006:456037; 2006:320952;
                         2006:121748; 2005:1037120; 2005:436099; 2005:150225;
                         2005:141376; 2005:113259; 2004:1125099; 2004:1120628;
                         2004:1043350; 2004:1038074; 2004:875181; 2004:830500;
                         2004:719141; 2004:697522; 2004:621102; 2004:499011;
                         2004:470034; 2004:407871
REFERENCE COUNT:
                   50
                        THERE ARE 50 CITED REFERENCES AVAILABLE FOR THIS
                        RECORD.
                   (1) Bielecki, M; J Chem Soc, Perkin Trans 1999, V2, P449
                   (2) Cao, H; Org Lett 2002, V4(9), P1503 ZCAPLUS
                   (3) Chen, R; J Capillary Electrophor 1996, V5, P243
                   (4) Clarke, W: Diabetes Res Clin Pract 1988, V4, P209
                             MEDLINE
                   (5) Das, B; J Indian Med Assoc 1995, V93(4), P127 MEDLINE
                   (6) Daum, K; Invest Ophthalmol Vis Sci 1982, V22(4), P509
                             ZCAPLUS
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                   (8) Dicesare, N; Anal Biochem 2001, V294, P154 ZCAPLUS
                   (9) Dicesare, N; Biomed Opt J 2002, V7(4), P538 ZCAPLUS
                   (10) Dicesare, N; Chem Commun 2001, P2022 ZCAPLUS
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                   (12) Dicesare, N; J Photochem Photobiol, A 2001, V143, P39
                             ZCAPLUS
                   (13) Dicesare, N; J Phys Chem A 2001, V105, P6834 ZCAPLUS
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                             ZCAPLUS
                   (17) Eggert, H; J Org Chem 1999, V64, P3846 ZCAPLUS
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                   (23) James, T; J Am Chem Soc 1995, V117, P8982 ZCAPLUS
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- V32, P121 ZCAPLUS
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- IT 50-99-7, D-Glucose, analysis
 - RL: ANT (Analyte); ANST (Analytical study)

(noninvasive continuous monitoring of physiol. glucose using monosaccharide-sensing contact lens

- RN 50-99-7 ZCAPLUS
- CN D-Glucose (CA INDEX NAME)

Absolute stereochemistry.

- IT 661459-48-9 661459-49-0
 - RL: ARU (Analytical role, unclassified); ANST (Analytical study) (noninvasive continuous monitoring of physiol. glucose using monosaccharide-sensing contact lens

10/566406

RN 661459-48-9 ZCAPLUS

CN Boronic acid, [4-[(2E)-3-[4-(dimethylamino)phenyl]-1-oxo-2propenyl]phenyl]- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 661459-49-0 ZCAPLUS

CN Boronic acid, [4-[(2E,4E)-5-[4-(dimethylamino)phenyl]-1-oxo-2,4pentadienyl]phenyl]- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

L51 ANSWER 13 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN

ACCESSION NUMBER: 2003:1006923 ZCAPLUS Full-text

DOCUMENT NUMBER: 140:59511

ENTRY DATE: Entered STN: 26 Dec 2003

TITLE: Preparation of boronic chalcone derivatives as

anticancer agents

INVENTOR(S): Khan, Saeed R.

PATENT ASSIGNEE(S): Johns Hopkins University, USA SOURCE:

PCT Int. Appl., 56 pp.

CODEN: PIXXD2 Patent

LANGUAGE: English

CLASSIFICATION: 27-10 (Heterocyclic Compounds (One Hetero Atom))

Section cross-reference(s): 1, 25, 29

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

DOCUMENT TYPE:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2003106384	A2	20031224	WO 2003-US18962	20030612

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WO 2003106384 A3 20040617
        W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
            CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
            GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
            LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
            PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ,
            UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
        RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
            KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES,
            FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR,
            BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
                    A1 20031231 AU 2003-243594 20030612
    AU 2003243594
    US 20050176988
                       A1 20050811 US 2005-517781
                                                                20050420
    US 7514579
                       B2 20090407
    US 20090227542
                       A1 20090910 US 2009-403288
                                                                 20090312
PRIORITY APPLN. INFO.:
                                          US 2002-388255P
                                                            P 20020613
                                          US 2003-444429P P 20030203

WO 2003-US18962 W 20030612

US 2005-517781 A3 20050420
PATENT CLASSIFICATION CODES:
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
WO 2003106384
              IPCI C07C [ICM, 7]
                IPCR C07F0005-02 [I,A]
                ECLA C07F0005-02C
AU 2003243594
               IPCI C07F0005-02 [ICM, 7]
                IPCR C07F0005-02 [I.A]
                ECLA C07F0005-02C
US 20050176988 IPCI C07F0005-02 [I,A]
                IPCR C07F0005-02 [I.A]
                NCL
                      562/007.000
                     C07F0005-02C
                ECLA
US 20090227542 IPCI C07F0005-02 [I,A]; A61K0031-69 [I,A]; A61P0035-00
                      [I,A]; A61P0035-04 [I,A]
                IPCR C07F0005-02 [I,A]; A61K0031-69 [I,A]; A61P0035-00
                      [I.A]: A61P0035-04 [I.A]
                NCL
                      514/064.000; 548/405.000; 562/007.000
                ECLA C07F0005-02C
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Ar OH OH (CH2) n-b-OH I

OTHER SOURCE(S):

GRAPHIC IMAGE:

ABSTRACT:

The present invention relates to novel boronic chalcone derivs. I [Ar =

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

MARPAT 140:59511

(un)substituted heteroaryl, etc.; W = H, etc.; X = Zn, etc.; n = 0 or any integer; Z = (un)substituted alkylene, etc.] which are useful as antitumor/anticancer agents. The activity of compds. of this invention against the growth of human breast cancer cell lines was demonstrated.

SUPPL. TERM: boronic chalcone deriv anticancer agent prepn

INDEX TERM: Bone, neoplasm

(Ewing's sarcoma; preparation of boronic chalcone derivs. as

anticancer agents)

INDEX TERM: Sarcoma

(Ewing's; preparation of boronic chalcone derivs. as

anticancer agents)

INDEX TERM: Sarcoma

(Kaposi's; preparation of boronic chalcone derivs, as

anticancer agents)

INDEX TERM: Skin, neoplasm

(basal cell carcinoma; preparation of boronic chalcone

derivs.
INDEX TERM:

as anticancer agents)

Carcinoma

(basal cell; preparation of boronic chalcone derivs. as

anticancer agents)

INDEX TERM: Intestine, neoplasm

(colon; preparation of boronic chalcone derivs. as anticancer

agents)

INDEX TERM: Intestine, neoplasm

(colorectal; preparation of boronic chalcone derivs. as

anticancer agents)

INDEX TERM: Neoplasm

Neoplasm

(head and neck; preparation of boronic chalcone derivs. as

anticancer agents) Acute myeloid leukemia

INDEX TERM:

Antitumor agents Bladder, neoplasm Brain, neoplasm %ye, neoplasm

Head and Neck, neoplasm

Head and Neck, neoplasm

Human

Kidney, neoplasm Lung, neoplasm Lymphocytic leukemia

Lymphoma

Mammary gland, neoplasm

Melanoma

Mouth, neoplasm Neoplasm

Ovary, neoplasm

Pharynx, neoplasm Prostate gland, neoplasm

Stomach, neoplasm Testis, neoplasm

Uterus, neoplasm

(preparation of boronic chalcone derivs. as anticancer

agents)

INDEX TERM: Carcinoma

(squamous cell; preparation of boronic chalcone derivs. as

anticancer agents)

INDEX TERM: 562823-84-1# 562823-85-2P 562823-86-3P 562823-87-4P 562823-88-5P 562823-90-9F

562823-91-0P 562823-92-1P

562823-93-2P 562823-94-3P 562823-95-4P

637347-03-6P

ROLE: PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological

study); PREP (Preparation); USES (Uses)

(preparation of boronic chalcone derivs. as anticancer agents)

OS.CITING REF COUNT: 6 THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (7

CITINGS)
DATE LAST CITED: Date last citing reference entered STN: 02 Jun 2010

OS.CITING.REFS: CAPLUS 2010:642561; 2010:105774; 2006:977103; 2005:612309; 2004:1127078; 2004:1089421

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS

RECORD.

REFERENCE(S): (1) Anon; US 5814622 A ZCAPLUS

(2) Anon; US 6083903 A ZCAPLUS(3) Anon; US 6297217 B1 ZCAPLUS

(3) Anon; 03 6237217 B1 2CAPLO IT 562823-84-1P 562823-90-9P 562823-91-0P

562823-92-1P 562823-93-2P

RI: PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(preparation of boronic chalcone derivs. as anticancer agents)

RN 562823-84-1 ZCAPLUS

Double bond geometry as shown.

RN 562823-90-9 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3,4-dichlorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-91-0 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3-chloro-4-fluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-92-1 ZCAPLUS

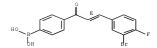
CN Boronic acid, B-[4-[(2E)-3-(3,4-difluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-93-2 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3-bromo-4-fluoropheny1)-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.



L51 ANSWER 14 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2003:410905 ZCAPLUS Full-text

DOCUMENT NUMBER: 139:117464

ENTRY DATE: Entered STN: 30 May 2003
TITLE: Design, Synthesis, and Ev

TITLE: Design, Synthesis, and Evaluation of Novel
Boronic-Chalcone Derivatives as Antitumor Agents
AUTHOR(S): Kumar, Srinivas K.; Hager, Erin; Pettit, Catherine;
Gurulingappa, Hallur; Davidson, Nancy E.; Khan, Saeed

CORPORATE SOURCE: Division of Experimental Therapeutics, Sidney Kimmel

Comprehensive Cancer Center at Johns Hopkins,

Baltimore, MD, 21231, USA

SOURCE: Journal of Medicinal Chemistry (2003), 46(14),

2813-2815

CODEN: JMCMAR; ISSN: 0022-2623
PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

CLASSIFICATION: 29-4 (Organometallic and Organometalloidal Compounds)

Section cross-reference(s): 1

OTHER SOURCE(S): CASREACT 139:117464

ABSTRACT:

A series of boronic-chalcone derivs., e.g. 4-IC6H4CH:CHCOC6H4B(OH)2-4, were synthesized and tested for antitumor activity against human breast cancer cell lines. The results show the boronic-chalcones are more toxic to breast cancer cells compared to normal breast cells than other known chalcones.

SUPPL. TERM: boronic chalcone deriv prepn antitumor activity human breast

cancer

INDEX TERM: Human

(design, synthesis, and evaluation of novel

boronic-chalcone derivs. as antitumor agents) Antitumor agents

INDEX TERM: Antitumor age

(mammary gland; design, synthesis, and evaluation of novel boronic-chalcone derivs. as antitumor agents)

INDEX TERM: 562823-85-2P

ROLE: BSU (Biological study, unclassified); RCT (Reactant); SPN (Synthetic preparation); BIOL (Biological study); PREP

(Preparation); RACT (Reactant or reagent) (design, synthesis, and evaluation of novel

boronic-chalcone derivs. as antitumor agents)
22563-48-0P 562823-86-1P 562823-86-3P

INDEX TERM: 22563-48-0P **562823-84-1P** 562823-86-3P 562823-87-4P 562823-88-5P 562823-89-6P

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562823-90-9P 562823-91-0P
                   562823-92-1P 562823-93-2P
                                               562823-94-3P
                   562823-95-4P
                   ROLE: BSU (Biological study, unclassified); SPN (Synthetic
                  preparation); BIOL (Biological study); PREP (Preparation)
                      (design, synthesis, and evaluation of novel
                      boronic-chalcone derivs. as antitumor agents)
INDEX TERM:
                  99-93-4, 4-Acetylphenol 6287-38-3,
                  3.4-Dichlorobenzaldehyde 15164-44-0, 4-Iodobenzaldehyde
                   34036-07-2, 3,4-Difluorobenzaldehyde 34328-61-5
                   77771-02-9
                               149104-90-5, 4-Acetylphenylboronic acid
                   166330-03-6
                   ROLE: RCT (Reactant); RACT (Reactant or reagent)
                      (design, synthesis, and evaluation of novel
                      boronic-chalcone derivs. as antitumor agents)
OS.CITING REF COUNT: 122 THERE ARE 122 CAPLUS RECORDS THAT CITE THIS RECORD
                         (122 CITINGS)
DATE LAST CITED: Date last citing reference entered STN: 17 Oct 2011
OS.CITING.REFS: CAPLUS 2011:881455; 2011:1190265; 2011:991168; 2011:596846;
                        2011:458294; 2011:492175; 2011:466790; 2011:443610;
                         2011:439588; 2011:374823; 2011:272137; 2011:65469;
                         2011:11629; 2010:1451501; 2010:1454435; 2010:1346932;
                        2010:1440858; 2010:1394766; 2010:1372747;
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                         2009:1283172; 2009:1117164; 2009:655687; 2009:430280;
                         2009:503779; 2009:489090; 2009:480379; 2009:471012;
                         2009:166506; 2009:396625; 2009:241555; 2008:1334426
REFERENCE COUNT:
                  34
                         THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS
                         RECORD.
REFERENCE(S):
                   (1) Baker, S; Science 1990, V249, P912 MEDLINE
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- 562823-92-1P 562823-93-2P
 - RL: BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)
 - (design, synthesis, and evaluation of novel boronic-chalcone derivs. as antitumor agents)
- RN 562823-84-1 ZCAPLUS
- CN Boronic acid, B-[4-[(2E)-3-(4-iodophenyl)-1-oxo-2-propen-1-yl]phenyl](CA INDEX NAME)

Double bond geometry as shown.

- RN 562823-90-9 ZCAPLUS
- CN Boronic acid, B-[4-[(2E)-3-(3,4-dichlorophenyl)-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-91-0 ZCAPLUS

CN Boronic acid, B=[4-[(2E)-3-(3-chloro-4-fluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-92-1 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3,4-difluorophenyl)-1-oxo-2-propen-1yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

RN 562823-93-2 ZCAPLUS

CN Boronic acid, B-[4-[(2E)-3-(3-bromo-4-fluorophenyl)-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

Double bond geometry as shown.

ACCESSION NUMBER: 2002:211081 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 137:134122
ENTRY DATE: Entered STN: 20 Mar 2002

TITLE: Chalcone-analogue fluorescent probes for

saccharides signaling using the boronic acid group

AUTHOR(S): DiCesare, Nicolas; Lakowicz, Joseph R.

CORPORATE SOURCE: Center for Fluorescence Spectroscopy, School of

Medicine, University of Maryland, Baltimore, MD,

21201, USA

SOURCE: Tetrahedron Letters (2002), 43(14), 2615-2618

CODEN: TELEAY; ISSN: 0040-4039
PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal

LANGUAGE: English

CLASSIFICATION: 80-2 (Organic Analytical Chemistry)

Section cross-reference(s): 33, 73
ABSTRACT:

Two new fluorescent probes based on 1,3-diphenylprop-2-en-1-one and on

1,5-diphenylpenta-2,4-dien-1-one structures are presented. Both probes possess one electron-donating dimethylamino group and one boronic acid group

(electron-withdrawing group). The change between the neutral and the anionic form of the boronic acid group induced at high pH and/or in presence of sugar, induces optical changes for both probes. Spectroscopic data, pKa

and dissociation consts. for different monosaccharides are presented and discussed in terms of sugar detection.

arseassea in terms or sagas acception.

SUPPL. TERM: chalcone boronic acid fluorescent probe saccharide detection

INDEX TERM: Monosaccharides

ROLE: ANT (Analyte); ANST (Analytical study)
(analytes; chalcone-analog fluorescent probes
for saccharides signaling using the boronic

acid group)

INDEX TERM: Optical sensors

INDEX TERM:

(fluorescent; chalcone-analog

fluorescent probes for saccharides

signaling using the boronic acid group)

Dissociation constant

Fluorescence Molar absorptivity

(of 1,3-diphenylprop-2-en-1-one and on

1,5-diphenylpenta-2,4-dien-1-one containing dimethylaniline

and boronic acid group)

INDEX TERM: 50-99-7, Glucose, analysis 57-48-7,

Fructose, analysis 59-23-4, Galactose, analysis

ROLE: ANT (Analyte); ANST (Analytical study)
(analytes; chalcone-analog fluorescent probes

for saccharides signaling using the boronic

acid group)

INDEX TERM: 406719-92-4 406719-94-6

ROLE: ARU (Analytical role, unclassified); DEV (Device

component use); PRP (Properties); ANST (Analytical study);

USES (Uses)

(chalcone-analog fluorescent probes for

saccharides signaling using the boronic acid

```
group)
INDEX TERM:
                   100-10-7
                             6203-18-5 149104-90-5
                   ROLE: RCT (Reactant); RACT (Reactant or reagent)
                      (in preparation of 1,3-diphenylprop-2-en-1-one and on
                      1,5-diphenylpenta-2,4-dien-1-one containing dimethylaniline
                      and boronic acid group)
OS.CITING REF COUNT: 53 THERE ARE 53 CAPLUS RECORDS THAT CITE THIS RECORD (53
DATE LAST CITED: Date last citing reference entered STN: 05 Aug 2011
OS.CITING.REFS: CAPLUS 2011:928818; 2008:169850; 2011:330297; 2010:571477;
                         2010:430797; 2010:82926; 2009:1313614; 2009:503779;
                         2009:480379; 2008:499923; 2007:1388279; 2007:719542;
                         2007:686863; 2007:521464; 2007:266837; 2007:266452;
                         2007:239704; 2007:2736; 2006:529462; 2006:496977;
                         2006:456037; 2006:166391; 2005:1148227; 2005:1114291;
                         2005:1051890; 2005:649464; 2005:612309; 2005:601944;
                         2005:590446; 2005:533620; 2005:483103; 2005:320752;
                         2005:150225; 2005:141376; 2005:113259; 2004:1125099;
                         2004:1043350; 2004:1010934; 2004:938446; 2004:806116;
                         2004:697536; 2004:697529; 2004:697522; 2004:621102;
                         2004:407871; 2004:290806; 2004:115138; 2003:1011575;
                         2003:814314; 2003:410905
                        THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                 17
                         RECORD.
REFERENCE(S):
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     50-99-7, Glucose, analysis
ΙT
     RL: ANT (Analyte); ANST (Analytical study)
        (analytes: chalcone-analog fluorescent probes for
        saccharides signaling using the boronic acid group)
RN
     50-99-7 ZCAPLUS
CN
     D-Glucose (CA INDEX NAME)
Absolute stereochemistry.
```

IT 406719-92-4 406719-94-6

RL: ARU (Analytical role, unclassified); DEV (Device component use); PRP (Properties); ANST (Analytical study); USES (Uses)

(chalcone-analog fluorescent probes for saccharides signaling using the boronic acid group)

RN 406719-92-4 ZCAPLUS

CN Boronic acid, B-[4-[3-[4-(dimethylamino)pheny1]-1-oxo-2-propen-1-yl]phenyl]- (CA INDEX NAME)

RN 406719-94-6 ZCAPLUS

CN Boronic acid, B-[4-[5-[4-(dimethylamino)phenyl]-1-oxo-2,4-pentadien-1yl]phenyl]- (CA INDEX NAME)

L51 ANSWER 16 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2002:73583 ZCAPLUS Full-text

DOCUMENT NUMBER:

136:275539 Entered STN: 28 Jan 2002

ENTRY DATE:

New sensitive and selective fluorescent probes for

TITLE: New sensitive and selective : fluoride using boronic acids

AUTHOR(S): DiCesare, Nicolas; Lakowicz, Joseph R.
CORPORATE SOURCE: Center for Fluorescence Spectroscopy, University of

Maryland, School of Medicine, Baltimore, MD, 21201,

USA

SOURCE: Analytical Biochemistry (2002), 301(1), 111-116 CODEN: ANBCA2; ISSN: 0003-2697

PUBLISHER: Academic Press

DOCUMENT TYPE: Journal LANGUAGE: English

CLASSIFICATION: 9-5 (Biochemical Methods)

Section cross-reference(s): 14, 61, 63

ABSTRACT:

We report the spectroscopic characterization of six fluorescent probes for fluoride seasing and/or monitoring. All probes are based on the ability of the boronic acid group to interact with fluoride. The probes combine electron donor and withdrawing groups and involve the excited charge transfer mechanism. The change between the neutral form of the boronic acid group [R-B(OH)2], which is an electron withdrawing group, and the anionic trifluoro form [R-BF-3], which is an electron donating group, is at the origin of the different spectral changes observed for the investigated probes. Two probes are based on the stilbene structure where the boronic group in the 4 position is coupled with a cyano group, in one case, and the dimethylamino group in the other case, both at the 4' position. Another probe is based on the diphenyl-1,4-butadiene possessing the boronic acid group in the 4' position and a dimethylamino group in the 4' position. One probe is based on the diphenyloxazole structure having both the boronic acid and the dimethylamino groups in para positions. The two last probes reported are based on the benzalacetophenone (chalcone) structure, again coupling the boronic acid and dimethylamino groups. All probes show spectral shifts and/or intensity changes in the presence of fluoride resulting in most of the cases to a wavelength-ratiometric way for the detection and/or anal. of fluoride. Selectivity and stability consts. are also presented and discussed. (c) 2002 Academic Press.

SUPPL. TERM: fluorescence probe fluoride detn

INDEX TERM: Electron donors Electron transfer

Fluorescent indicators

Fluorometry

(fluorescent probes for fluoride using boronic

acids)

INDEX TERM: 16984-48-8, Fluoride, analysis

ROLE: ANT (Analyte); ANST (Analytical study)

(fluorescent probes for fluoride using boronic acids)

406719-91-3 406719-92-4 406719-93-5 INDEX TERM:

406719-95-7 406719-96-8 406719-94-6

ROLE: ARU (Analytical role, unclassified); ANST (Analytical study)

(fluorescent probes for fluoride using boronic

acids)

OS.CITING REF COUNT: 56 THERE ARE 56 CAPLUS RECORDS THAT CITE THIS RECORD (57 CITINGS)

DATE LAST CITED: Date last citing reference entered STN: 31 May 2011 OS.CITING.REFS: CAPLUS 2011:581418; 2011:330297; 2010:938709; 2010:731911;

2010:372843; 2010:430797; 2009:1508405; 2009:1284960; 2009:1192373; 2009:1167680; 2009:513974; 2009:708542; 2009:326427; 2009:467324; 2009:405592; 2009:180417; 2008:1150229; 2008:1148843; 2008:1126750; 2008:982227; 2008:898411; 2008:601305; 2007:1463534; 2007:1259331;

> 2007:1154244; 2007:1011172; 2007:1001495; 2007:864498; 2007:781248; 2007:596098; 2007:552904; 2007:544629;

2007:402430; 2007:347836; 2007:258546; 2006:1295356; 2006:496977; 2005:1148419; 2005:1016898; 2005:980494; 2005:953114; 2005:601944; 2005:567416; 2005:460733; 2004:1010934; 2004:938446; 2004:715428; 2004:621102; 2004:581074; 2004:495754

REFERENCE COUNT:

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IT

RL: ARU (Analytical role, unclassified); ANST (Analytical study) (fluorescent probes for fluoride using boronic acids)

RN 406719-92-4 ZCAPLUS

CN Boronic acid, B-[4-[3-[4-(dimethylamino)phenyl]-1-oxo-2-propen-1vl]phenvl]- (CA INDEX NAME)

RN 406719-94-6 ZCAPLUS

CN Boronic acid, B-[4-[5-[4-(dimethylamino)phenyl]-1-oxo-2,4-pentadien-1yl]phenyl]- (CA INDEX NAME)

L51 ANSWER 17 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 2001:186546 ZCAPLUS Full-text

DOCUMENT NUMBER: 134:366929

ENTRY DATE: Entered STN: 18 Mar 2001

TITLE: Synthesis, structure and spectroscopic characteristics of 2"-boryl-4"-dimethylaminochalcones. Effect of an intramolecular boron-oxygen coordinate bond to the

conjugated system

AUTHOR(S): Murafuji, Toshihiro; Sugimoto, Kenji; Yanagimoto, Sachiko; Moriya, Tomokazu; Sugihara, Yoshikazu;

Mikata, Yuji; Kato, Masako; Yano, Shigenobu

CORPORATE SOURCE: Department of Chemistry, Faculty of Science, Yamaguchi

University, Yamaquchi, 753-8512, Japan

SOURCE: Heterocycles (2001), 54(2), 929-942

CODEN: HTCYAM; ISSN: 0385-5414

PUBLISHER: Japan Institute of Heterocyclic Chemistry

DOCUMENT TYPE: Journal

LANGUAGE: English

CLASSIFICATION: 29-4 (Organometallic and Organometalloidal Compounds)

Section cross-reference(s): 22, 75

OTHER SOURCE(S): CASREACT 134:366929

ABSTRACT: 2'-Diethylboryl-4"-dimethylaminochalcone (1) and the related compds. (4)-(7)

bearing a dioxyboryl group in the 2'-position were synthesized, and the effect of the intramol. B-O coordinate bond on the spectroscopic characteristics of 4"-dimethylaminochalcone chromophore was examined by comparison with

4"-dimethylaminochalcone (2) using UV/visible and fluorescence spectra.

SUPPL. TERM: crystal structure cyclic boronic ester

boryldimethylaminochalcones; mol structure cyclic boronic ester boryldimethylaminochalcones; boronic ester cyclic

prepn crystal structure fluorescence spectra;

aminochalcone boryl prepn crystal structure fluorescence

spectra

INDEX TERM: Crystal structure

Molecular structure

(of cyclic boronic esters)

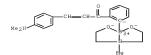
INDEX TERM: Fluorescence

		(and best a should not an about a should be the
		(synthesis, structure and spectroscopic characteristics of boryldimethylaminochalcones)
INDEX	TERM:	Acetalization
		(transacetalization; and cyclization reactions between
		diols and (ethylenedioxyboryl)dimethylaminochalcone)
INDEX	TERM:	5419-55-6, Triisopropoxyborane
		ROLE: RCT (Reactant); RACT (Reactant or reagent)
		(borylation reaction of
		[(ethylenedioxy)ethyl]bromobenzene by)
INDEX	TERM:	100-10-7, 4-Dimethylaminobenzaldehyde 7397-46-8,
		Diethylmethoxyborane ROLE: RCT (Reactant); RACT (Reactant or reagent)
		(condensation reaction with acetophenone trimethylsilyl
		enolate)
INDEX	TERM:	681448-39-5P
		ROLE: SPN (Synthetic preparation); PREP (Preparation)
		(condensation reaction with acetophenone trimethylsilyl
		enolate)
INDEX	TERM:	13735-81-4, Acetophenone trimethylsilyl enol ether
		ROLE: RCT (Reactant); RACT (Reactant or reagent)
		(condensation reaction with diethylmethoxyborane)
INDEX	TERM:	50777-64-5P
		ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
		(preparation and borylation reaction of)
INDEX	TERM:	243140-13-8P
		ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
		(Preparation); RACT (Reactant or reagent)
		(preparation and condensation reactions with
		dimethylaminobenzaldehyde)
INDEX	TERM:	22965-98-6P 340131-44-4P
		ROLE: PRP (Properties); SPN (Synthetic preparation); PREP
		(Preparation)
TNDEY	TERM:	(preparation and crystal structure of) 308103-40-4P
INDEA	I LIMIT.	ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
		(Preparation); RACT (Reactant or reagent)
		(preparation and cyclization reactions of)
INDEX	TERM:	243140-14-9P
		ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
		(Preparation); RACT (Reactant or reagent)
		(preparation and sequential deacetalization and reaction with
T110011	TERM:	dimethylaminobenzaldehyde) 340131-41-1P
INDEX	TERM:	ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
		(Preparation); RACT (Reactant or reagent)
		(preparation and transacetalization with diols to give cyclic
		boronic esters)
INDEX	TERM:	340131-40-0P 340131-42-2P 340131-43-3P
		ROLE: SPN (Synthetic preparation); PREP (Preparation)
		(preparation of)
INDEX	TERM:	340131-39-7P
		ROLE: PRP (Properties); RCT (Reactant); SPN (Synthetic
		preparation); PREP (Preparation); RACT (Reactant or reagent)

INDEX NAME)

```
(preparation, crystal structure and reaction with hexylamine)
INDEX TERM:
                   340131-38-6P
                   ROLE: PRP (Properties); RCT (Reactant); SPN (Synthetic
                   preparation); PREP (Preparation); RACT (Reactant or reagent)
                      (preparation, mol. structure and reaction with hexylamine)
INDEX TERM:
                   111-26-2, n-Hexylamine
                   ROLE: RCT (Reactant); RACT (Reactant or reagent)
                      (reaction with diethylboryldimethylaminochalcone)
INDEX TERM:
                   2142-69-0, 2'-Bromoacetophenone
                   ROLE: RCT (Reactant); RACT (Reactant or reagent)
                      (sequential acetalization reaction and borylation
                      reaction of)
                   76-09-5, Pinacol
                                      105-59-9, N-Methyldiethanolamine
INDEX TERM:
                   504-63-2, 1,3-Propanediol
                   ROLE: RCT (Reactant); RACT (Reactant or reagent)
                      (transacetalization/cyclization reactions with
                      (ethylenedioxyboryl)dimethylaminochalcone)
OS.CITING REF COUNT: 3
                        THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3
                         CITINGS)
DATE LAST CITED: Date last citing reference entered STN: 16 Feb 2009
OS.CITING.REFS: CAPLUS 2007:258510; 2005:1275420; 2002:974642
REFERENCE COUNT:
                   13
                        THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS
                         RECORD.
                   (1) Kamlet, M; J Org Chem 1983, V48, P2877 ZCAPLUS
REFERENCE(S):
                   (2) Klein, J; J Org Chem 1976, V41, P3307 ZCAPLUS
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                             V20 ZCAPLUS
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ΙT
    340131-43-3P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of)
RN
     340131-43-3 ZCAPLUS
CN
    Boron, [2-[(2E)-3-[4-(dimethylamino)phenyl]-1-oxo-2-propenyl]phenyl][[2,2'-
```

 $(methylimino-\kappa N)$ bis $[ethanolato-\kappa O]](2-)]-, (T-4)-(9CI)$ (CA



L51 ANSWER 18 OF 18 ZCAPLUS COPYRIGHT 2011 ACS on STN ACCESSION NUMBER: 1999:422273 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 131:214326

ENTRY DATE: Entered STN: 08 Jul 1999

TITLE: Structure and spectroscopic characteristics of

2'-diethylboryl-4''-dimethylaminochalcone bearing an intramolecular boron-oxygen coordinate bond

AUTHOR(S): Murafuji, Toshihiro; Sugihara, Yoshikazu; Moriya,

Tomokazu; Mikata, Yuji; Yano, Shiqenobu

CORPORATE SOURCE: Department of Chemistry, Faculty of Science, Yamaguchi

University, Yamaguchi City, 753-8512, Japan RCE: New Journal of Chemistry (1999), 23(7), 683-685

SOURCE: New Journal of Chemistry (1999), 23(7) CODEN: NJCHE5; ISSN: 1144-0546

PUBLISHER: Royal Society of Chemistry

DOCUMENT TYPE: Journal

LANGUAGE: English

CLASSIFICATION: 29-4 (Organometallic and Organometalloidal Compounds)

Section cross-reference(s): 75

How formation of an intramol. coordinate bond affects mol. structure was examined in the structural comparison of (E)-2'-diethylboryl-4''-dimethylaminochalcone

(1) and 2'-ethylenedioxyboryl-4''-dimethylaminochalcone (2) with

chloro{2-[(4-dimethylaminostyry1)carbonyl]phenyl}(4-methylphenyl)bismuthane (3) and (B)-4''-dimethylaminochalcone (4). 1-4 Were prepared and characterized spectroscopically and Nordy cystallog. Structures were determined for 1 and 4.

SUPPL. TERM: crystal structure aminochalcone borylchalcone; mol structure

aminochalcone borylchalcone; aminochalcone ethylboryl ethylenedioxyboryl tolylbismuthino prepn; boron oxygen bond

intramol coordinate borylchalcone; lewis acidity

ethylenedioxyboryl ethylboryl comparison chalcone; solvatochromism borylchalcone chalcone comparison

INDEX TERM: Bond

(boron-oxygen, intramol., coordinate; of

(ethylboryl)aminochalcone)

INDEX TERM: Crystal structure

Molecular structure (of (ethylboryl)aminochalcone and aminochalcone)

INDEX TERM: Solvatochromism

(of ethylboryl(aminochalcone) compared to aminochalcone)

INDEX TERM: Lewis acidity

(of ethylenedioxyboryl group compared to ethylboryl group

INDEX TERM:	in substituted aminochalcones) Linear free energy relationship (solvation energy; between solvent shifts and Kamlet-Taft parameter for UV/VIS spectra of (ethylboryl)aminochalcone
INDEX TERM:	parameter 10 000000 spectra of (ethyloofyl) aminochalcone and aminochalcone) 100-10-7, 4-Dimethylamiobenzaldehyde ROLE: RCT (Reactant); RACT (Reactant or reagent)
INDEX TERM:	(aldol condensation with acetophenone) 98-86-2, Acetophenone, reactions ROLE: RCT (Reactant); RACT (Reactant or reagent)
INDEX TERM:	(aldol condensation with aminobenzaldehyde) 7397-46-8, Methyl diethylborinate 51752-29-5, Chlorodi(p-tolyl)bismuthine ROLE: RCT (Reactant); RACT (Reactant or reagent)
INDEX TERM:	(condensation with phenyl(silyloxy)ethene) 13735-81-4 ROLE: RCT (Reactant); RACT (Reactant or reagent)
INDEX TERM:	(condensation with toly1(chloro)bismuthine) 167771-87-1P ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
INDEX TERM:	(Preparation); RACT (Reactant or reagent) (preparation and condensation with aminobenzaldehyde) 243140-13-8P ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
INDEX TERM:	(Preparation); RACT (Reactant or reagent) (preparation and condensations with aminobenzaldehyde) 243140-09-2P 243140-16-1P
INDEX TERM:	ROLE: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (preparation and crystal structure of) 22965-98-6F
INDEX TERM:	ROLE: RRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (preparation and mol. structure of)
INDEX TERM:	243140-15-0P ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RRCT (Reactant or reagent)
INDEX TERM:	(preparation and substitution by chloride) 243140-14-9P ROLE: RCT (Reactant); SPN (Synthetic preparation); PREP
INDEX TERM:	(Preparation); RACT (Reactant or reagent) (preparation and transacetalization followed by condensation with aminobenzaldehyde) 243140-10-5P 243140-11-6P 243140-12-7P ROLE: SPN (Synthetic preparation); PREP (Preparation)
INDEX TERM:	(preparation of) 50777-64-5 ROLE: RCT (Reactant); RACT (Reactant or reagent)
OS.CITING REF COUN	(substitution with iso-Pr borate followed by hydrolysis) T: 12 THERE ARE 12 CAPLUS RECORDS THAT CITE THIS RECORD (12 CITINGS)
	ate last citing reference entered STN: 14 Aug 2009 APLUS 2009;677921, 2008:886631; 2008:116304; 2007:258510; 2006:1295356; 2006:1171115; 2005:1255223; 2004:552272;

2004:470247; 2002:974642; 2002:575977; 2001:186546

REFERENCE COUNT: 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD.

REFERENCE(S): (1) Jiang, Y; J Photochem Photobiol A: Chem 1994, V81, P205

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(2) Kamlet, M; J Org Chem 1983, V48, P2877 ZCAPLUS

(3) Lippert, E; Z Naturforsch A 1955, V10, P541

(4) Mataga, N; Bull Chem Soc Jpn 1955, V28, P690 ZCAPLUS

(5) Mataga, N; Bull Chem Soc Jpn 1956, V29, P465 ZCAPLUS

(6) Murafuji, T; Organometallics 1995, V14, P3848 ZCAPLUS

(7) Onsager, L; J Am Chem Soc 1936, V58, P1486 ZCAPLUS

(8) Sugihara, Y; J Chem Soc Perkin Trans 1 1995, V22, P2813(9) Sugihara, Y; New J Chem 1998, V22, P1031 ZCAPLUS

(9) Suginara, Y; New J Chem 1998, V22, P1031 2CAPLUS (10) Suzuki, H; J Chem Soc Perkin Trans 1 1993, P1169

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(11) van Walree, C; J Chem Soc Chem Commun 1995, P35 ZCAPLUS

(12) Wang, P; J Photochem Photobiol A: Chem 1995, V86, P109 ZCAPLUS

IT 243140-10-5P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 243140-10-5 ZCAPLUS

CN 2-Propen-1-one, 3-[4-(dimethylamino)phenyl]-1-[2-(1,3,2-dioxaborolan-2-yl)phenyl]- (CA INDEX NAME)

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           STRUCTURE UPLOADED
L*** DEL
           16 S L1
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               E W02004-EP8825/APPS
L1
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L4
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              D L3
L5
              STRUCTURE UPLOADED
L6
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L7
              STRUCTURE UPLOADED
L8
            1 SEA SSS SAM L7
              D SCA
1.9
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               SAVE TEMP L9 WIN406STR7L/A
               D SCA
L10
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T.14
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        325389 SEA SPE=ON ABB=ON PLU=ON ?SACCHARID?
L16
1.17
           490 SEA SPE=ON ABB=ON PLU=ON ?OPTHALM?
L18
        32416 SEA SPE=ON ABB=ON PLU=ON ?OPHTHALM?
L19
        166464 SEA SPE=ON ABB=ON PLU=ON EYE
L20
        717477 SEA SPE=ON ABB=ON PLU=ON ?FLUORESC?
      1438953 SEA SPE-ON ABB-ON PLU-ON OPTIC?
L22
       444462 SEA SPE=ON ABB=ON PLU=ON SENSOR?
        25592 SEA SPE=ON ABB=ON PLU=ON OCULAR?
L23
L24
         7970 SEA SPE=ON ABB=ON PLU=ON CONTACT LEN?
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L25 2633827 SEA SPE=ON ABB=ON PLU=ON POLYMER?

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1.26
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L27
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               D SCA
L28
             7 SEA SPE=ON ABB=ON PLU=ON L13 NOT L27
               D SCA
L29
        411686 SEA SPE=ON ABB=ON PLU=ON SUGAR?
L30
             4 SEA SPE=ON ABB=ON PLU=ON L13 AND L29
L31
             4 SEA SPE=ON ABB=ON PLU=ON BLOOD? AND L13
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L32
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1.33
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L34
         19268 SEA SPE=ON ABB=ON PLU=ON SMITH D?/AU, AUTH
L35
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L36
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L37
L38
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L39
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             1 SEA SPE=ON ABB=ON PLU=ON L39 AND BORO?
L40
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L41
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L43
               OR ?SACCHARID? OR SUGAR)
L44
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L46
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L47
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L49
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                     ANSWERS '8-9' FROM FILE MEDLINE
                     ANSWER '10' FROM FILE BIOSIS
                     ANSWERS '11-12' FROM FILE WPIX
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                D IALL HIT L50 11-12
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               D STAT QUE L31
L51
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     FILE HOME
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     Property values tagged with IC are from the ZIC/VINITI data file
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     STRUCTURE FILE UPDATES: 17 OCT 2011 HIGHEST RN 1337015-67-4
    DICTIONARY FILE UPDATES: 17 OCT 2011 HIGHEST RN 1337015-67-4
    CAS Information Use Policies apply and are available at:
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     TSCA INFORMATION NOW CURRENT THROUGH June 24, 2011.
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    FILE ZCAPLUS
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FILE COVERS 1907 - 18 Oct 2011 VOL 155 ISS 17
FILE LAST UPDATED: 17 Oct 2011 (20111017/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Aug 2011
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Aug 2011

ZCAplus now includes complete International Patent Classification (IPC) reclassification data for the second quarter of 2011.

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This file contains CAS Registry Numbers for easy and accurate substance identification.

FILE MEDLINE

FILE LAST UPDATED: 15 Oct 2011 (20111015/UP). FILE COVERS 1946 TO DATE.

 $\texttt{MEDLINE}\left(R\right)$ is a registered trademark of the U.S. National Library of Medicine (NLM).

MEDLINE and LMEDLINE have been updated with the 2011 Medical Subject Headings (MeSH) vocabulary and tree numbers from the U.S. National Libra of Medicine (NLM). Additional information is available at:

http://www.nlm.nih.gov/pubs/techbull/nd10/nd10_medline_data_changes_2011.

The 2011 Medline reload was completed on January 22, 2011. See HELP RLOAD for details.

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See HELP RANGE before carrying out any RANGE search.

FILE EMBASE

FILE COVERAGE: EMBASE-originated material 1947 to 18 Oct 2011 (20111018/E Unique MEDLINE content 1948 to present

EMBASE is now updated daily. SDI frequency remains weekly (default) and biweekly.

This file contains CAS Registry Numbers for easy and accurate substance identification.

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FILE BIOSIS

FILE COVERS 1926 TO DATE.

CAS REGISTRY NUMBERS AND CHEMICAL NAMES (CNs) PRESENT

FROM JANUARY 1926 TO DATE.

RECORDS LAST ADDED: 12 October 2011 (20111012/ED)

BIOSIS has been augmented with 1.8 million archival records from 1926 through 1968. These records have been re-indexed to match current BIOSIS indexing.

FILE WPIX

FILE LAST UPDATED: 14 OCT 2011 <20111014/UP>
MOST RECENT UPDATE: 201166 <201166/DW>
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>> Now containing more than 1.8 million chemical structures in DCR <<<

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 No update date (UP) has been created for the reclassified documents, but they can be identified by the reclassified specific update codes (see HELP CLA for details) <<</p>
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